

## FTlite Database for FactSage 8.3

The FTlite Database is designed for thermodynamic and phase equilibrium calculations involving **Al** alloys, **Mg** alloys and **Ti** alloys using the FactSage 8.0 Thermochemical Software Package (and later versions). The FTlite Database can also be used to perform calculations involving mixtures of **Li-Na-K-Mg-Ca-Sr-Ba** with several other elements.

Al Alloys
<b>Ag, Al, As, Au, B, Ba, Be, Bi, C, Ca, Ce, Co, Cr, Cu, Dy, Er, Eu, Fe, Ga, Gd, Ge, H, Hf, Ho, In, K, La, Li, Lu, Mg, Mn, Mo, N, Na, Nb, Nd, Ni, O, P, Pb, Pr, Pt, Sb, Sc, Si, Sm, Sn, Sr, Ta, Tb, Ti, Tm, V, W, Y, Yb, Zn, Zr</b>
Mg Alloys
<b>Ag, Al, B, Ba, Be, Bi, C, Ca, Ce, Co, Cr, Cu, Dy, Er, Eu, Fe, Ga, Gd, Ge, H, Ho, In, K, La, Li, Lu, Mg, Mn, Na, Nb, Nd, Ni, O, Pb, Pr, Pt, Sb, Sc, Si, Sm, Sn, Sr, Tb, Ti, Tm, V, Y, Yb, Zn, Zr</b>
Ti Alloys
<b>Ag, Al, B, Ba, C, Ca, Ce, Co, Cr, Cu, Dy, Er, Eu, Fe, Ga, Gd, H, Ho, K, La, Li, Lu, Mg, Mn, Mo, N, Na, Nb, Nd, Ni, O, Pr, Sc, Si, Sm, Sn, Sr, Ta, Tb, Ti, Tm, V, W, Y, Yb, Zn, Zr</b>
Color codes
<b>Red</b> : Al or Mg
<b>Blue</b> : Major alloying elements (full optimisations of binary systems with <b>Al, Mg</b> and <b>Ti</b> and with several minor alloying elements, <b>Al-Mg-Xx</b> ternary systems evaluated (good for Al+Mg-rich regions), several quaternary systems included);
<b>Green</b> : Minor alloying elements (full optimisations of binary systems with Al and Mg);
<b>Black</b> : Optimized for the <b>M-Zz</b> system and few <b>M-Xx-Zz</b> and <b>M-Yy-Zz</b> systems (where <b>M</b> is Al, Mg or Ti);

### Composition Ranges

The database is intended to allow calculations over all ranges of composition, although the assessed data are often most reliable for light metal rich composition ranges (Al-rich, Mg-rich and Ti-rich compositions. Alkali metal-rich compositions). The database can be used for Al alloys in the commercial series 1000, 2000, 3000, 4000, 5000, 6000 and 7000, and for a wide range cast alloys.

Figure 1 gives the list of optimized binary subsystems with the red, blue and green elements given above.

In FTlite 8.3, platinum (Pt) is added as a new element, and many updates are available for Pt-, Ga- and Ge- containing subsystems, for Al-, Mg- and Ti-alloys (except Ge-Ti).

### Temperature Ranges

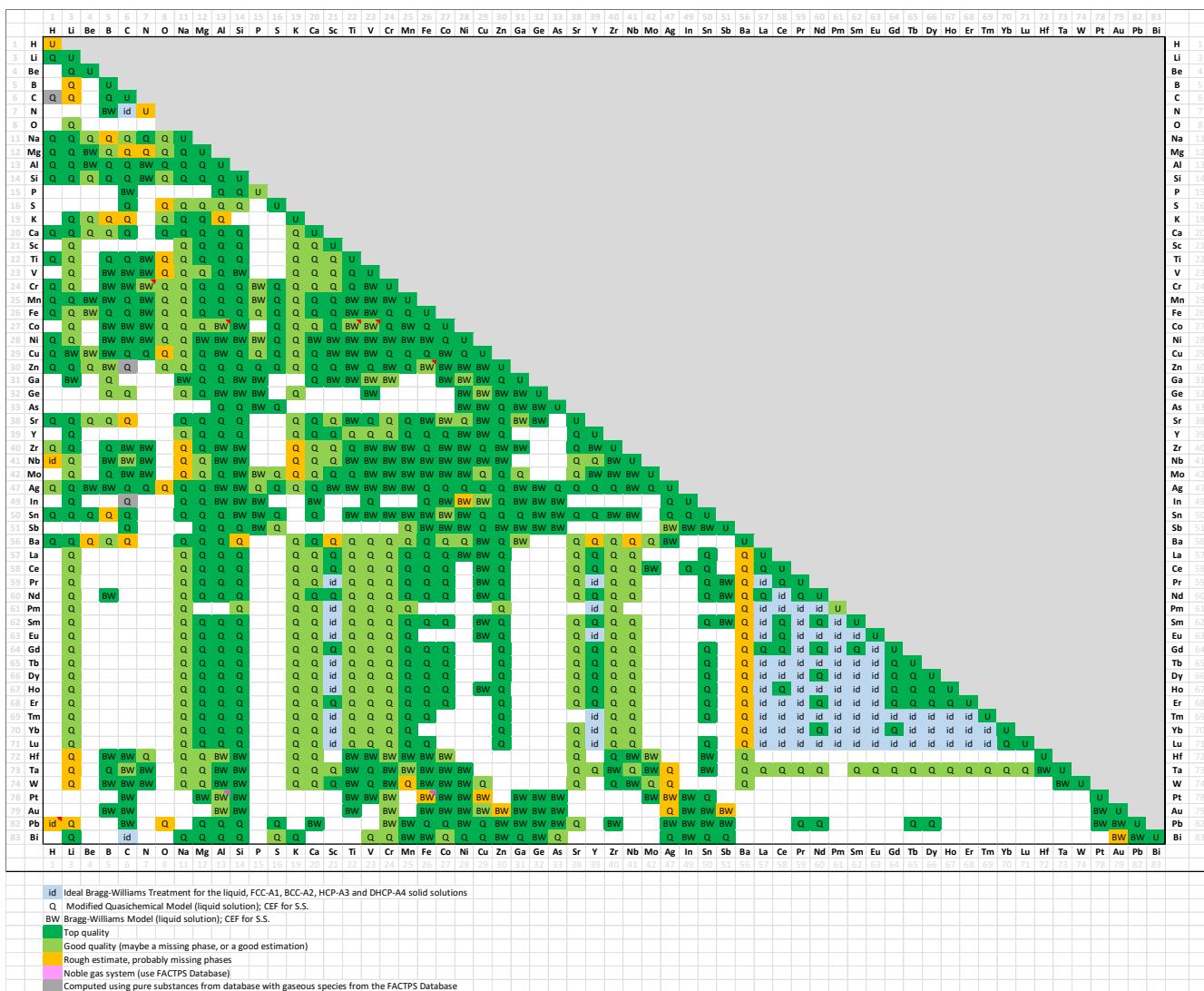
The database is generally valid for the temperature range of approximately room temperature to 2200°C (please remember that kinetics effects can strongly affect phase relationships in alloys in general and in particular at low temperature), although for some alloys containing high melting point metals the data are reliable to still higher temperatures.

In **Table 4**, comments appearing in red can be found for some binary systems which have spurious stable phases at high temperatures, such as inverted liquid-liquid miscibility gaps, or the presence of an inverted solid solutions. The temperature above which the problem occurs is given in **Table 4** as an indication. These inverted stability regions can cause problems in the computation of liquidus projection of ternary or higher order systems in the PHASE DIAGRAM Module and affect the computation of precipitation or formation targets in the EQUILIB Module.

### Physical Properties

The density, viscosity and surface tension of the liquid phase can be computed for most important elements present in the database. The computation of the **surface tension** was a new feature in 8.2. Please be aware that we **use an approach that considers the presence of impurities**, such as oxygen, that have a very strong influence on the surface tension of the melt. As such, you should expect that high values of the surface tension of pure liquids and molten alloys will be computed if oxygen is not included in the calculations. However, the presence, **even at the ppm level**, of oxygen can substantially decrease the surface tension of the liquid. **For example, for pure molten Al, a surface tension of the order of 1.15 N/m is usually measured near the melting point, but this value falls to 0.85 N/m when 1 to 2 ppm of oxygen is present.** The model and its database will be able to compute such phenomena. Similar behaviour can be found for other metals. Special care in the selection of the proper amount of oxygen as impurity is necessary (for example, the surface tension of Al-Mg liquid alloys can be computed by considering  $MgO_{(s)}$ ,  $Al_2O_3_{(s4)}$  and  $MgAl_2O_4_{(s)}$  in an EQUILIB calculation to compute the O solubility limit at solid oxide saturation, these specific solid oxides being present in the database). Coupling FTlite with FToxic may be necessary for computing the oxygen content of a complex liquid alloy saturated with solid oxides.

The thermal conductivity of most non-metallic stoichiometric solids (ex.:  $TiB_2$ ,  $SiC$ ,  $Al_2O_3$ , etc.) can be computed with the FTlite 8.3 database. See FactSage (EQUILIB) documentation on physical properties, and the list of phases at the end of the present document.



**Figure 1:** Optimized binary systems in the FTlite 8.3 Database (see also Table 4)

In 8.3, a total of **992 binary systems** (vs 932 in 8.2) have been evaluated, for most of them over the entire range of composition and for all stable phases. A matrix of assessed binary systems is given in Figure 1, for the major and minor alloying elements of Al-, Mg- and Ti-alloys.

Several dozens of ternary systems have been assessed, and important quaternary systems have also been evaluated.

The FTlite 8.3 database contains **317 solution phases** (vs 287 in 8.2) and **2145 pure compounds** (vs 1926 in 8.2) with **2621 stoichiometric phases** counting allotropic forms (2374 in 8.2). **Table 1A** lists the important solutions for calculations involving Al alloys, while **Table 1B** is for solutions in Mg-alloys and **Table 1C** is for solutions in Ti-alloys. **Table 2** is a detailed list of all solutions. **Table 3** is a generic list of the pure compounds and their associated phases. **Table 4** gives the references for the binary data assessments.

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## Use of the Database

The phase diagrams of all the binary systems listed above have been checked using FactSage 8.3.

Phase selection in the EQUILIB or PHASE DIAGRAM modules using FTlite is simple: simply follow these instructions:

- For pure solid compounds:
  - Right-click on “pure solids”
  - Then click “Select/Clear” > “Add all species from database” > “FTlite”
- For solutions:
  - Click on the “Select” button below the “Solution species” list box
  - Then click “Add all phases from database” > “FTlite”
  - Apply the recommendations related to the CBCC-A12, D82 and D88 solutions, as described in the warning text box in the following page;
- There is no need to select pure liquid phases (the FTlite-Liqu solution contains the liquid species). They may be selected as dormant (metastable, option “!”) for purposes of computing their chemical activity.
- Click “Use V & phys. property data” in the EQUILIB Module if you intend to have density, viscosity, thermal conductivity and surface tension to be calculated for phases, when available. We recommend not to click this option in the PHASE DIAGRAM Module.

There might be cases when a chemical system with many elements results in more than 150 possible solutions to be selected. In these cases, it is recommended to change the selection of some solutions where the “I-option” (“possible 2-phase immiscibility”) or the “J-option” (“possible 3-phase immiscibility”) has been automatically selected, to either “+” (for “I”) or “I-option” (for “J-option”). The solutions where this change must be done should then be selected at the user’s discretion based on his/her knowledge of which solutions might exhibit immiscibility in the composition range the user is interested in.

Within the solution database are hard coded default selections such as “+”, “I” or “J” (for single phase, 2-phase immiscibility and 3-phase immiscibility). However, when the system has more than 5 elements, several specific solution phases will see their default selection changed to “!”, “!I” or “!J”, meaning they become “dormant” (metastable). If in the calculation(s) that will follow one of these phases has a calculated activity higher than 1.0, you should overwrite the “!” (dormant) selection.

The following solution phases have this selection option:

A3” (hcp-Zn)	D88 ( $\text{Mn}_5\text{Si}_3$ family)
A4 (diamond Si, Ge, $\alpha$ -Sn)	D88b ( $\text{RE}_5\text{Sb}_3$ )
A12 ( $\alpha$ -Mn, $\gamma$ - $\text{Al}_{12}\text{Mg}_{17}$ , $\text{Mg}_{24}\text{RE}_5$ )	D8b ( $\sigma$ -FeCr)
D82 ( $\gamma$ -brass family $\text{Cu}_5\text{Zn}_8$ )	cF96 ( $\text{Ti}_2\text{Co}$ , $\text{Ti}_2\text{Ni}$ , $\text{Hf}_2\text{Co}$ )
D85 ( $\mu$ -Fe <sub>7</sub> W <sub>6</sub> )	oP36 ( $\text{RE}_5\text{Si}_4$ , $\text{RE}_5\text{Sn}_4$ )

### CBCC-A12

It is recommended to select the CBCC-A12 solid solution with the “!l” option (metastable + 2-phase immiscibility) in calculations in the EQUILIB module for Al alloys when more than 6 elements are present in the system. This is due to the very numerous number of end-members (4860) in that solution, which increases substantially the computation time. If after a calculation the activity of the CBCC-A12 solution is greater than 1.0, then redo the calculation with the standard “l” option.

Note that for Mg-alloys, the CBCC-A12 phase is very often stable with Mg-HCP-A3 as both  $\alpha$ -Mn,  $\beta$ -Mg<sub>17</sub>Al<sub>12</sub> and Mg<sub>24</sub>RE<sub>5</sub> solutions.

In Al-alloys, the CBCC-A12 phase is NOT often stable with Al-FCC-A1 but may form upon reaction of Al concentrating Mn, REE and/Mg in residual phases.

### D82 Prototype Cu<sub>5</sub>Zn<sub>8</sub>

Similarly it is recommended to select the D82 solid solution (Cu<sub>5</sub>Zn<sub>8</sub>, Al<sub>8</sub>M<sub>5</sub> with M = Fe, Cr, V) with the “!l” option (metastable + 2-phase immiscibility) in calculations in the EQUILIB module for Al alloys when more than 6 elements are present in the system. This is due to the very numerous number of end-members (8712) in that solution, which increases substantially the computation time. If after a calculation the activity of the D82 solution is greater than 1.0, then redo the calculation with the standard “l” option.

In both Al and Mg alloys, the D82 solution is rarely stable with either Al-FCC-A1 or Mg-HCP-A3 but may form upon reactions of Al and/or Mg concentrating D82-forming elements in residual phases.

### Gamma-brass D88

Similarly it is recommended to select the D88 solid solution with the “!l” option (metastable + 2-phase immiscibility) in calculations in the EQUILIB module for Al alloys when more than 6 elements are present in the system. This is due to the very numerous number of end-members (6084) in that solution, which increases substantially the computation time. If after a calculation the activity of the D88 solution is greater than 1.0, then redo the calculation with the standard “l” option.

In both Al and Mg alloys, the D88 solution is rarely stable with either Al-FCC-A1 or Mg-HCP-A3 but may form upon reactions of Al and/or Mg concentrating D88-forming elements in residual phases.

## Updates in 8.3 from 8.2

Here is a list of new assessments or new assessments integrated from publications in 8.3:

- 1) Ag – N
- 2) Ag – Pt
- 3) Ag – Y
- 4) Al – Pt
- 5) Al – Ta (replaced)
- 6) As – Pt
- 7) As – Zn (replaced)
- 8) Au – Ga
- 9) Au – Pt
- 10) B – Mo (replaced)
- 11) B – Ta
- 12) Ba – Ga
- 13) Bi – Mn
- 14) C – Pt
- 15) Ca – Cu (replaced)
- 16) Ca – Ga
- 17) Ca – Ge
- 18) Co – Ga
- 19) Co – Pt
- 20) Cr – Ga
- 21) Cr – Pt
- 22) Cu – N
- 23) Cu – Pt
- 24) Fe – In (replaced)
- 25) Fe – Pt
- 26) Ga – Li
- 27) Ga – Mo
- 28) Ga – N
- 29) Ga – Ni
- 30) Ga – Pt
- 31) Ga – Sc
- 32) Ga – Sr
- 33) Ga – V
- 34) Ga – Zr
- 35) Ge – K
- 36) Ge – Na

- 37) Ge – Ni
- 38) Ge – Pt
- 39) Ge – Sr
- 40) Ge – V
- 41) H – Sn
- 42) Hf – Li
- 43) Hf – N
- 44) Hf – Ni
- 45) In – Pt
- 46) Li – Sb
- 47) Li – Ta
- 48) Li – W
- 49) Mg – Pt
- 50) Mn – Sb
- 51) Mo – Pt
- 52) Ni – Pt
- 53) P – Sn
- 54) Pb – Pt
- 55) Pt – Si
- 56) Pt – Sn
- 57) Pt – Ta
- 58) Pt – Ti
- 59) Pt – V
- 60) Sn – Ta

Many existing solutions were modified to introduce the new systems (*i.e.* Liquid metal, FCC-A1, BCC-A2, HCP-A3, etc.). Those are not necessarily listed below. Also, many existing solutions were modified to introduce new end-members:

- A3' addition of Au(Sn)
- Diamond-A4 addition of In and V
- CBCC-A12 addition of Sb
- CUB-A13 addition of Sb
- A15 addition of Ga, Ge, Pt
- BCC-B2!BCC-A2 addition of Pt (AlPt, MnNi, CoTi, CoHf, CoPt)
- BCC-B2a addition of Ag (AgY, AgSc);  
improved Gibbs energy of metastable end-members
- BCC-B2b addition of Au and Pt (PtTi, NiPt, AuTi)
- B8x (B81/B82) addition of Ge (Ni-Ge), improved Al-Ti-Zr, new Ga-Ni-Ti,  
improved Cu-Ti-Sn, Fe-Mn-Sn, Cu-Fe-Sn, Mn-Bi
- B11 addition of Au for AuTi
- B20 addition of Ni, Pt, Ga, Mg for NiSi, PtAl, PtGa, PtMg
- B27 addition of Cr, Mo, Ni for CrB, MoB, NiB
- B31 addition of Ga, Ge, Pt
- B32 addition of Ga for GaLi
- B33 addition of Fe, Mo, Ge for FeB, MoB, CaGe
- Bg addition of Zr, Ga, Cr, Fe, Ti for borides and ZrGa
- C1a addition of Pt, Mn, Ga for GaPt, AlAu, AlPt;  
improved Gibbs energy of metastable end-members
- C2 addition of As, Ag and Pt for PtAs<sub>2</sub>, PtSb<sub>2</sub> and Ag solubility;  
improved Gibbs energy of metastable end-members
- C11 addition of Y for Ag<sub>2</sub>Y
- C11b improved Gibbs energy of metastable end-members
- C12 addition of Ge and Sn for CaGe<sub>2</sub> and CaSn<sub>2</sub>
- C14 improved Gibbs energy of metastable end-members
- C15 improved Gibbs energy of metastable end-members
- C15a addition of Au and Bi for Au<sub>2</sub>Bi
- C15b addition of Hf
- C16 addition of Ga, Mg, Cr, Mo for Zr<sub>2</sub>Ga and  
many ternary boride solutions (M<sub>2</sub>B)
- C23a addition of Ge and Pt
- C32 addition of Ca, Sr, Ba, Ga, Ta, Mo for BaGa<sub>2</sub>, SrGa<sub>2</sub> and  
many ternary borides (MB<sub>2</sub>)
- C36 addition of Mn for Mg-Mn-Ni; improved Al<sub>2</sub>Ta

- C38 addition of Mn and Bi for  $Mn_2Sb$  with Bi solubility
- C40 addition of Ge for  $VGe_2$
- Ca addition of Mn for Mg-Mn-Ni
- Cb addition of Fe and Mo for solubility in  $Cr_2B$
- D0e addition of boron for  $Co_3B$  borides and its solutes
- D03 improved  $Li_3Sb$ ; improved Gibbs energy of metastable end-members
- D011 addition of Ge and Mo for  $Ni_3Ge$ ; improved solubility in  $Ni_3B$  of  $M_3B$  borides and borides in  $Fe_3C$
- D018 improved Gibbs energy of metastable end-members
- D019 addition of Pt for  $Mo_3Pt$ ;
- D022 improved Gibbs energy of metastable end-members
- D023 addition of Ga for  $Ga_3Zr$
- D024 addition of Au and Ga for  $Au_3Ga$
- D0a addition of Pt
- D13 addition of Ga for  $Ga_4Sr$  and  $Ga_4Ba$ ;
- D1g improved Gibbs energy of metastable end-members
- D59 addition of Si for solubility in  $B_4C$
- D510 modification to the sublattice structure for  $Zn_3As_2$
- D513 addition of Ni
- D5a addition of Ga and Pt
- D7b addition of Ga, B, Cr, Mo, Fe, V, Nb for borides (i.e.  $Zr_3B_2$ , ...)  
and  $Ga_2Hf_3$
- D85 addition of Ta, Fe, Mo for  $M_3B_4$  borides
- D8l improvement in many ternary alloy systems;
- D8I addition of Ga, Ge, Cr, Fe, B, Mo for  $M_5B_3$  borides and germanides;
- L10 addition of Mn, Pt, Zr, Co, Ni for CoPt and NiPt;
- L12 improved MnNi; improved Gibbs energy of metastable end-members
- L12 addition of Ga and Pt for  $Ga_3Sc$  and several  $Pt_3M$  end-members;
- L12c improved Gibbs energy of metastable end-members
- L21 addition of Ge and Pt for several  $Pt_3M$  end-members;  $AlPt_3$ ,  $Ni_3Ge$ ;
- L'2b improved Gibbs energy of metastable end-members
- cF96 addition of Pt for  $SiPt_2$
- cP60 addition of Pt
- hP18 addition of Ga for  $Ga_7Sr_8$ ,  $Ga_7Ba_8$
- mC18 addition of Zr and Al for  $Zr_5Ga_4$  and  $Zr_5Al_4$ ;
- mC22 addition of Hf for  $Ni_7Hf_2$
- mC22 addition of Ga for  $Ga_8Mo_3$

- mP24 restructruration of sublattices for ZnAs<sub>2</sub> and ZnP<sub>2</sub>
- mS20 addition of Pt for InPt
- mS44 addition of Pt and Ga for Ni<sub>13</sub>Ga<sub>9</sub> and Pt<sub>13</sub>In<sub>9</sub>
- oC20 addition of Fe for solubility in M<sub>2</sub>B<sub>3</sub> borides;  
improved Gibbs energy of metastable end-members
- oF40 addition of Ga for Ga<sub>3</sub>Zr
- oI6 addition of Pt, Fe, Mo for MoPt<sub>2</sub>, VPt<sub>2</sub> and solubility
- o20P addition of Au
- oP56 addition of W
- oS12 addition of Ga and Zr for ZrGa<sub>2</sub>
- tP20 addition of Ti for solubility of Al<sub>2</sub>Ti<sub>3</sub>
- tP32 improved Nb<sub>3</sub>P
- R-phase several improvements in ternary systems involving Mo, W for steels

Moreover, the following new solid solution phases were created :

- 1) BCC-B2e Prototype CsCl  
CoGa with vacancies
- 2) B19 Prototype Beta' AuCd  
Au-Ti, (Mo,V,Ti)Pt
- 3) D0c Prototype SiU3  
SiPt<sub>3</sub> is stable with Zr solubility
- 4) D82a Prototype Cu5Zn8  
V<sub>6</sub>Ga<sub>7(ht)</sub>
- 5) D8i Prototype-Mo2B5  
Mo<sub>2</sub>B<sub>5</sub>, V<sub>2</sub>B<sub>5</sub>
- 6) L11 Prototype-CuPt  
CuPt
- 7) L12e Prototype-AuCu3  
Pt<sub>3</sub>Ga, Pt<sub>3</sub>Ge, Ni<sub>3</sub>Ga
- 8) C432 Prototype Ta39Al69  
Ta<sub>39</sub>Al<sub>69</sub>
- 9) cP64 Prototype-KGe  
KGe, KSi
- 10) h22P Prototype Ti6Sn5  
β-Ti<sub>6</sub>Sn<sub>5</sub> and V<sub>6</sub>Ga<sub>5</sub>
- 11) h15R Prototype Al2Li3  
Al<sub>2</sub>Li<sub>3</sub>, Ga<sub>2</sub>Li<sub>3</sub>, In<sub>2</sub>Li<sub>3</sub>
- 12) mP86 Prototype Ta22Al21  
Ta<sub>22</sub>Al<sub>21</sub>
- 13) mS16 Prototype Pt<sub>3</sub>Ge  
Pt<sub>3</sub>Ge, Pt<sub>3</sub>Si
- 14) o12I Prototype KHg2  
Al<sub>2</sub>Sr, Ga<sub>2</sub>Sc, Cu<sub>2</sub>Pr, Cu<sub>2</sub>Sm
- 15) oP4 Prototype AuCd  
α-PtTi
- 16) oP4a Prototype Ag3Sb  
Ag<sub>3</sub>Sb with Au solubility
- 17) C2Y3 Prototype Co2Y3  
Co<sub>2</sub>Y<sub>3</sub> with REE solubility
- 18) oP24 Prototype GeAs2  
GeAs<sub>2</sub>, SiP<sub>2</sub>, SiAs<sub>2</sub>
- 19) o24P Prototype BaSi2  
(Ba,Sr)(Ge,Si)<sub>2</sub>

- 20) oP28            Prototype CeCu<sub>6</sub>  
                   CeCu<sub>6</sub>, SmCu<sub>6</sub>
- 21) oP52            Prototype Ca<sub>7</sub>Sn<sub>6</sub>  
                   Ca<sub>7</sub>Ge<sub>6</sub>, Ca<sub>7</sub>Sn<sub>6</sub>
- 22) oS8            Prototype TII  
                   CoY, GaCa, GaSc, GeSr, NiLa, NiZr, PbSr
- 23) o16S            Prototype Pt<sub>5</sub>Ga<sub>3</sub>  
                   Pt<sub>5</sub>Ga<sub>3</sub>, Ni<sub>5</sub>Ga<sub>3</sub>
- 24) i160            Prototype Cs<sub>3</sub>As<sub>2</sub>  
                   Zn<sub>3</sub>As<sub>2</sub>
- 25) tP16            Prototype Pt<sub>3</sub>Ga  
                   Pt<sub>3</sub>Ga<sub>(ht)</sub>, Pt<sub>3</sub>Al
- 26) P-phase        ternary CrMoNi Frank-Kasper solution  
                   with Fe-solubility
- 27) P1-phase        M<sub>3</sub>Ca<sub>2</sub>  
                   (Cu,Mg)<sub>3</sub>Ca<sub>2</sub>
- 28) P2-phase        MCu<sub>3</sub>  
                   (Ca,Mg)Cu<sub>3</sub>

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**Volumetric and thermal conductivity parameters were introduced for 102 pure solid phases.**

Following the 8.2 version where the volumetric properties (density at 298.15K, volumetric thermal expansivity, compressibility and derivative of the Bulk Modulus) and thermal conductivity parameters (based on the Debye Temperature, Gruneisen, and atoms per crystallographic cell) were introduced for 102 pure solid phases, the volumetric and conductivity models have been extended for more compounds.

Carbides (25 phases) – red = new in 8.3

WC	ZrC	Fe3C_S1	Ti <sub>2</sub> C	Ta <sub>2</sub> C	Nb <sub>2</sub> C	Mn <sub>7</sub> C <sub>3</sub>	Mn <sub>5</sub> C <sub>2</sub>
Fe <sub>5</sub> C <sub>2</sub> (S2)	Fe <sub>7</sub> C <sub>3</sub>	Cr <sub>23</sub> C <sub>6</sub>	Cr <sub>7</sub> C <sub>3</sub>	Cr <sub>3</sub> C <sub>2</sub>	Cr <sub>3</sub> C	BaC <sub>2</sub> (S1)	BaC <sub>2</sub> (S2)
Cr <sub>2</sub> C	SrC <sub>2</sub>	AlTi <sub>2</sub> C	AlTi <sub>3</sub> C	Fe <sub>6</sub> W <sub>6</sub> C	Mn <sub>3</sub> AlC	Mn <sub>5</sub> SiC	SiC
Al <sub>4</sub> C <sub>3</sub>							

Nitrides (24 phases)

h-BN	BN_(S2)	h-AlN	AlN_(S2)	AlN_(S3)	TiN	ZrN	InN
GaN	WN_(S1)	WN_(S2)	VN	TaN_(S1)	TaN_(S2)	Si <sub>3</sub> N <sub>4</sub>	NbN
Mn <sub>2</sub> N	Mn <sub>4</sub> N	Fe <sub>2</sub> N	Cr <sub>2</sub> N	CrN	CoN	CoN <sub>3</sub>	Nb <sub>2</sub> N

Borides (55 phases)

ZrB <sub>2</sub>	TiB <sub>2</sub>	HfB <sub>2</sub>	NbB <sub>2</sub>	NiB	Ni <sub>3</sub> B	ZrB <sub>12</sub>	ZrB
WB_(S1)	WB_(S2)	V <sub>3</sub> B <sub>4</sub>	VB	VB <sub>2</sub>	V <sub>2</sub> B <sub>3</sub>	V <sub>3</sub> B <sub>2</sub>	Ti <sub>3</sub> B <sub>4</sub>
TiB	NbB	Nb <sub>5</sub> B <sub>6</sub>	Ni <sub>2</sub> B	Ni <sub>4</sub> B <sub>3</sub>	Nb <sub>3</sub> B <sub>4</sub>	Nb <sub>2</sub> B <sub>3</sub>	Nb <sub>3</sub> B <sub>2</sub>
NbB <sub>2</sub>	Mn <sub>2</sub> B	MnB <sub>4</sub>	Mn <sub>3</sub> B <sub>4</sub>	MnB	MnB <sub>2</sub>	Fe <sub>2</sub> B	FeB
Fe <sub>3</sub> B	CrB	Cr <sub>3</sub> B <sub>4</sub>	CrB <sub>2</sub>	CoB	Co <sub>2</sub> B	Co <sub>3</sub> B	AlB <sub>2</sub>
BaB <sub>6</sub>	CaB <sub>6</sub>	Cr <sub>2</sub> B	Cr <sub>5</sub> B <sub>3</sub>	CrB <sub>4</sub>	KB <sub>6</sub>	LiB <sub>3</sub>	MgB <sub>2</sub>
MgB <sub>4</sub>	MgB <sub>7</sub>	NaB <sub>15</sub>	SrB <sub>6</sub>	W <sub>2</sub> B	AlCr <sub>2</sub> B <sub>2</sub>	AlCr <sub>3</sub> B <sub>4</sub>	

Phosphides (12 phases) – red = new in 8.3

AlP	Ca <sub>3</sub> P <sub>2</sub>	Fe <sub>2</sub> P	Fe <sub>3</sub> P	FeP	FeP <sub>2</sub>	Mg <sub>3</sub> P <sub>2</sub>	Mn <sub>2</sub> P
MnP	Ni <sub>12</sub> P <sub>5</sub>	Ni <sub>2</sub> P	Ni <sub>3</sub> P				

Oxides (9 phases) – red = new in 8.3

Al <sub>2</sub> O <sub>3</sub>	MgO	CuO	PbO	PbO <sub>2</sub>	Pb <sub>3</sub> O <sub>4</sub>		
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**Table 1a: Important phases in Al alloys present in the FTlite 8.3 Database**

Common names	FTlite		Struktur-bericht	Pearson	Space Group (index)
	Soln-Nickname or CMPD*	Solution Full name or compound stoichiometry			
Liquid metal	Liqu	Liquid	-	-	
Al, $\alpha$ , FCC, (Al), Pb	A1	FCC-A1	A1	cF4	Fm-3m (225)
Si also Ge, $\alpha$ -Sn	A4	Diamond-A4	A4	cF8	Fd-3m (227)
$\theta$ -Al <sub>2</sub> Cu	C16	C16 Prototype-Al <sub>2</sub> Cu	C16	tI12	I4/mcm (140)
$\beta$ -Al <sub>3</sub> Mg <sub>2</sub>	Beta	Beta Prototype-Mg <sub>28</sub> Al <sub>45</sub>	-	cF183 <sub>2</sub>	Fd-3m (227)
Mg <sub>2</sub> Si also Mg <sub>2</sub> Sn, Mg <sub>2</sub> Ge and Mg <sub>2</sub> Pb	C1a	aC1 Prototype-CaF <sub>2</sub>	C1	cF12	Fm-3m (225)
Al <sub>3</sub> Sc, Al <sub>3</sub> RE (Er,Tm,Yb,Lu), Al <sub>3</sub> (Cu,Li,Mg) <sub>metastable</sub>	L12	L12 Prototype-AuCu <sub>3</sub>	L1 <sub>2</sub>	cP4	Pm-3m (221)
Al <sub>3</sub> Ti(ht) also $\alpha$ -Al <sub>3</sub> Hf, Al <sub>3</sub> Nb, Al <sub>3</sub> Ta, Al <sub>3</sub> V	D022	D022 Prototype-TiAl <sub>3</sub>	D0 <sub>22</sub>	tI8	I4/mmm (139)
Al <sub>3</sub> Zr also $\beta$ -Al <sub>3</sub> Hf	D023	D023 Prototype-ZrAl <sub>3</sub>	D0 <sub>23</sub>	tI16	I4/mmm (139)
T-phase Mg <sub>32</sub> (Al,Zn) <sub>49</sub> $\tau$ -AlMgZn, $\tau$ -AlMgCu, $\tau$ -AlMgAg	D8e	Tau Prototype-Mg <sub>32</sub> (Al,Zn) <sub>49</sub>	D8 <sub>e</sub>	cI162	Im-3 (204)
V-phase Mg <sub>2</sub> Zn <sub>11</sub> also Mg <sub>2</sub> Cu <sub>6</sub> Al <sub>5</sub>	D22	V Prototype-Mg <sub>2</sub> Zn <sub>11</sub>	D2 <sub>2</sub>	cP39	Pm-3 (200)
Al <sub>6</sub> Mn, Al <sub>6</sub> (Fe,Mn)	D2h	D2h Prototype-Al <sub>6</sub> Mn	D2 <sub>h</sub>	oC28	Cmcm (63)
$\varepsilon$ -Al <sub>3</sub> Ti(lt)	tI32	tI32 Prototype-TiAl <sub>3</sub>	-	tI32	I4/mmm (139)
S-Al <sub>2</sub> CuMg	S	S Prototype-Al <sub>2</sub> CuMg	-	oC16?	Cmcm (63)
Q-Phase	-	Al <sub>7</sub> Cu <sub>3</sub> Mg <sub>6(S)</sub>			
Q-Phase	Q	Q Prototype-Th <sub>7</sub> S <sub>12</sub> (Al <sub>5</sub> Cu <sub>2</sub> Mg <sub>3</sub> Si <sub>7</sub> )			P63/m (176)
$\pi$ -Phase	-	Al <sub>8</sub> FeMg <sub>3</sub> Si <sub>6(S)</sub>			
$\alpha$ -AlFeSi or $\tau_5$ chinese script / rods	Tau5	AlFeSi_alpha	$\sim$ Al <sub>7</sub> Fe <sub>2</sub> Si	hP245	P63/mmc (194)
$\beta$ -AlFeSi or $\tau_6$ plates / acicular	Tau6	AlFeSi_beta	$\sim$ Al <sub>5</sub> FeSi		
$\gamma$ -AlFeSi or $\tau_2$	Tau2	AlFeSi_gamma	$\sim$ Al <sub>3</sub> FeSi	m?	
$\delta$ -AlFeSi	-	Al <sub>11</sub> Fe <sub>3</sub> Si <sub>6(S)</sub>			
$\alpha$ -AlMnSi	AIM1	AlMnSi_alpha	$\sim$ Al <sub>9</sub> Mn <sub>2</sub> Si		
$\beta$ -AlMnSi	AIM2	AlMnSi_alpha	$\sim$ Al <sub>9</sub> Mn <sub>3</sub> Si		

$\alpha$ -AlFeMnSi	AFMS	AlFeMnSi_alpha	$\sim$ Al <sub>18</sub> (Fe, Mn) <sub>4</sub> Si <sub>3</sub>		
Al <sub>7</sub> Cu <sub>2</sub> Fe, Al <sub>7</sub> CuMn <sub>2</sub>	E9a	E9a Prototype-Al <sub>7</sub> Cu <sub>2</sub> Fe	E9 <sub>a</sub>	tP40	P4/mnc (128)
<b>E-phase</b> Al <sub>18</sub> Cr <sub>2</sub> Mg <sub>3</sub> also Al <sub>18</sub> Mn <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> Ta <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> Ti <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> V <sub>2</sub> Mg <sub>3</sub> , Al <sub>20</sub> (Cr,Ti,V) <sub>2</sub> RE	E	E Prototype-CeCr <sub>2</sub> Al <sub>20</sub>		cF184	Fd-3m (227)
Al <sub>13</sub> Cr <sub>4</sub> Si <sub>4</sub>	-	Al <sub>13</sub> Cr <sub>4</sub> Si <sub>4(S)</sub>	-	cF84	F-43m
$\tau_3$ -Al <sub>11</sub> Mn <sub>3</sub> Zn <sub>2</sub>	-	Al <sub>11</sub> Mn <sub>3</sub> Zn <sub>2(S)</sub>		o?	
Al <sub>7</sub> Cr or Al <sub>45</sub> Cr <sub>7</sub> or Al <sub>13</sub> Cr <sub>2</sub>	-	Al <sub>7</sub> Cr <sub>(S)</sub>		mC104	C2/m (12)
'Al <sub>7</sub> V' or Al <sub>45</sub> V <sub>7</sub>	-	Al <sub>45</sub> V <sub>7(S)</sub>		mC104	C2/m (12)
Al <sub>10</sub> V	-	Al <sub>10</sub> V <sub>(S)</sub>		cF176	Fd-3m (227)
Al <sub>9</sub> Co <sub>2</sub>	D8d	D8d Prototype-Al <sub>9</sub> Co <sub>2</sub>	D8 <sub>d</sub>	mP22	P2 <sub>1</sub> /c (14)
Al <sub>4</sub> Mn, Al <sub>4</sub> (Mn,Fe) also Al <sub>4</sub> Cr	Al4M	Al4Mn	-	hP574	P6 <sub>3</sub> /mmc
<b>Al<sub>12</sub>Mn</b> also Al <sub>12</sub> Mo, Al <sub>12</sub> W	cl26	cl26 Prototype-Al <sub>12</sub> W	-	cl26	Im-3 (204)
'Al <sub>3</sub> Fe' or Al <sub>13</sub> Fe <sub>4</sub>	m102	mC102 Prototype-Al <sub>13</sub> Fe <sub>4</sub>	-	mC102	C2/m (12)
Al <sub>4</sub> Sr also Al <sub>4</sub> Ca, Al <sub>4</sub> Ba, Al <sub>4</sub> Eu	D13	D13 Prototype-Al <sub>4</sub> Ba	D1 <sub>3</sub>	tI10	I4/mmm (139)
Al <sub>2</sub> CaZn <sub>2</sub> , Al <sub>2</sub> REZn <sub>2</sub>	D13a	Al <sub>2</sub> CaZn <sub>2</sub>	D1 <sub>3</sub>	tI10	I4/mmm (139)
Al <sub>3</sub> CuRE	D13c	D13c Prototype-BaNiSn <sub>3</sub>	D1 <sub>3</sub>	tI10	I4/mmm (139)
TiB <sub>2</sub> also AlB <sub>2</sub> , CrB <sub>2</sub> , HfB <sub>2</sub> , MgB <sub>2</sub> , MnB <sub>2</sub> , MoB <sub>2</sub> , NbB <sub>2</sub> , VB <sub>2</sub> , ZrB <sub>2</sub>	C32	C32 Prototype-AlB <sub>2</sub>	C32	hP3	P6/mmm (191)
AlP also AlAs, AlIn, AlSb	B3	B3 Prototype-ZnS	B3	cF8	F-43m (216)
AlLi	B32	B32 Prototype-NaTl	B32	cF16	Fd-3m (227)
<b>Al<sub>4</sub>C<sub>3</sub></b>	D71	D71 Prototype-Al <sub>4</sub> C <sub>3</sub>	D7 <sub>1</sub>	hR7	R-3m (166)
Al <sub>11</sub> RE <sub>3</sub> (La,Ce,Pr,Nd)	oI28	oI28 Prototype-Al <sub>11</sub> La <sub>3</sub>	-	oI28	Immm
Al <sub>3</sub> Gd, Al <sub>3</sub> Sm	D019	D019 Prototype-Ni <sub>3</sub> Sn	D0 <sub>19</sub>	hP8	P6 <sub>3</sub> /mmc (194)
Al <sub>3</sub> Ni, Fe <sub>3</sub> C-cementite, Ni <sub>3</sub> B	D011	D011 Prototype-Fe <sub>3</sub> C	D0 <sub>11</sub>	oP16	Pnma (62)
Al <sub>3</sub> Dy, Ni <sub>3</sub> Ti	D024	D024 Prototype-Ni <sub>3</sub> Ti	D0 <sub>24</sub>	hP16	P6 <sub>3</sub> /mmc (194)
Al <sub>3</sub> Ho	hR60	hR60 Prototype_HoAl <sub>3</sub>		hR60	R-3m (166)
Be	A3	HCP-A3	A3	hP2	P6 <sub>3</sub> /mmc (194)
Zn or (Zn)	A3"	HCP-ZN Prototype-Mg	A3	hP2	P6 <sub>3</sub> /mmc (194)
$\beta$ -Sn	A5	BCT-A5 Prototype-Sn	A5	tI4	I4 <sub>1</sub> /amd (141)
Bi	A7	RHOM-A7 Prototype-As	A7	hR2	R-3m (166)
Al <sub>9</sub> (Co,Fe,Ni) <sub>2</sub> (Sr,Ba)	hP12	hP12 Prototype-Al <sub>9</sub> Co <sub>2</sub> Sr		hP12	P6/mmm (191)

<i>REFe<sub>2</sub>Al<sub>10</sub></i>	<i>oC52</i>	<i>oC52 Prototype-YbFe<sub>2</sub>Al<sub>10</sub></i>		<i>oC52</i>	<i>Cmcm (63)</i>
<i>Al<sub>3</sub>Tb</i>	<i>A3RE</i>	<i>hR12 Prototype-BaPb<sub>3</sub></i>		<i>hR12</i>	<i>R-3m (166)</i>
<i>AlMgAg</i>	-	<i>AlMgAg<sub>(s)</sub></i>	<i>C14</i>	<i>hP12</i>	<i>P6<sub>3</sub>/mmc (191)</i>
<i>AlNaSi</i>	-	<i>AlNaSi<sub>(s)</sub></i>	-		<i>P4/nmm</i>
<i>Al<sub>21</sub>Pt<sub>5</sub></i>	-		-	<i>cF416</i>	<i>F-43m (216)</i>
<i>Al<sub>2</sub>Si<sub>2</sub>Sr</i>	<i>hP5</i>	<i>hP5 Prototype-Ce<sub>2</sub>SO<sub>2</sub></i>	-	<i>hP5</i>	<i>P-3m1 (164)</i>
<i>Al<sub>2</sub>Si<sub>2</sub>Ba</i>	-	<i>BaAl<sub>2</sub>Si<sub>2(s)</sub></i>	-	<i>o?</i>	<i>Pnma</i>
<i>Al<sub>2</sub>Si<sub>2</sub>Sc</i>	-	<i>ScSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>La</i>	-	<i>LaSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Pr</i>	-	<i>PrSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Nd</i>	-	<i>NdSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Sm</i>	-	<i>SmSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Eu</i>	-	<i>EuSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Gd</i>	-	<i>GdSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Tb</i>	-	<i>TbSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Dy</i>	-	<i>DySi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Ho</i>	-	<i>HoSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>2</sub>Si<sub>2</sub>Yb</i>	-	<i>YbSi<sub>2</sub>Al<sub>2(s)</sub></i>	-		
<i>Al<sub>3</sub>Si<sub>2</sub>Er</i>	-	<i>ErSi<sub>2</sub>Al<sub>3(s)</sub></i>	-	<i>mS14</i>	<i>C2/m</i>
<i>AlCeSi<sub>2</sub></i>	-	<i>AlCeSi<sub>2(s)</sub></i>	<i>h?</i>		
<i>ScCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>ScCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>YCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>YCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>LaCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>LaCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>CeCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>CeCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>PrCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>PrCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>NdCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>NdCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>SmCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>SmCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>EuCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>EuCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>GdCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>GdCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>TbCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>TbCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>DyCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>DyCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>HoCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>HoCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>
<i>ErCu<sub>4</sub>Al<sub>8</sub></i>	-	<i>ErCu<sub>4</sub>Al<sub>8(s)</sub></i>		<i>tl26</i>	<i>I4/mmm (139)</i>

TmCu <sub>4</sub> Al <sub>8</sub>	-	TmCu <sub>4</sub> Al <sub>8(s)</sub>		tI26	I4/mmm (139)
YbCu <sub>4</sub> Al <sub>8</sub>	-	YbCu <sub>4</sub> Al <sub>8(s)</sub>		tI26	I4/mmm (139)
LuCu <sub>4</sub> Al <sub>8</sub>	-	LuCu <sub>4</sub> Al <sub>8(s)</sub>		tI26	I4/mmm (139)
Al <sub>8</sub> EuFe <sub>2</sub>	-	Al <sub>8</sub> EuFe <sub>2(s)</sub>		oP44	Pbam (55)
Al <sub>8</sub> CrY	-	Al <sub>8</sub> CrY <sub>(s)</sub>	-		
Al <sub>2</sub> O <sub>3</sub> $\alpha$ -corundum	-	Al <sub>2</sub> O <sub>3(s4)</sub>	D51	hR30	R-3c (167)
MgO magnesia	-	MgO <sub>(s)</sub>	B1	cF8	Fm-3m (225)
MgAl <sub>2</sub> O <sub>4</sub> spinel	-	MgAl <sub>2</sub> O <sub>4(s)</sub>	H11	cF56	Fd-3m (227)

**Table 1b: Important phases in Mg alloys present in the FTlite 8.3 Database**

Common names	FTlite		Struktur-bericht	Pearson	Space Group (index)
	Soln-Nickname or CMPD*	Solution Full name or compound stoichiometry			
Liquid metal	Liqu	Liquid	-	-	
Mg, $\alpha$ , HCP, (Mg), $\alpha$ -Ti, $\alpha$ -Zr, $\alpha$ -Be	A3	HCP-A3	A3	hP2	P6 <sub>3</sub> /mmc (194)
$\gamma$ -Fe <sub>(austenite)</sub> , InMg, $\gamma$ -Mn	A1	FCC-A1	A1	cF4	Fm-3m (225)
$\alpha$ -Fe <sub>(ferrite)</sub> , Cr, Li, K, Na, $\beta$ -Be, $\delta$ -Mn, $\beta$ -Sc, V	A2	BCC-A2	A2	cI2	Im-3m (229)
C <sub>(graphite)</sub>	A9	A9 Prototype-C	A9	hP4	P6 <sub>3</sub> /mmc (194)
$\alpha$ -Mn, $\beta$ -Al <sub>12</sub> Mg <sub>17</sub> , Mg <sub>24</sub> RE <sub>5</sub> (RE = Y, Tb, Dy, Ho, Er, Tm, Lu)	A12	CBCC-A12 Prototype-Mn	A12	cI58	I-43m (217)
$\beta$ -Mn	A13	CUB-A13 Prototype-Mn	A13	cP20	P4 <sub>1</sub> 3 <sub>2</sub> (213)
MgSc, AlSc	B2_a	BCC-B2 Prototype-CsCl	B2	cP2	Pm-3m (221)
AlNi	B2_c	BCC-B2c Prototype-CsCl	B2	cP2	Pm-3m (221)
Mg <sub>2</sub> Si also Mg <sub>2</sub> Sn, Mg <sub>2</sub> Ge and Mg <sub>2</sub> Pb	C1a	aC1 Prototype-CaF <sub>2</sub> (anti-fluorite)	C1	cF12	Fm-3m (225)
AlCr <sub>2</sub>	C11b	C11b Prototype-MoSi <sub>2</sub>	C11 <sub>b</sub>	tI6	I4/mmm (139)
Mg <sub>2</sub> Ca, Mg <sub>2</sub> Yb, Li <sub>2</sub> Ca, (Al,Mg) <sub>2</sub> Lu	C14	C14 Prototype-MgZn <sub>2</sub>	C14	hP12	P6 <sub>3</sub> /mmc (194)
Na <sub>2</sub> K	-	Na <sub>2</sub> K <sub>(S)</sub>	C14	hP12	P6 <sub>3</sub> /mmc (194)
Al <sub>2</sub> Ca, Al <sub>2</sub> Ba, (Al,Cu) <sub>2</sub> Mg, Al <sub>2</sub> RE (RE = Sc, Y, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Tm, Er, Yb, Lu)	C15	C15 Prototype-MgCu <sub>2</sub>	C15	cF24	Fd-3m (227)
Al <sub>4</sub> MgY	-	Al <sub>4</sub> MgY <sub>(S)</sub>	C15	cF24	Fd-3m (227)
(Al,Mg) <sub>2</sub> Ca	C36	C36 Prototype-MgNi <sub>2</sub>	C36	hP24	P6 <sub>3</sub> /mmc (194)
Mg <sub>2</sub> Cu	Cb	Cb Prototype-CuMg <sub>2</sub>	C <sub>b</sub>	oF48	Fddd (70)
AlTi	L10	L10 Prototype-AuCu	L1 <sub>0</sub>	tP2	P4/mmm (123)
Al <sub>3</sub> Sc, Al <sub>3</sub> RE (Er, Tm, Yb, Lu)	L12	L12 Prototype-AuCu <sub>3</sub>	L1 <sub>2</sub>	cP4	Pm-3m (221)
Mg <sub>7</sub> Zn <sub>3</sub> or Mg <sub>51</sub> Zn <sub>20</sub>	-	Mg <sub>51</sub> Zn <sub>20(S)</sub>	-	oI142	Immm
MgZn or Mg <sub>12</sub> Zn <sub>13</sub>	MgZn	Mg <sub>12</sub> Zn <sub>13</sub>	-	-	-
Mg <sub>17</sub> Ba <sub>2</sub>	-	Mg <sub>17</sub> Ba <sub>2(S)</sub>	-	hP57	R3m
AgMg <sub>4</sub>	hP_G	AgMg4 gamma_hp	-	h?	?

$\text{Ag}_{17}\text{Mg}_{54}$	A17M	$\text{oP}142$ Prototype- $\text{Hf}54\text{Os}17$		$\text{oP}142$	I <sub>mmm</sub> (71)
$\text{AgMg}_3$	D018	D018 Prototype- $\text{Na}3\text{As}$	D0 <sub>18</sub>	hP24	P <sub>6<sub>3</sub></sub> cm (185)
$\text{Al}_3RE$ ( $RE = \text{Y}, \text{La}, \text{Ce}, \text{Pr}, \text{Nd}, \text{Sm}$ ), $\text{Ti}_3\text{Al}$	D019	D019 Prototype- $\text{Ni}3\text{Sn}$	D0 <sub>19</sub>	hP8	P <sub>6<sub>3</sub></sub> /mmc (194)
$\text{Al}_3\text{Ti}$ also $\alpha\text{-Al}_3\text{Hf}$ , $\text{Al}_3\text{Nb}$ , $\text{Al}_3\text{Ta}$ , $\text{Al}_3\text{V}$	D022	D022 Prototype- $\text{TiAl}3$	D0 <sub>22</sub>	tI8	I <sub>4/mmm</sub> (139)
$\text{Al}_3\text{Zr}$ also $\beta\text{-Al}_3\text{Hf}$	D023	D023 Prototype- $\text{ZrAl}3$	D0 <sub>23</sub>	tI16	I <sub>4/mmm</sub> (139)
$\sigma\text{-FeCr}$ , $\sigma\text{-AlNb}$ , $\sigma\text{-AlTa}$	D8b	Sigma Prototype- $\text{FeCr}$	D8 <sub>b</sub>	tP30	P <sub>4<sub>2</sub></sub> /mnm (136)
<b>T-phase</b> $\text{Mg}_{32}(\text{Al}, \text{Zn})_{49}$ or $\tau\text{-AlMgZn}$	D8e	Tau Prototype- $\text{Mg}32(\text{Al}, \text{Zn})49$	D8 <sub>e</sub>	cI162	I <sub>m-3</sub> (204)
$\text{Mg}_{12}RE$ ( $RE = \text{La}, \text{Ce}, \text{Pr}$ )	D2b	D2b Prototype- $\text{Mn}12\text{Th}$	D2 <sub>b</sub>	tI26	I <sub>4/mmm</sub> (139)
<b>Al<sub>6</sub>Mn</b> , <b>Al<sub>6</sub>(Fe,Mn)</b>	D2h	D2h Prototype- $\text{Al}6\text{Mn}$	D2 <sub>h</sub>	oC28	Cmcm (63)
<b>S-phase</b> $\text{Al}_2\text{CuMg}$	S	S Prototype- $\text{Al}2\text{CuMg}$	-	oS16	Cmcm (63)
<b>ϕ-Phase</b> or $(\text{Al}, \text{Zn})_5\text{Mg}_6$	Phi	Phi $\text{AlMgZn}$	-	o?	Pbcm (57)
<b>Q-Phase</b>	-	$\text{Al}_7\text{Cu}_3\text{Mg}_{6(s)}$			
<b>Q-Phase</b>	Q	Q Prototype- $\text{Th}7\text{S}12$ $\text{Al}_5\text{Cu}_2\text{Mg}_7\text{Si}_7$			P6 <sub>3/m</sub> (176)
<b>π-Phase</b>	-	$\text{Al}_8\text{FeMg}_3\text{Si}_{6(s)}$			
$\text{Sr}_4\text{Al}_{38}\text{Mg}_{58}$	-	$\text{Sr}_4\text{Al}_{38}\text{Mg}_{58(s)}$			
$\alpha\text{-AlFeSi}$ or $\tau_5$ chinese script / rods	Tau5	AlFeSi_alpha	$\sim\text{Al}_7\text{Fe}_2\text{Si}$	hP245	P6 <sub>3/mmc</sub> (194)
$\beta\text{-AlFeSi}$ or $\tau_6$ plates / acicular	mC52	AlFeSi_beta	$\sim\text{Al}_5\text{FeSi}$	mC52	C <sub>2/c</sub> (15)
$\gamma\text{-AlFeSi}$ or $\tau_2$	Tau2	AlFeSi_gamma	$\sim\text{Al}_3\text{FeSi}$	m?	
$\delta\text{-AlFeSi}$	-	$\text{Al}_{11}\text{Fe}_3\text{Si}_{6(s)}$			
$\alpha\text{-AlMnSi}$ or $\tau_9$	AlM1	AlMnSi_alpha	$\sim\text{Al}_9\text{Mn}_2\text{Si}$	cP138	Pm-3 (200)
$\beta\text{-AlMnSi}$	hP26	AlMnSi_beta	$\sim\text{Al}_9\text{Mn}_3\text{Si}$	hP26	P6 <sub>3/mmc</sub> (194)
$\alpha\text{-AlFeMnSi}$	AFMS	AlFeMnSi_alpha	$\sim\text{Al}_{18}(\text{Fe}, \text{Mn})_4\text{Si}_3$		
$\text{Al}_7\text{Cu}_2\text{Fe}$ , $\text{Al}_7\text{CuMn}_2$	E9a	E9a Prototype- $\text{Al}7\text{Cu}2\text{Fe}$	E9 <sub>a</sub>	tP40	P4/mnc (128)
<b>E-phase</b> $\text{Al}_{18}\text{Cr}_2\text{Mg}_3$ also $\text{Al}_{18}\text{Mn}_2\text{Mg}_3$ , $\text{Al}_{18}\text{Ta}_2\text{Mg}_3$ , $\text{Al}_{18}\text{Ti}_2\text{Mg}_3$ , $\text{Al}_{18}\text{V}_2\text{Mg}_3$	E	E Prototype- $\text{CeCr}2\text{Al}20$	-	cF184	Fd-3m (227)
$\text{Mg}_2\text{Ni}$	Ca	Ca Prototype- $\text{NiMg}2$	C <sub>a</sub>	hP18	P <sub>6<sub>2</sub>2<sub>2</sub></sub>
$\text{Mg}_5RE$ ( $RE = \text{Gd}, \text{Sm}$ )	Mg5X	cF448 Prototype- $\text{Mg}5\text{Gd}$	-	cF448	F-43m (216)
$\text{Mg}_{41}RE_5$ ( $RE = \text{Nd}, \text{Sm}$ )	tI92	tI92 Prototype- $\text{Mg}41\text{Ce}5$	-	tI92	I <sub>4/m</sub> (87)
$\text{Mg}_{17}\text{Sr}_2$ , $\text{Mg}_{17}\text{Sc}_2$ , $\text{Mg}_{17}\text{Eu}_2$ , $\text{Mg}_{17}\text{Pm}_2$ , $\text{Mg}_{17}\text{Tm}_2$	hP38	hP38 Prototype- $\text{Ni}17\text{Th}2$	-	hP38	P <sub>6<sub>3</sub></sub> /mmc (194)

$\alpha\text{-Mg}_3\text{Bi}_2$ , $\alpha\text{-Mg}_3\text{Sb}_2$	P3m1	alpha_Mg3X2(low-T)	-	-	P-3m1
Al <sub>3</sub> Ni <sub>2</sub>	D513	D513 Prototype-Al3Ni2	D5 <sub>13</sub>	hP5	P-3m1 (164)
Al <sub>8</sub> V <sub>5</sub>	-	Al <sub>8</sub> V <sub>5(S)</sub>	D8 <sub>2</sub>	cI52	I-43m (217)
Al <sub>8</sub> Mn <sub>5</sub>	D810	D810 Prototype-Al8Cr5	D8 <sub>10</sub>	hR26	R3m (160)
Al <sub>13</sub> Cr <sub>4</sub> Si <sub>4</sub>	-	Al <sub>13</sub> Cr <sub>4</sub> Si <sub>4(S)</sub>	-	cF84	F-43m
$\tau_3\text{-Al}_{11}\text{Mn}_3\text{Zn}_2$	-	Al <sub>11</sub> Mn <sub>3</sub> Zn <sub>2(S)</sub>		o?	
$\beta\text{-Al}_8\text{Cr}_5$ (HT)	-	Al <sub>8</sub> Cr <sub>5(S)1</sub>			
$\alpha\text{-Al}_8\text{Cr}_5$ (LT)	-	Al <sub>8</sub> Cr <sub>5(S)2</sub>	D8 <sub>10</sub>	hR26	R3m (160)
$\alpha\text{-Al}_9\text{Cr}_4$	-	Al <sub>9</sub> Cr <sub>4(S)</sub>	-		
$\beta\text{-Al}_9\text{Cr}_4$	-	Al <sub>9</sub> Cr <sub>4(S)2</sub>	-	cI52	I-43m
'Al <sub>7</sub> Cr' or Al <sub>45</sub> Cr <sub>7</sub> or Al <sub>13</sub> Cr <sub>2</sub>	-	Al <sub>13</sub> Cr <sub>2(S)</sub>		mC104	C2/m
'Al <sub>7</sub> V' or Al <sub>45</sub> V <sub>7</sub>	-	Al <sub>7</sub> V <sub>(S)</sub>		mC104	C2/m
Al <sub>10</sub> V	-	Al <sub>10</sub> V <sub>(S)</sub>		cF?	Fd-3m
$\alpha\text{-Al}_3\text{Ta}_2$ (It)	-	Al <sub>3</sub> Ta <sub>2(S)2</sub>			
Al <sub>9</sub> Co <sub>2</sub>	D8d	D8d Prototype-Al9Co2	D8 <sub>d</sub>	mP22	P2 <sub>1</sub> /c (14)
Al <sub>4</sub> Mn, Al <sub>4</sub> (Mn,Fe) also Al <sub>4</sub> Cr	Al4M	Al4Mn	-	hP574	P6 <sub>3</sub> /mmc
Al <sub>11</sub> Mn <sub>4</sub>	aP15	aP15 Prototype-Al11Mn4		aP15	P-1 (2)
Al <sub>12</sub> Mn also Al <sub>12</sub> Mo, Al <sub>12</sub> W	cI26	cI26 Prototype-Al12W	-	cI26	Im-3 (204)
'Al <sub>2</sub> Fe' or Al <sub>61</sub> Fe <sub>31</sub>	-	Al <sub>61</sub> Fe <sub>31(S)</sub>		aP18	P1-1
$\eta\text{-Al}_5\text{Fe}_2$	Eta	Eta Prototype-Al5Fe2	-	oS24	Cmcm (63)
'Al <sub>3</sub> Fe' or Al <sub>13</sub> Fe <sub>4</sub>	m102	mC102 Prototype-Al13Fe4	-	mC102	C2/m (12)
Al <sub>4</sub> Sr, Al <sub>4</sub> Ba, Al <sub>4</sub> Eu	D13	D13 Prototype-Al4Ba	D1 <sub>3</sub>	tI10	I4/mmm (139)
TiB <sub>2</sub> also AlB <sub>2</sub> , CrB <sub>2</sub> , HfB <sub>2</sub> , MgB <sub>2</sub> , ZrB <sub>2</sub>	C32	C32 Prototype-AlB2	C32	hP3	P6/mmm (191)
AlLi, InLi, ZnLi, InNa	B32	B32 Prototype-NaTl	B32	cF16	Fd-3m (227)
Al <sub>4</sub> C <sub>3</sub>	D71	D71 Prototype-Al4C3	D7 <sub>1</sub>	hR7	R-3m (166)
Al <sub>11</sub> RE <sub>3</sub> (La,Ce,Pr,Nd)	oI28	oI28 Prototype-Al11La3	-	oI28	Immm (71)
Al <sub>3</sub> Ni, Fe <sub>3</sub> C-cementite	D011	D011 Prototype-Fe3C	D0 <sub>11</sub>	oP16	Pnma (62)
Al <sub>3</sub> Dy	D024	D024 Prototype-Ni3Ti	D0 <sub>24</sub>	hP16	P6 <sub>3</sub> /mmc (194)
Al <sub>3</sub> Ho	hR60	hR60 Prototype-HoAl3		hR60	R-3m (166)
MgBe <sub>13</sub>	-	MgBe <sub>13(S)</sub>	D2 <sub>3</sub>	cF112	Fm-3c
Al <sub>2</sub> Ti	-	Al <sub>2</sub> Ti <sub>(S)</sub>		tI24	I4 <sub>1</sub> /amd (141)

$\text{Mg}_3\text{P}_2$	-	$\text{Mg}_3\text{P}_{2(\text{s})}$			
MgO magnesia	-	$\text{MgO}_{(\text{s})}$	B1	cF8	Fm-3m (225)
MgAl <sub>2</sub> O <sub>4</sub> spinel	-	$\text{MgAl}_2\text{O}_{4(\text{s})}$	H11	cF56	Fd-3m (227)
Mg <sub>5</sub> Ga <sub>2</sub>	D8g	D8g Prototype-Ga <sub>2</sub> Mg <sub>5</sub>	D8 <sub>g</sub>	oI28	Ibam (72)

**Table 1c: Important phases in Ti alloys present in the FTlite 8.3 Database**

Common names	FTlite		Struktur-bericht	Pearson	Space Group (index)
	Soln-Nickname or CMPD*	Solution Full name or compound stoichiometry			
Liquid metal	Liqu	Liquid	-	-	
Ti, $\alpha$ , HCP, (Ti), Dy, Er, Gd, Hf, Ho, Lu, Mg, Sc, Tb, Tm, Y, $\alpha$ -Zr	A3	HCP-A3	A3	hP2	P6 <sub>3</sub> /mmc (194)
Ti, $\beta$ , BCC, (Ti), Ba, Ca, Ce, Dy, Eu, Hf, K, Li, Mo, Na, Nb, Sr, Ta, V, W, Yb	A2	BCC-A2	A2	cI2	Im-3m (229)
TiC, TiN, $\beta$ -TiO, Ca, Ce, La, Sr, Yb	A1	FCC-A1	A1/B1	cF4	Fm-3m (225)
La, Nd, Pr, Au	A3'	DHCP-A3	A3'	hP4	P6 <sub>3</sub> /mmc (194)
C <sub>(graphite)</sub>	A9	A9 Prototype-C	A9	hP4	P6 <sub>3</sub> /mmc (194)
Mg <sub>24</sub> RE <sub>5</sub> (RE = Y, Tb, Dy, Ho, Er, Tm, Lu)	A12	CBCC-A12 Prototype-Mn	A12	cI58	I-43m (217)
Ti <sub>3</sub> Au, Ti <sub>3</sub> Pt	A15	CUB-A15 Prototype-Cr <sub>3</sub> Si	A15	cP8	Pm-3n (223)
FeTi, AgMg, AlFe, AlMn	B2	BCC-B2 Prototype-CsCl	B2	cP2	Pm-3m (221)
MgSc	B2_a	BCC-B2 Prototype-CsCl	B2	cP2	Pm-3m (221)
ZnTi	-	ZnTi <sub>(s)</sub>	B2	cP2	Pm-3m (221)
(Ag,Cu)Ti	B11	B11 Prototype-CuTi	B11	tP4	P4/nmm (129)
TiSi, TiB	B27	B27 Prototype-FeB	B27	oP8	Pnma (62)
Ti <sub>2</sub> Ga, Zr <sub>2</sub> Al, Ti <sub>2</sub> Sn	B8x	B81/B82 NiAs/InNi2	B8 <sub>1</sub> /B8 <sub>2</sub>	hP6	P6 <sub>3</sub> /mmc (194)
$\delta$ -TiH <sub>2</sub>	C1	C1 Prototype-CaF2	C1	cF12	Fm-3m (225)
Rutile TiO <sub>2</sub>	-	TiO <sub>2(s)</sub>	C4	tP6	P4 <sub>2</sub> /mnmm (136)
Ti <sub>2</sub> N	-	Ti <sub>2</sub> N <sub>(s)</sub>	C4	tP6	P4 <sub>2</sub> /mnmm (136)
Anatase TiO <sub>2</sub>	-	TiO <sub>2(s2)</sub>	C5	tI12	I4 <sub>1</sub> /amd (141)
(Ag,Cu,Zn)Ti <sub>2</sub>	C11b	C11b Prototype-MoSi2	C11 <sub>b</sub>	tI6	I4/mmm (139)
Cr <sub>2</sub> Ti, Mn <sub>2</sub> Ti, Mg <sub>2</sub> Ca, Mg <sub>2</sub> Yb, Li <sub>2</sub> Ca	C14	C14 Prototype-MgZn2	C14	hP12	P6 <sub>3</sub> /mmc (194)
Na <sub>2</sub> K	-	Na <sub>2</sub> K <sub>(s)</sub>	C14	hP12	P6 <sub>3</sub> /mmc (194)
Cr <sub>2</sub> Ti	C15	C15 Prototype-MgCu2	C15	cF24	Fd-3m (227)
Al <sub>4</sub> MgY	-	Al <sub>4</sub> MgY <sub>(s)</sub>	C15	cF24	Fd-3m (227)
TiB <sub>2</sub> , MB <sub>2</sub> (M = Al, Cr, Hf, Mg, Mn, Nb, Ti, V, Zr)	C32	C32 Prototype-AlB2	C32	hP3	P6/mmm (191)

Cr <sub>2</sub> Ti	C36	C36 Prototype-MgNi2	C36	hP24	P6 <sub>3</sub> /mmc (194)
<b>TiSi<sub>2</sub>, ZrSn<sub>2</sub></b>	<b>C54</b>	C54 Prototype-TiSi2	C54	oF24	Fddd (70)
<b><math>\gamma</math>-TiAl</b>	<b>L10</b>	L10 Prototype-AuCu	L1 <sub>0</sub>	tP2	P4/mmm (123)
Mg <sub>17</sub> Ba <sub>2</sub>	-	Mg <sub>17</sub> Ba <sub>2(s)</sub>	-	hP57	R3m
AgMg <sub>4</sub>	hP_G	AgMg4 gamma_hp	-	h?	?
Ag <sub>17</sub> Mg <sub>54</sub>	A17M	oP142 Prototype-Hf54Os17		oP142	Immm (71)
AgMg <sub>3</sub>	D018	D018 Prototype-Na3As	D0 <sub>18</sub>	hP24	P6 <sub>3</sub> cm (185)
<b><math>\alpha_2</math>-Ti<sub>3</sub>Al, Ti<sub>3</sub>Ga, Ti<sub>3</sub>Sn</b>	<b>D019</b>	D019 Prototype-Ni3Sn	D0 <sub>19</sub>	hP8	P6 <sub>3</sub> /mmc (194)
Al <sub>3</sub> Ti also Al <sub>3</sub> Hf, Al <sub>3</sub> Nb, Al <sub>3</sub> Ta, Al <sub>3</sub> V	D022	D022 Prototype-TiAl3	D0 <sub>22</sub>	tI8	I4/mmm (139)
Mg <sub>12</sub> RE (RE = La,Ce,Pr)	D2b	D2b Prototype-Mn12Th	D2 <sub>b</sub>	tI26	I4/mmm (139)
<b>Ti<sub>3</sub>B<sub>4</sub>, M<sub>3</sub>B<sub>4</sub> (M = Cr, Hf, Mn, Nb, Ti, V)</b>	<b>D7b</b>	D7b Prototype-Ta3B4	D7 <sub>b</sub>	oI14	Immm (71)
<b>Ti<sub>5</sub>Si<sub>3</sub></b>	<b>D88</b>	D88 Prototype-Mn5Si3	D8 <sub>8</sub>	hP16	P6 <sub>3</sub> /mcm (193)
Ti <sub>2</sub> O <sub>3</sub>	-	Ti <sub>2</sub> O <sub>3(s)</sub>	D5 <sub>1</sub>	hR30	R-3c (167)
Ti <sub>3</sub> O <sub>2</sub>	-	Ti <sub>3</sub> O <sub>2(s)</sub>	-	hR30	R-3c (167)
Al <sub>7</sub> Cu <sub>2</sub> Fe, Al <sub>7</sub> CuMn <sub>2</sub>	E9a	E9a Prototype-Al7Cu2Fe	E9 <sub>a</sub>	tP40	P4/mnc (128)
(Co,Ni)Ti <sub>2</sub>	cF96	cF96 Prototype-Ti2Ni	-	cF96	Fd-3m (227)
Mg <sub>5</sub> RE (RE = Gd,Sm)	Mg5X	cF448 Prototype-Mg5Gd	-	cF448	F-43m (216)
Mg <sub>41</sub> RE <sub>5</sub> (RE = Nd,Sm)	tI92	tI92 Prototype-Mg41Ce5	-	tI92	I4/m (87)
Mg <sub>17</sub> Sr <sub>2</sub> , Mg <sub>17</sub> Sc <sub>2</sub> , Mg <sub>17</sub> Eu <sub>2</sub> , Mg <sub>17</sub> Pm <sub>2</sub> , Mg <sub>17</sub> Tm <sub>2</sub>	hP38	hP38 Prototype-Ni17Th2	-	hP38	P6 <sub>3</sub> /mmc (194)
$\alpha$ -Mg <sub>3</sub> Bi <sub>2</sub> , $\alpha$ -Mg <sub>3</sub> Sb <sub>2</sub>	P3m1	alpha_Mg3X2(low-T)	-	-	P-3m1
$\beta$ -MnTi	-	"MnTi <sub>(s)</sub> "	-	tP30	(136)
<b>Ti<sub>3</sub>Si, M<sub>3</sub>Si (M = Hf,Nb, Ta,Ti,Zr)</b>	<b>tP32</b>	tP32 Prototype-Ti3P	-	tP32	P4 <sub>2</sub> /n (86)
Sm	-	Sm <sub>(s)</sub>	-	hR9	R-3m (166)
$\alpha$ -TiO	-	TiO <sub>(s)</sub>	-	mS20	C2/m (12)
<b>Al<sub>2</sub>O<sub>3</sub> <math>\alpha</math>-corundum</b>	-	<b>Al<sub>2</sub>O<sub>3(s4)</sub></b>	D51	hR30	R-3c (167)
MgO magnesia	-	MgO <sub>(s)</sub>	B1	cF8	Fm-3m (225)
MgAl <sub>2</sub> O <sub>4</sub> spinel	-	MgAl <sub>2</sub> O <sub>4(s)</sub>	H11	cF56	Fd-3m (227)
Mg <sub>5</sub> Ga <sub>2</sub>	D8g	D8g Prototype-Ga2Mg5	D8 <sub>g</sub>	oI28	Ibam (72)

**Table 2: Solution names with their major components (not including minor solutes)**

- **Important solutions in Al alloys**
- **Important solutions in Mg alloys**
- **Important solutions in Ti alloys**
- **Important solutions in Al and Mg alloys**

Common names	FTlite		Major phase components
	Nickname	Full name	
Liquid metal	Liqu	Liquid	most elements
Al, FCC, TiC, TiN, TiO	A1	FCC-A1	Ag, Al, Cu, $\gamma$ -Fe <sub>austenite</sub> , Ni, Pb, Pt, ... with C, H, N, O interstitials ( $\rightarrow$ B1)
BCC, $\beta$ -Ti	A2	BCC-A2	$\alpha$ -Fe <sub>ferrite</sub> , Cr, K, Li, Na, $\beta$ -Ti, $\beta$ -Zr, Mo, Nb, Ta, V... ... with C, H, N, O interstitials
Mg, HCP, $\alpha$ -Ti	A3	HCP-A3	Mg, $\alpha$ -Ti, $\alpha$ -Zr, Co, Rare-earth elements... ... with C, H, N, O interstitials
DHCP	A3'	DHCP-A3'	Rare-earth elements, Au(Sn)
Zn	A3''	HCP-Zn	Zn with 17 solutes
Si	A4	Diamond-A4	C <sub>diamond</sub> , Ge, Si, $\alpha$ -Sn with 13 solutes
Sn	A5	BCT-A5	$\beta$ -Sn with 16 solutes
In	A6	TET-A6	$\alpha$ -In, with 12 solutes
InSn	A6o	TET-A6o	$\beta$ -InSn with 5 solutes
As, Bi, Sb	A7	RHOM-A7	$\alpha$ -As, Bi, Sb with 7 solutes
C, graphite	A9	A9	C <sub>graphite</sub> with boron
Ga	A11	ORTH-A11	$\alpha$ -Ga with 2 solutes
$\alpha$ -Mn, Mg <sub>24</sub> RE <sub>5</sub> , ( $\gamma$ - or $\beta$ -) Al <sub>12</sub> Mg <sub>17</sub>	A12	CBCC-A12	$\alpha$ -Mn, $\beta$ / $\gamma$ -Al <sub>12</sub> Mg <sub>17</sub> , $\chi$ -FeCr(Mo,W) Mg <sub>24</sub> RE <sub>5</sub> (with RE = Y,Tb,Dy,Ho,Er,Tm,Lu)
$\beta$ -Mn	A13	CUB-A13	$\beta$ -Mn, Ag <sub>3</sub> Al with 25 solutes
Cr <sub>3</sub> Si	A15	CUB-A15	Cr <sub>3</sub> Si, Cr <sub>3</sub> Ga, Cr <sub>3</sub> Pt, Mo <sub>3</sub> Al, Nb <sub>3</sub> Al, Nb <sub>3</sub> Sn, Ta <sub>3</sub> Sn, V <sub>3</sub> Al, V <sub>3</sub> Co, V <sub>3</sub> Ga, V <sub>3</sub> Ni, V <sub>3</sub> Si, V <sub>3</sub> Sn
$\gamma$ -InSn	Af	HEX-Af	$\gamma$ -InSn with Ag solubility
$\beta$ -B, boron, A <sub>g</sub>	Ag	Ag	$\beta$ -B-tetragonal with carbon solubility
$\alpha$ -B	aRho	aRHOM	$\alpha$ -B with carbon solubility
$\beta$ -B	bRho	bRHOM	$\beta$ -B-rhombohedral with 4 solutes
P(red)	RedP	Red-Phosphorus	P(red) with As solubility
RESb	B1	FCC-B1	CeSb, NdSb, PrSb, SmSb, MS (M=Ca,Mg,Mn,Pb)
ordered BCC	B2_a	BCC-B2a	AlSc, ZnRE and MgRE (with RE = Y,Sc,La,Ce, Pr,Nd,Sm,Eu, Gd,Tb,Dy,Ho,Tm,Er,Yb,Lu), CoSc
AlFe, AlMn, FeSi, AgLi, AgMg, CaIn, $\beta$ -NiZn, $\beta$ '-CuZn	B2	BCC_B2!BCC_A2	AgLi, AgMg, CaIn, $\beta$ -NiZn, $\beta$ '-CuZn with order/disorder transf.

TiNi	B2_b	BCC-B2b	non-stoichiometric TiNi with Au, Pt
AlNi	B2_c	BCC-B2c	non-stoichiometric AlNi with vacancies
AlMo	B2_d	BCC-B2d	non-stoichiometric AlMo
CoGa	B2_e	BDD-B2e	non-stoichiometric CoGa
<b>AIP, <math>\beta</math>-BN, <math>\beta</math>-SiC-3C</b>	B3	B3 (sphalerite)	AlAs, AIP, AlSb, $\beta$ -BN, GaAs, GaP, GaSb, InAs, InP, InSb, $\beta$ -SiC-3C
<b>AlN, <math>\gamma</math>-BN, <math>\alpha</math>-SiC-2H</b>	B4	B4 (wurtzite)	AlN, $\gamma$ -BN, GaN, InN, $\alpha$ -SiC-2H
$\varepsilon$ -AlCu, L'3, B8 <sub>1</sub>	B8x	B81/B82 NiAs/InNi2	B8 <sub>1</sub> : $\varepsilon$ -AlCu, $\varepsilon$ -FeSb, NiBi, NiAs, $\alpha$ -MnBi, AuSn, $\eta$ -CuSn B8 <sub>2</sub> : $\beta$ -BiMn, InNi <sub>2</sub> , $\theta$ -SiNi <sub>2</sub> , Sn(Fe,Ti) <sub>2</sub>
$\gamma$ -CuTi	B11	B11 Prototype-CuTi	$\gamma$ -CuTi, $\alpha$ -AuTi, Ag(Ti,Zr), $\beta$ 1-NiZn
TiPt, MoPt, VPt, TiAu	B19	B19 Prototype-beta'AuCd	(Mo,V,Ti)Pt, $\beta$ -TiAu
FeSi, PtGa, PtMg	B20	B20 Prototype-FeSi	(Fe,Co,Cr,Mn)Si, PtGa, PtMg
FeB, <b>TiB, TiSi</b>	B27	B27 Prototype-FeB	CoB, FeB, HfB, TiB, ZrB, HfSi, TiSi, ZrSi, RESi (with RE = La,Ce,Pr,Nd,Pm,Sm,Eu,Gd, Tb,Dy,Ho,Er,Tm,Yb,Lu)
FeP, MnP, CoP, NiSi	B31	B31 Prototype-FeAs	(Co,Fe,Mn)P, NiSi, PtGe, NiGe, PtSi
<b>AlLi</b>	B32	B32 Prototype-NaTl	AlLi, InLi, InNa, ZnLi
CrB	B33	B33 Prototype-CrB	CrB, NiB, VB, $\beta$ -WB, NbB, CaSi, SrSi, BaSi, CaSn, LaSn, SrSn, AlY, YSi, ScSi, AlHf
CoSn	B35	B35 Prototype-CoSn	CoSn, FeSn, NiIn
$\eta$ -AgZn	Bb	Bb Prototype-AgZn	$\eta$ -AgZn (also called $\beta'$ )
$\alpha$ -WB	Bg	Bg Prototype-MoB	$\alpha$ -WB, $\alpha$ -MoB, ZrGa
$\delta$ -WC	Bh	Bh Prototype-WC	$\gamma$ -MoC, $\delta$ -WC, MoP, WP, WN
$\alpha$ -BN	Bk	Bk Prototype-BN	$\alpha$ -BN with AlN solubility
<b>TiH<sub>2</sub></b>	C1	C1 Prototype-CaF <sub>2</sub>	TiH <sub>2</sub> , $\delta$ -ZrH <sub>2</sub> , $\delta$ -HfH <sub>2</sub>
<b>Mg<sub>2</sub>Si</b>	C1a	aC1 Prototype -C1	Mg <sub>2</sub> Si, Mg <sub>2</sub> Sn, Mg <sub>2</sub> Ge, Mg <sub>2</sub> Pb, Al <sub>2</sub> Au, Si <sub>2</sub> Co, Si <sub>2</sub> Ni, Ga <sub>2</sub> Au, Ga <sub>2</sub> Pt
FeS <sub>2</sub> , NiS <sub>2</sub> , AuSb <sub>2</sub>	C2	C2 Prototype-FeS2	AuSb <sub>2</sub>
Cu <sub>2</sub> O	C3	C3 Prototype-Cu2O	(Cu,Ag) <sub>2</sub> O
<b>CuTi<sub>2</sub></b>	C11b	C11b Prototype-MoSi2	MoSi <sub>2</sub> , WSi <sub>2</sub> , AgTi <sub>2</sub> , AgZr <sub>2</sub> , AlCr <sub>2</sub> , CuTi <sub>2</sub> , CuZr <sub>2</sub> , ZnTi <sub>2</sub> , ZnZr <sub>2</sub> , TaNi <sub>2</sub> , ScAg <sub>2</sub> , ScCu <sub>2</sub> , TiAu <sub>2</sub> , YAg <sub>2</sub>
CaSi <sub>2</sub>	C12	C12 Prototype-CaSi2	CaSi <sub>2</sub> , CaGe <sub>2</sub>
<b>MgZn<sub>2</sub>, Laves-C14, Hexagonal Laves</b>	C14	C14 Prototype-MgZn <sub>2</sub> (Laves)	Al <sub>2</sub> Zr, Co <sub>2</sub> Mg, Co <sub>2</sub> Ta, Cr <sub>2</sub> Nb, Cr <sub>2</sub> Ti, Cr <sub>2</sub> Ta, Cr <sub>2</sub> Hf, Cr <sub>2</sub> Zr, Cu <sub>2</sub> Mg, Fe <sub>2</sub> Hf, Fe <sub>2</sub> Mo, Fe <sub>2</sub> Nb, Fe <sub>2</sub> Sc, Fe <sub>2</sub> Ta, Fe <sub>2</sub> Ti, Fe <sub>2</sub> W, Li <sub>2</sub> Ca, Mg <sub>2</sub> Ba, Mg <sub>2</sub> Ca, Mg <sub>2</sub> Dy, Mg <sub>2</sub> Er, Mg <sub>2</sub> Eu, Mg <sub>2</sub> Ho, Mg <sub>2</sub> Lu, Mg <sub>2</sub> Sr, Mg <sub>2</sub> Tb, Mg <sub>2</sub> Tm, Mg <sub>2</sub> Y, Mg <sub>2</sub> Yb, Mn <sub>2</sub> Er, Mn <sub>2</sub> Ho, Mn <sub>2</sub> Lu, Mn <sub>2</sub> Nb, Mn <sub>2</sub> Nd, Mn <sub>2</sub> Sm, Mn <sub>2</sub> Tm, Mn <sub>2</sub> Sc, Mn <sub>2</sub> Ta, Mn <sub>2</sub> Ti, Mn <sub>2</sub> Zr, V <sub>2</sub> Ta, Zn <sub>2</sub> Mg, Zn <sub>2</sub> Ti, Zr <sub>2</sub> Ag

<b>MgCu<sub>2</sub>, Laves-C15, Cubic Laves</b>	C15	C15 Prototype-MgCu <sub>2</sub> (Laves)	Al <sub>2</sub> Ba, Al <sub>2</sub> Ca, Al <sub>2</sub> Ce, Al <sub>2</sub> Dy, Al <sub>2</sub> Er, Al <sub>2</sub> Eu, Al <sub>2</sub> Gd, Al <sub>2</sub> Ho, Al <sub>2</sub> La, Al <sub>2</sub> Lu, Al <sub>2</sub> Nd, Al <sub>2</sub> Pr, Al <sub>2</sub> Sc, Al <sub>2</sub> Sm, Al <sub>2</sub> Tb, Al <sub>2</sub> Tm, Al <sub>2</sub> Y, Al <sub>2</sub> Yb, Co <sub>2</sub> Ce, Co <sub>2</sub> Dy, Co <sub>2</sub> Er, Co <sub>2</sub> Gd, Co <sub>2</sub> Ho, Co <sub>2</sub> Nd, Co <sub>2</sub> Pr, Co <sub>2</sub> Sc, Co <sub>2</sub> Sm, Co <sub>2</sub> Ti, Co <sub>2</sub> Y, Co <sub>2</sub> Zr, Co <sub>2</sub> Nb, Co <sub>2</sub> Ta, Co <sub>2</sub> Hf, Cr <sub>2</sub> Ti, Cr <sub>2</sub> Zr, Cr <sub>2</sub> Nb, Cr <sub>2</sub> Ta, Cr <sub>2</sub> Hf, Cu <sub>2</sub> Mg, Cu <sub>2</sub> Zr, Fe <sub>2</sub> Ce, Fe <sub>2</sub> Dy, Fe <sub>2</sub> Er, Fe <sub>2</sub> Gd, Fe <sub>2</sub> Ho, Fe <sub>2</sub> Lu, Fe <sub>2</sub> Sc, Fe <sub>2</sub> Sm, Fe <sub>2</sub> Tb, Fe <sub>2</sub> Tm, Fe <sub>2</sub> Y, Fe <sub>2</sub> Yb, Fe <sub>2</sub> Hf, Fe <sub>2</sub> Ti, Fe <sub>2</sub> Zr, Mg <sub>2</sub> Ce, Mg <sub>2</sub> Gd, Mg <sub>2</sub> La, Mg <sub>2</sub> Nd, Mg <sub>2</sub> Pr, Mg <sub>2</sub> Sm, Mn <sub>2</sub> Dy, Mn <sub>2</sub> Gd, Mn <sub>2</sub> Tb, Mn <sub>2</sub> Y, Ni <sub>2</sub> Ca, Ni <sub>2</sub> Sc, V <sub>2</sub> Zr, V <sub>2</sub> Ta, V <sub>2</sub> Hf, Zn <sub>2</sub> Zr
Laves-Ag <sub>2</sub> Na	C15a	C15a Prototype-MgCu <sub>2</sub> (Laves)	Ag <sub>2</sub> Na, Au <sub>2</sub> Bi
Laves-Ni <sub>5</sub> Zr	C15b	C15b Prototype-MgSnCu <sub>4</sub> (Laves)	Ni <sub>5</sub> Zr, Ni <sub>5</sub> Hf
<b>θ-Al<sub>2</sub>Cu</b>	C16	C16 Prototype-Al <sub>2</sub> Cu	θ-Al <sub>2</sub> Cu, (Co,Cr,Fe,Mn,Mo,Ni,Ta,W) <sub>2</sub> B, Hf <sub>2</sub> Al, Hf <sub>2</sub> Si, Pb <sub>2</sub> Au, Sn <sub>2</sub> (Co,Fe,Mn), Ta <sub>2</sub> (Co,Ni), (Ta,Zr) <sub>2</sub> Si, (Hf,Ti,Zr) <sub>2</sub> Ni, Zr <sub>2</sub> (Co,Fe,Ni,Si), Zr <sub>2</sub> Ga, CoSc <sub>2</sub> , Mg <sub>2</sub> Pt
Fe(As,Sb) <sub>2</sub>	C18	C18 Prototype-FeS <sub>2</sub>	(Co,Fe,Ni)(As,Sb) <sub>2</sub>
α-Sm	C19	C19 Prototype-Sm	α-Sm with 17 solutes
Fe <sub>2</sub> P	C22	C22 Prototype-Fe <sub>2</sub> P	Cu <sub>2</sub> P, Fe <sub>2</sub> P, Mn <sub>2</sub> P, Ni <sub>2</sub> P, FeNiP
CaH <sub>2</sub> , α-SrH <sub>2</sub>	C23	C23 Prototype-PbCl <sub>2</sub>	CaH <sub>2</sub> , α-SrH <sub>2</sub> , α-BaH <sub>2</sub>
AlRE <sub>2</sub> , Co <sub>2</sub> Si, AlPt <sub>2</sub> (ht)	C23a	C23 Prototype-Co <sub>2</sub> Si	AlPt <sub>2</sub> (ht), AlPr <sub>2</sub> , AlNd <sub>2</sub> , AlSm <sub>2</sub> , AlGd <sub>2</sub> , AlTb <sub>2</sub> , AlDy <sub>2</sub> , AlHo <sub>2</sub> , AlEr <sub>2</sub> , Ca <sub>2</sub> Sn, Sr <sub>2</sub> Sn, Ca <sub>2</sub> Si, Sr <sub>2</sub> Si, θ-Ni <sub>2</sub> Si, Ni <sub>2</sub> Ge, Ca <sub>2</sub> Cu, Co <sub>2</sub> Si, Pb(Ca,Sr) <sub>2</sub>
<b>TiB<sub>2</sub></b>	C32	C32 Prototype-AlB <sub>2</sub>	AlB <sub>2</sub> , CrB <sub>2</sub> , HfB <sub>2</sub> , MgB <sub>2</sub> , MnB <sub>2</sub> , NbB <sub>2</sub> , TiB <sub>2</sub> , VB <sub>2</sub> , ZrB <sub>2</sub>
RESi <sub>2</sub> , RE <sub>2</sub> Si <sub>3</sub> , RE <sub>3</sub> Si <sub>5</sub>	C32o	C32o Prototype-AlB <sub>2</sub>	RESi <sub>2</sub> (with RE = Y) RE <sub>2</sub> Si <sub>3</sub> (with RE = Sc, Nd) RE <sub>3</sub> Si <sub>5</sub> (with RE = Sm, Gd, Tb, Dy, Ho, Er)
<b>Laves-MgNi<sub>2</sub></b>	C36	C36 Prototype-MgNi <sub>2</sub> (Laves)	MgNi <sub>2</sub> , Cr <sub>2</sub> (Ti,Zr), Cu <sub>2</sub> Mg, ternary phases in Mg-Al-Ca-Sr-Ba, Co <sub>2</sub> Ta, Co <sub>2</sub> Ti
η-Cu <sub>2</sub> Sb	C38	C38 Prototype-Cu <sub>2</sub> Sb	η-Cu <sub>2</sub> Sb, Cu <sub>2</sub> As, Mn <sub>2</sub> Sb
CrSi <sub>2</sub>	C40	C40 Prototype-CrSi <sub>2</sub>	CrSi <sub>2</sub> , NbSi <sub>2</sub> , TaSi <sub>2</sub> , VSi <sub>2</sub> , HfSn <sub>2</sub> , WAl <sub>2</sub> , WSi <sub>2</sub>
REZn <sub>2</sub> , Al <sub>2</sub> Sr	C42	C42 Prototype-CeCu <sub>2</sub>	Al <sub>2</sub> Sr, REZn <sub>2</sub> (with all RE), BaZn <sub>2</sub> , SrZn <sub>2</sub> , RECu <sub>2</sub> (with RE = Ce, Nd, Pr, Sm), Ag <sub>2</sub> Ba
Si <sub>2</sub> Zr	C49	C49 Prototype-Si <sub>2</sub> Zr	Si <sub>2</sub> Zr, Si <sub>2</sub> Hf, Sn <sub>2</sub> RE (RE = Y, Gd, Tb, Dy, Ho, Er, Tm, Lu)
<b>TiSi<sub>2</sub></b>	C54	C54 Prototype-TiSi <sub>2</sub>	TiSi <sub>2</sub> , ZrSn <sub>2</sub>
NiMg <sub>2</sub> C <sub>a</sub>	Ca	Ca Prototype-NiMg <sub>2</sub>	NiMg <sub>2</sub> with Cu solubility
CuMg <sub>2</sub> , Mn <sub>2</sub> B	Cb	Cb Prototype-CuMg <sub>2</sub>	CuMg <sub>2</sub> , Cr <sub>2</sub> B, Mn <sub>2</sub> B, NbSn <sub>2</sub> , VSn <sub>2</sub>

$RESi_2$ , $RE_3Si_5$ $RE_6Si_{11}$ C <sub>c</sub>	Cc	Cc Prototype-ThSi2	$RESi_2$ (with RE = Y, La, Ce, Sm, Gd, Dy) $RE_3Si_5$ (with RE = Y) $RE_6Si_{11}$ (with RE = Pr, Nd, Ho)
SiPt <sub>3</sub>	D0c	D0c Prototype-SiU3	SiPt <sub>3</sub>
$M_3P$ ( $M=Co,Cr,Fe$ , $Mn,Ni,V$ ), Fe <sub>3</sub> B	D0e	D0e Prototype-Ni3P	Co <sub>3</sub> P, Cr <sub>3</sub> P, Fe <sub>3</sub> P, Mn <sub>3</sub> P, Ni <sub>3</sub> P, V <sub>3</sub> P(ht), Fe <sub>3</sub> B
Kieftite Sb <sub>3</sub> Co	D02	D02 Prototype-As3Co	Sb <sub>3</sub> Co
Mg <sub>3</sub> RE, Zn <sub>3</sub> RE, $\beta$ -Ni <sub>3</sub> Sn, $\gamma$ -Cu <sub>3</sub> Sn, Cu <sub>3</sub> Sb	D03	D03 Prototype-BiF3	Mg <sub>3</sub> RE (with RE = Ce,La,Pr,Nd,Sm,Gd,Tb, Dy,Tm,Lu) Zn <sub>3</sub> RE (with RE = Y,Ce,Nd,Sm,Pr,Gd,Tb,Dy, Er,Ho,Lu,Tm) $\gamma$ -Cu <sub>3</sub> Sn, Sn <sub>3</sub> Cu, $\beta$ -Ni <sub>3</sub> Sn, Ni <sub>3</sub> Sb, Cu <sub>3</sub> Sb, Mn <sub>3</sub> Si, Li <sub>3</sub> Bi, K <sub>3</sub> Bi
Fe <sub>3</sub> Si	D03a	D03a Prototype-BiF3	Fe <sub>3</sub> Si with Al solubility
Li <sub>2</sub> MgSn	D03b	D03b Prototype-BiF3	Li <sub>2</sub> MgSn
$\theta$ -Fe <sub>3</sub> C, Al <sub>3</sub> Ni	D011	D011 Prototype-Fe3C	Co <sub>3</sub> B, Ni <sub>3</sub> B, $\theta$ -Fe <sub>3</sub> C, Mn <sub>3</sub> C, Al <sub>3</sub> Ni, Y <sub>3</sub> Ni, La <sub>3</sub> Ni, Ni <sub>3</sub> Si, Ni <sub>3</sub> Ge
CoY <sub>3</sub>	CoY3	D011 Prototype-Fe3C	CoRE <sub>3</sub> (with RE = Y,La,Pr,Nd,Sm,Gd.Dy, Ho,Er)
Na <sub>3</sub> As, Na <sub>3</sub> P	D018	D018 Prototype-Na3As	Na <sub>3</sub> As, Na <sub>3</sub> Bi, Na <sub>3</sub> P
$\alpha_2$ -Ti <sub>3</sub> Al, Al <sub>3</sub> X, X <sub>3</sub> Al D0 <sub>19</sub>	D019	D019 Prototype-Ni3Sn	Al <sub>3</sub> RE (with RE = Y,La,Ce,Pr,Nd,Sm,Gd), RE <sub>3</sub> Al (with RE = La,Ce,Pr,Nd), Nb <sub>3</sub> Al, Ta <sub>3</sub> Al, $\alpha_2$ -Ti <sub>3</sub> Al, Ti <sub>3</sub> Ga, Zr <sub>3</sub> Al, Ti <sub>3</sub> Sn, Ni <sub>3</sub> Zr, Ni <sub>3</sub> In, Ni <sub>3</sub> Sn(lt), Co <sub>3</sub> W, Mo <sub>3</sub> Al, Mo <sub>3</sub> Pt, Mn <sub>3</sub> Sn
Cu <sub>3</sub> P	D021	D021 Prototype-Cu3P	Cu <sub>3</sub> P, Na <sub>3</sub> P, K <sub>3</sub> Bi
Al <sub>3</sub> X D0 <sub>22</sub>	D022	D022 Prototype-TiAl3	$\alpha$ -Al <sub>3</sub> Hf, Al <sub>3</sub> Cr, Al <sub>3</sub> Mo, Al <sub>3</sub> Nb, Al <sub>3</sub> Ta, $\varepsilon$ - Al <sub>3</sub> Ti, Al <sub>3</sub> V, Al <sub>3</sub> Zr, Ga <sub>3</sub> Ti, Ni <sub>3</sub> V, Pt <sub>3</sub> V
Al <sub>3</sub> Zr D0 <sub>23</sub>	D023	D023 Prototype-ZrAl3	Al <sub>3</sub> Zr, $\beta$ -Al <sub>3</sub> Hf, Ga <sub>3</sub> Zr
Ni <sub>3</sub> Ti D0 <sub>24</sub>	D024	D024 Prototype-Ni3Ti	Ni <sub>3</sub> Ti, Al <sub>3</sub> Dy, Au <sub>3</sub> Ga
Ag <sub>3</sub> Sn D0 <sub>a</sub>	D0a	D0a Prototype-Cu3Ti	Ag <sub>3</sub> Sn, Cu <sub>3</sub> Sb, $\gamma$ -Ni <sub>3</sub> Mo, Ni <sub>3</sub> Nb, Ni <sub>3</sub> Sb, Ni <sub>3</sub> Ta
Cr <sub>7</sub> C <sub>3(hT)</sub> , Mn <sub>7</sub> C <sub>3</sub> D10 <sub>1</sub>	D101	D101 Prototype-Mn7C3	Cr <sub>7</sub> C <sub>3</sub> , Mn <sub>7</sub> C <sub>3</sub> with boride solubility
Cr <sub>7</sub> C <sub>3(lT)</sub>	D102	D102 Prototype- Th7Fe3	Cr <sub>7</sub> C <sub>3</sub>
Al <sub>4</sub> (Ca,Sr,Ba)	D13	D13 Prototype-Al4Ba	Al <sub>4</sub> Ca, Al <sub>4</sub> Sr, Al <sub>4</sub> Ba, Al <sub>4</sub> Eu, Ga <sub>4</sub> Sr, Ga <sub>4</sub> Ba
Al <sub>2</sub> CaZn <sub>2</sub>	D13a	D13a Al2CaZn2	Al <sub>2</sub> CaZn <sub>2</sub> , Al <sub>2</sub> REZn <sub>2</sub> (with RE = La,Ce,Pr,Nd)
Al <sub>11</sub> RE <sub>3</sub>	D13b	D13b Prototype-Al4Ba	Al <sub>11</sub> RE <sub>3</sub> (with RE = La,Ce,Pr,Nd,Sm)
Al <sub>3</sub> CuRE	D13c	D13c Prototype- BaNiSn3	Al <sub>3</sub> CuRE (with RE = Y,La,Ce)
Sc(Ag,Cu) <sub>4</sub>	D1a	D1a Prototype-MoNi4	ScAg <sub>4</sub> , ScCu <sub>4</sub> , TiAu <sub>4</sub> , MoNi <sub>4</sub> , WNi <sub>4</sub>
B <sub>4</sub> C	D1g	D1g Prototype-B4C	B <sub>4</sub> C with Si solubility

(Ca,Sr,Ba)B <sub>6</sub>	D21	D21 Prototype-CaB6	CaB <sub>6</sub> , KB <sub>6</sub> , NaB <sub>6</sub> , NdB <sub>6</sub> , SrB <sub>6</sub> , BaB <sub>6</sub>
<b>V-phase Mg<sub>2</sub>Zn<sub>11</sub>, Mg<sub>2</sub>Cu<sub>6</sub>Al<sub>5</sub></b>	D22	D22 Prototype-Mg <sub>2</sub> Zn <sub>11</sub>	Mg <sub>2</sub> Zn <sub>11</sub> , Mg <sub>2</sub> Cu <sub>6</sub> Al <sub>5</sub>
NaZn <sub>13</sub>	D23	D23 Prototype-NaZn <sub>13</sub>	NaZn <sub>13</sub> , CaZn <sub>13</sub> , SrZn <sub>13</sub> , BaZn <sub>13</sub> , LaZn <sub>13</sub> , EuZn <sub>13</sub>
Co <sub>13</sub> La	D23a	D23 Prototype-NaZn <sub>13</sub>	Co <sub>13</sub> La
<b>Mg<sub>12</sub>RE, Zn<sub>12</sub>RE</b> Al <sub>8</sub> (Fe,Cr) <sub>4</sub> RE	D2b	D2b Prototype-Mn <sub>12</sub> Th	Mg <sub>12</sub> RE (with RE = La,Ce,Pr) Mn <sub>12</sub> RE (with RE = Y,Gd,Tb,Dy,Ho,Er,Tm) Zn <sub>12</sub> RE (with RE = Sc,Y,La,Ce,Pr,Nd,Sm, Pm,Eu,Gd,Tb,Dy,Ho,Er,Tm,Yb,Lu) Al <sub>8</sub> Cr <sub>4</sub> RE (with RE = Y,La,Pr,Nd,Sm, Gd,Ho,Er,Yb) Al <sub>8</sub> Fe <sub>4</sub> RE (with RE = Sc,Y,La,Ce,Pr,Nd,Sm, Gd,Tb,Ho,Tm,Yb,Lu)
REZn <sub>5</sub> , (Ca,Sr)Zn <sub>5</sub> , (Ca,Sr)Cu <sub>5</sub> , (Sr,Ba)Ag <sub>5</sub>	D2d	D2d Prototype-CaCu5	REZn <sub>5</sub> (with RE = La,Ce,Eu), CaZn <sub>5</sub> , SrZn <sub>5</sub> , CaCu <sub>5</sub> , CaNi <sub>5</sub> , SrCu <sub>5</sub> , SrAg <sub>5</sub> , BaAg <sub>5</sub> , ScNi <sub>5</sub> , LaNi <sub>5</sub> , RECo <sub>5</sub> (with RE = Sc,Y,Ce,Pr,Nd,Sm, Gd,Dy, Ho,Er)
<b>Al<sub>6</sub>Mn, Al<sub>6</sub>(Fe,Mn)</b>	D2h	D2h Prototype-Al <sub>6</sub> Mn	Al <sub>6</sub> Mn (Al <sub>6</sub> Fe) <sub>metastable</sub>
α-Mg <sub>3</sub> (Bi,Sb) <sub>2</sub>	D52	D52 Prototype-La <sub>2</sub> O <sub>3</sub>	Mg <sub>3</sub> (Bi,Sb) <sub>2</sub>
β-Mg <sub>3</sub> (Bi,Sb) <sub>2</sub>	D53	D53 Prototype-Mn <sub>2</sub> O <sub>3</sub>	Mg <sub>3</sub> (Bi,Sb) <sub>2</sub>
Zn <sub>3</sub> P <sub>2</sub>	D59	D59 Prototype-Zn <sub>3</sub> P <sub>2</sub>	Zn <sub>3</sub> As <sub>2</sub> , Zn <sub>3</sub> P <sub>2</sub>
Cr <sub>3</sub> C <sub>2</sub>	D510	D510 Prototype-Cr <sub>3</sub> C <sub>2</sub>	(Cr,Fe,Ni) <sub>3</sub> C <sub>2</sub> with boron solubility
Al <sub>3</sub> Ni <sub>2</sub>	D513	D513 Prototype-Al <sub>3</sub> Ni <sub>2</sub>	Al <sub>3</sub> Ni <sub>2</sub> , Al <sub>3</sub> Pt <sub>2</sub> , Ga <sub>3</sub> Pt <sub>2</sub> , In <sub>3</sub> Ni <sub>2</sub>
RE <sub>3</sub> Si <sub>2</sub>	D5a	D5a Prototype-Si <sub>2</sub> U <sub>3</sub>	RE <sub>3</sub> Si <sub>2</sub> (with RE = La,Ce), Si <sub>2</sub> Sc <sub>2</sub> Al, Si <sub>2</sub> Hf <sub>3</sub> , Si <sub>2</sub> Zr <sub>3</sub> , Ga <sub>2</sub> Zr <sub>3</sub> , (Nb,V) <sub>3</sub> B <sub>2</sub> , MoCr <sub>2</sub> B <sub>2</sub>
<b>Al<sub>4</sub>C<sub>3</sub></b>	D71	D71 Prototype-Al <sub>4</sub> C <sub>3</sub>	Al <sub>4</sub> C <sub>3</sub> with Si and V solubility
Sb <sub>3</sub> RE <sub>4</sub>	D73a	D73a Prototype-Th <sub>3</sub> P <sub>4</sub>	Sb <sub>3</sub> RE <sub>4</sub> (with RE = Ce,Nd,Pr,Sm)
<b>Ti<sub>3</sub>B<sub>4</sub></b>	D7b	D7b Prototype-Ta <sub>3</sub> B <sub>4</sub>	(Cr,Hf,Mn,Nb,Ta,Ti,V) <sub>3</sub> B <sub>4</sub> with Fe & Mo
Ag <sub>5</sub> Zn <sub>8</sub> , γ-AlCu	D82	D82 Prototype-Cu <sub>5</sub> Zn <sub>8</sub>	(Ag,Cu,Fe) <sub>5</sub> Zn <sub>8</sub> , γ-AlCu, γ-NiZn, ε-Al <sub>8</sub> (Cr,Fe,V) <sub>5</sub>
Ga <sub>6</sub> V <sub>7</sub>	D82a	D82 Prototype-Cu <sub>5</sub> Zn <sub>8</sub>	Ga <sub>6</sub> V <sub>7</sub>
Cu <sub>9</sub> Al <sub>4</sub> , 'InAg <sub>2</sub> '	D83	D83 Prototype-Cu <sub>9</sub> Al <sub>4</sub>	γ-Cu <sub>9</sub> Al <sub>4</sub> , γ-Co <sub>9</sub> Zn <sub>4</sub> , In <sub>5</sub> Ag <sub>8</sub>
Cr <sub>23</sub> C <sub>6</sub> , τ-Ni <sub>20</sub> Al <sub>3</sub> B <sub>6</sub>	D84	D84 Prototype-Cr <sub>23</sub> C <sub>6</sub>	τ-Ni <sub>20</sub> Al <sub>3</sub> B <sub>6</sub> , Cr <sub>23</sub> C <sub>6</sub> , Mn <sub>23</sub> C <sub>6</sub> with boron solubility
Fe <sub>7</sub> W <sub>6</sub>	D85	D85 Prototype-Fe <sub>7</sub> W <sub>6</sub> (Frank-Kasper)	(Co,Fe) <sub>7</sub> (Mo,Nb,Ta,W) <sub>6</sub> with Mn, Ni and Ti solubility
ε-Cu <sub>15</sub> Si <sub>4</sub>	D86	D86 Prototype-Cu <sub>15</sub> Si <sub>4</sub>	ε-Cu <sub>15</sub> Si <sub>4</sub>
Mn <sub>5</sub> Si <sub>3</sub> , <b>Ti<sub>5</sub>Si<sub>3</sub></b> , Ti <sub>5</sub> Sn <sub>3</sub>	D88	D88 Prototype-Mn <sub>5</sub> Si <sub>3</sub>	RE <sub>5</sub> Si <sub>3</sub> and RE <sub>5</sub> Sn <sub>3</sub> (with RE = Dy,Er,Gd,Ho,Sc,Sm,Tm,Y) Cr <sub>5</sub> Si <sub>3</sub> , Fe <sub>5</sub> Si <sub>3</sub> , Mn <sub>5</sub> Si <sub>3</sub> , Nb <sub>5</sub> Si <sub>3</sub> , Ta <sub>5</sub> Si <sub>3</sub> , Ti <sub>5</sub> Si <sub>3</sub> , V <sub>5</sub> Si <sub>3</sub> , W <sub>5</sub> Si <sub>3</sub> , Zr <sub>5</sub> Si <sub>3</sub> , Ti <sub>5</sub> Sn <sub>3</sub> , Zr <sub>5</sub> Sn <sub>3</sub>
Sb <sub>3</sub> RE <sub>5</sub>	D88b	D88b Prototype-Mn <sub>5</sub> Si <sub>3</sub>	Sb <sub>3</sub> RE <sub>5</sub> (with RE = Sc,Y,La,Ce,Pr,Nd,Sm,Eu, Gd,Tb,Dy,Ho,Er,Tm,Yb,Lu)

<b>Al<sub>8</sub>Mn<sub>5</sub> Al<sub>8</sub>Cr<sub>5</sub></b>	D810	D810 Prototype-Al8Cr5	Al <sub>8</sub> Mn <sub>5</sub> , Al <sub>8</sub> Cr <sub>5</sub> with Si, Cu, Fe solubility
Mn <sub>23</sub> RE <sub>6</sub> , Fe <sub>23</sub> RE <sub>6</sub> , Mg <sub>23</sub> Sr <sub>6</sub> , Li <sub>23</sub> Sr <sub>6</sub>	D8a	D8a Prototype-Mn23Th6	Mn <sub>23</sub> RE <sub>6</sub> (with RE = Y,Gd,Sc,Pr,Nd,Sm,Tb, Dy,Ho,Er,Tm,Lu) Fe <sub>23</sub> RE <sub>6</sub> (with RE = Y,Gd,Tb,Dy,Ho,Er,Tm) Mg <sub>23</sub> Sr <sub>6</sub> , Li <sub>23</sub> Sr <sub>6</sub> , Co <sub>23</sub> (Hf,Zr) <sub>6</sub>
σ-FeCr, σ-AlNb, σ-AlTa, σ-MnV, σ-NiV, σ-FeMo	D8b	D8b Prototype-FeCr	σ-FeCr, intermetallic phases in Al-Mo-Nb-Ta and Mn-Ni-V-W
Al <sub>9</sub> Co <sub>2</sub>	D8d	D8d Prototype-Al9Co2	Al <sub>9</sub> Co <sub>2</sub> with Cr, Fe, Ni solubility
<b>T-phase, Tau Mg<sub>32</sub>(Al,Zn)<sub>49</sub></b>	D8e	Tau Prototype-Mg <sub>32</sub> (Al,Zn)49	Ternary phases in Mg-Al-Cu-Zn-(Ag)
Mg <sub>5</sub> (Ga,In) <sub>2</sub>	D8g	D8g Prototype-Ga <sub>2</sub> Mg <sub>5</sub>	Mg <sub>5</sub> (Ga,In) <sub>2</sub>
B <sub>2</sub> W or W <sub>2</sub> B <sub>5</sub>	D8h	D8h Prototype-B2W	W <sub>2</sub> (B,Va) <sub>5</sub>
(Mo,V) <sub>2</sub> B <sub>5</sub>	D8i	D8i Prototype-Mo2B5	(Mo,V) <sub>2</sub> B <sub>5</sub>
RE <sub>5</sub> Si <sub>3</sub> D8l Cr <sub>5</sub> B <sub>3</sub>	D8l	D8l Prototype-Cr5B3	RE <sub>5</sub> Si <sub>3</sub> (with RE = La,Ce,Pr,Nd) Ca <sub>5</sub> Si <sub>3</sub> , Sr <sub>5</sub> Si <sub>3</sub> , Ca <sub>5</sub> Sn <sub>3</sub> , Sr <sub>5</sub> Sn <sub>3</sub> , Ca <sub>5</sub> Zn <sub>3</sub> , Sr <sub>5</sub> Pb <sub>3</sub> , Ca <sub>5</sub> Ag <sub>3</sub> , Cr <sub>5</sub> B <sub>3</sub> (with Fe and Mo substitution)
Cr <sub>5</sub> Si <sub>3</sub> , La <sub>5</sub> Sn <sub>3</sub>	D8m	D8m Prototype-W5Si3	(Cr,Mo,Nb,V,W) <sub>5</sub> Si <sub>3</sub> RE <sub>5</sub> Sn <sub>3</sub> (with RE = La,Ce,Pr)
FeZr <sub>3</sub> , CoZr <sub>3</sub>	E1a	E1a Prototype-FeZr3	(Co,Fe)Zr <sub>3</sub>
(Fe,Mn,Ti) <sub>3</sub> AlC	E21	E21 Prototype-CaTiO3	(Fe,Mn,Ti) <sub>3</sub> AlC
<b>Al<sub>7</sub>Cu<sub>2</sub>Fe, Al<sub>7</sub>CuMn<sub>2</sub></b>	E9a	E9a Prototype-Al7Cu2Fe	Al <sub>7</sub> Cu <sub>2</sub> Fe, Al <sub>7</sub> CuMn <sub>2</sub> , τ <sub>9</sub> -Al <sub>7</sub> Cu <sub>2</sub> Zr
(Co,Fe,Mo,Ni) <sub>3</sub> W <sub>3</sub> C	E93	E93 Prototype-Fe3W3C	(Co,Fe,Mo,Ni) <sub>3</sub> W <sub>3</sub> C
<b>Al<sub>4</sub>SiC<sub>4</sub>, Al<sub>5</sub>C<sub>3</sub>N</b>	E94	E94 Prototype-Al5C3N	Al <sub>4</sub> SiC <sub>4</sub> , Al <sub>5</sub> C <sub>3</sub> N
<b>γ-AlTi, MgIn, L1<sub>0</sub></b>	L10	L10 Prototype-AuCu	AlTi, MgIn, GaTi, CoPt, α-MnNi, NiPt
<b>CuPt L1<sub>1</sub></b>	L11	L11 Prototype-CuPt	CuPt
<b>Al<sub>3</sub>M, Al<sub>3</sub>RE L1<sub>2</sub></b>	L12	L12 Prototype-AuCu3	Al <sub>3</sub> RE with (RE = Sc,Er,Tm,Yb,Lu) metastable Al <sub>3</sub> M (with M = Cr,Cu,Li,Mg,Ti, V,Zr), Pb <sub>3</sub> RE (RE = Dy,Nd,Pr,Tb), Sn <sub>3</sub> RE (RE = Ce,Dy,Er,Gd,Ho,Nd,Pr,Sm,Y,Yb), Pt <sub>3</sub> Ti
AlRE <sub>3</sub> L1 <sub>2</sub>	L12b	L12b Prototype-AuCu3	AlRE <sub>3</sub> with (RE = Sc,Ce,Pr), SnRE <sub>3</sub> (RE = Ce,Pr)
Ni <sub>3</sub> (Al,Fe,Si) L1 <sub>2</sub>	L12c	L12-FCC!FCC-A1	Ag <sub>3</sub> Mg, Ni <sub>3</sub> Al, Ni <sub>3</sub> Fe, Ni <sub>3</sub> Si, Si <sub>3</sub> Ni, Ni <sub>3</sub> Ge
MgIn <sub>3</sub> , InMg <sub>3</sub> L1 <sub>2</sub>	L12d	L12d Prototype-AuCu3	MgIn <sub>3</sub> , InMg <sub>3</sub> , InAg <sub>3</sub>
Pt <sub>3</sub> Ga, Pt <sub>3</sub> Ge, Ni <sub>3</sub> Ga	L12e	L12e Prototype-AuCu3	Pt <sub>3</sub> Ga, Pt <sub>3</sub> Ge, Ni <sub>3</sub> Ga
AlCu <sub>2</sub> Mn, AlNi <sub>2</sub> (Hf,Ti)	L21	Heusler Prototype-AlCu2Mn	AlCu <sub>2</sub> Mn, AlNi <sub>2</sub> (Hf,Ti)
<b>TiH<sub>2</sub>(low-T), ZrH<sub>2</sub></b>	L'2	L'2 Prototype-TiH2	TiH <sub>2</sub> (low-T), ZrH <sub>2</sub> , SiPt <sub>2</sub>
AgMg <sub>3</sub>	AMg3	cF264 Prototype-AgMg3	AgMg <sub>3</sub> with Al, In, Sn solubility
Al <sub>11</sub> Mn <sub>4</sub> (low-T)	aP15	aP15 Prototype-Al11Mn4	Al <sub>11</sub> Mn <sub>4</sub> (low-T), Al <sub>11</sub> Cr <sub>4</sub> with Fe solub.
ξ-Al <sub>2</sub> Fe	aP18	aP18 Prototype-Al2Fe	ξ-Al <sub>2</sub> Fe
La <sub>2</sub> Sn <sub>3</sub>	aP20	aP20 Prototype-Nd2Sn3	RE <sub>2</sub> Sn <sub>3</sub> (with RE = La,Pr,Sm)

$(Al,In)_3Ca_8$	aP22	aP22 Prototype-Ca8In3	$(Al,In)_3Ca_8$
$Cu_7In_3$	aP40	aP40 Prototype-Cu7In3	$Cu_7In_3$
$\tau_3-REMgZn_2$	cF16	Tao3 Prototype-MnCu2Al	$REMgZn_2$ (with $RE = La,Ce,Nd,Sm,Pr,Gd,Ho,Tb,Er,Tm,LuY,Dy$ )
$Ti_2Co, Ti_2Ni, Hf_2Co$	cF96	cF96 Prototype-Ti2Ni	$Ti_2Co, Ti_2Ni, Hf_2Co, Sc_2Ni, Hf_2Fe$
$Al_{69}Ta_{39}$	C432	cF432 Prototype-Ta39Al69	$Al_{69}Ta_{39}$
$Al_{12}Mn, Al_{12}W$	cl26	cl26 Prototype-Al12W	$Al_{12}(Mn,Mo,W)$ ( $Al_{12}Fe$ -metastable)
$(Co,Fe,Mn)_4N$	cP5	cP5 Prototype-Fe4N	$(Co,Fe,Mn)_4N$ with Cr, Ni, C substitution
$Al_7Sr_8$	cP60	cP60 Prototype-Ba8Ga7	$Al_7Sr_8, Ga_7(Sr,Ba)_8$
K(Ge,Si)	cP64	cP64 Prototype-KGe	KGe, KSi
CaAlSi, SrAlSi	hP3_	hP3 Prototype-AlB2	CaAlSi, SrAlSi
$\delta$ -CuZn	hP3	CuZn_delta	$\delta$ -CuZn with Ag solubility
$(Ca,Sr,Y)Al_2Si_2$	hP5	hP5 Prototype-Ce2SO2	$(Ca,Sr,Y)Al_2Si_2$
Fe <sub>2</sub> Si	hP6	hP6 Prototype-Ni2Al	Ni <sub>2</sub> Al, Fe <sub>2</sub> Si
BaCu, SrCu	hP8	hP8 Prototype-BaCu	(Sr,Ba)Cu with Ca solubility
Cu <sub>6</sub> Y	hP8_	hP8 Prototype-TbCu7	"Cu <sub>6</sub> Y"
$M_2AlC$	hP8c	MAX Prototype-Cr2AlC	$M_2AlC$ ( $M = Cr, Ti, Nb, V$ )
LiB	hP8*	hP8 Prototype-LiB	LiB
$\delta$ -WC, $\eta$ -MoC	hP12	eta-MoC	$\delta$ -WC, $\eta$ -MoC with Ta and V solub.
Ti <sub>5</sub> Ga <sub>4</sub>	hP18	hP18 Prototype-Ti5Ga4	Ti <sub>5</sub> Ga <sub>4</sub> , Zr <sub>5</sub> Al <sub>4</sub> , Zr <sub>5</sub> Ga <sub>4</sub>
Co <sub>3</sub> RE <sub>4</sub>	hP22	hP22 Prototype-Co <sub>3</sub> Ho <sub>4</sub>	Co <sub>3</sub> RE <sub>4</sub> (with RE = Y, Gd, Ho)
Ti <sub>6</sub> Sn <sub>5</sub> , V <sub>6</sub> Ga <sub>5</sub>	H22P	hP22 Prototype-Ti6Sn5	Ti <sub>6</sub> Sn <sub>5</sub> , V <sub>6</sub> Ga <sub>5</sub>
$\beta$ -AlMnSi	hP26	AlMnSi_beta	$\beta$ -AlMnSi
Zn <sub>5</sub> RE	hP36	hP36 Prototype-ErZn5	Zn <sub>5</sub> RE (with RE = Er, Ho, Lu, Tm, Y)
$Mg_{17}Sr_2,$ $(Mg,Fe,Zn,Co)_{17}RE_2$	hP38	hP38 Prototype-Ni17Th2	Mg <sub>17</sub> Sr <sub>2</sub> , Mg <sub>17</sub> RE <sub>2</sub> (with RE = La, Ce, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Sc, Y) Zn <sub>17</sub> RE <sub>2</sub> (with RE = La, Ce, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Sc, Y) Fe <sub>17</sub> RE <sub>2</sub> (with RE = La, Ce, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Sc, Y) Co <sub>17</sub> RE <sub>2</sub> (with RE = Er – metastable : Ce, Pr, Nd, Sm, Gd, Dy, Ho)
Co <sub>11</sub> Ce <sub>24</sub>	hP70	Co11RE24 Prototype-Co11Ce24	Co <sub>11</sub> Ce <sub>24</sub> with Nd solubility
Mg <sub>38</sub> Sr <sub>9</sub>	hP94	hP94 Prototype-Mg38Sr9	Mg <sub>38</sub> Sr <sub>9</sub> with Al and Ca solubility
SbSn	hR3	hR3 Prototype-SbSn"	SbSn with Pb and Bi solubility
Fe <sub>3</sub> RE, Ni <sub>3</sub> Y, Co <sub>3</sub> RE	hR12	hR12 Prototype-Ni3Pu	Fe <sub>3</sub> RE (with RE = Y, Sm, Tb, Dy, Er, Tm, Lu), Co <sub>3</sub> RE (with RE = Y, Ce, Pr, Nd, Sm, Gd, Dy, Ho, Er), Ni <sub>3</sub> Y
$(Al,Ga,In)_2Li_3$	h15R	hR15 Prototype-Al2Li3	$(Al,Ga,In)_2Li_3$
V <sub>4</sub> C <sub>3</sub> (V <sub>4</sub> C <sub>2.65</sub> , 'V <sub>3</sub> C <sub>2</sub> ')	hR21	hR21 Prototype-Sn4P3	V <sub>4</sub> C <sub>3</sub>
Fe <sub>3</sub> Gd, Fe <sub>3</sub> Ho	hR36	hR36 Prototype-Be3Nb	Fe <sub>3</sub> RE (with RE = Gd, Ho)

Al <sub>5</sub> Mo	h36R	h36R Prototype-MoAl <sub>5</sub>	Al <sub>5</sub> Mo with Fe solubility
Mg <sub>3</sub> In	hR48	hR48 Prototype-Mg <sub>3</sub> In	Mg <sub>3</sub> In
RE <sub>2</sub> Co <sub>7</sub>	hR54	hR54 Prototype-Gd <sub>2</sub> Co <sub>7</sub>	RE <sub>2</sub> Co <sub>7</sub> (with RE = Y,La,Ce,Pr,Nd,Sm,Gd,Dy,Ho,Er)
Zn <sub>17</sub> RE <sub>2</sub> , Co <sub>17</sub> RE <sub>2</sub>	hR57	hR57 Prototype-Th <sub>2</sub> Zn <sub>17</sub>	Zn <sub>17</sub> RE <sub>2</sub> , Fe <sub>17</sub> RE <sub>2</sub> (with RE = Gd,Nd,Pr,Sm), Co <sub>17</sub> RE <sub>2</sub> (with RE = Y,Ce,Pr,Nd,Sm,Gd,Dy,Ho)
Al <sub>3</sub> Ho, Al <sub>3</sub> Dy	hR60	hR60 Prototype-HoAl <sub>3</sub>	Al <sub>3</sub> Ho, Al <sub>3</sub> Dy
Co <sub>19</sub> RE <sub>5</sub>	hR72	hR72 Prototype-Ce <sub>5</sub> Co <sub>19</sub>	Co <sub>19</sub> RE <sub>5</sub> (with RE = La,Ce,Pr,Nd,Sm)
Ni <sub>7</sub> Zr <sub>2</sub> , Co <sub>7</sub> Nb <sub>2</sub>	mC18	mC18 Prototype-Ni <sub>7</sub> Zr <sub>2</sub>	Ni <sub>7</sub> Zr <sub>2</sub> , Co <sub>7</sub> Nb <sub>2</sub>
Mo <sub>3</sub> Al <sub>8</sub>	mC22	mC22 Prototype-Mo <sub>3</sub> Al <sub>8</sub>	Mo <sub>3</sub> Al <sub>8</sub> with Ga, Si, Ti solubility
ζ-Zn <sub>13</sub> Fe, γ <sub>2</sub> -Zn <sub>13</sub> Co	mC28	mC28 Prototype-Zn <sub>13</sub> Co	ζ-Zn <sub>13</sub> Fe, γ <sub>2</sub> -Zn <sub>13</sub> Co
Ni <sub>5</sub> Sb <sub>2</sub>	m28C	mS28 Prototype-Ni <sub>5</sub> Sb <sub>2</sub>	Ni <sub>5</sub> Sb <sub>2</sub>
Al <sub>4</sub> Mo, Al <sub>4</sub> W	mC30	mC30 Prototype-Al <sub>4</sub> W	Al <sub>4</sub> Mo, Al <sub>4</sub> W with Fe solubility
β-AlFeSi (τ <sub>6</sub> )	mC52	AlFeSi_beta	β-AlFeSi (τ <sub>6</sub> )
(Ag,Cu)P <sub>2</sub>	mP12	mP12 Prototype-CuP <sub>2</sub>	(Ag,Cu)P <sub>2</sub>
ZnAs <sub>2</sub>	mP24	mP24 Prototype-ZnAs <sub>2</sub>	ZnAs <sub>2</sub> , β-ZnP <sub>2</sub>
RE <sub>12</sub> Co <sub>7</sub>	mP38	mP38 Prototype-Ho <sub>12</sub> Co <sub>7</sub>	RE <sub>12</sub> Co <sub>7</sub> (with RE = Dy,Ho,Er)
Co <sub>5</sub> Y <sub>8</sub>	mP52	mP52 Prototype-Co <sub>5</sub> Y <sub>8</sub>	Co <sub>5</sub> Y <sub>8</sub>
(Al,Ga,In) <sub>2</sub> Li <sub>3</sub>	mP86	mP86 Prototype-Ta <sub>22</sub> Al <sub>21</sub>	Ta <sub>22</sub> Al <sub>21</sub>
'Al <sub>3</sub> Fe'	m102	mC102 Prototype-Al <sub>13</sub> Fe <sub>4</sub>	Al <sub>13</sub> Fe <sub>4</sub> , Al <sub>13</sub> Co <sub>4</sub>
Mg <sub>2</sub> Zn <sub>3</sub>	m110	Mg <sub>2</sub> Zn <sub>3</sub>	Mg <sub>2</sub> Zn <sub>3</sub> with Al and Cu solubility
Ni <sub>3</sub> Sn <sub>4</sub>	mC14	mS14 Prototype-Ni <sub>3</sub> Sn <sub>4</sub>	Ni <sub>3</sub> Sn <sub>4</sub> with Cu solubility
Pt <sub>3</sub> (Ge,Si)	mS16	mS16 Prototype-Pt <sub>3</sub> Ge	Pt <sub>3</sub> Ge, Pt <sub>3</sub> Si(lt) with Ga solubility
η <sub>2</sub> -AlCu, θ-InCu	mS20	eta2 Prototype-AlCu	η <sub>2</sub> -AlCu, θ-InCu, InPt
GeAs, SiAs	mS24	mS24 Prototype-SiAs	(Ge,Si)(As,P)
Hägg χ-Fe <sub>5</sub> C <sub>2</sub>	mS28	Haegg Prototype-Mn <sub>5</sub> C <sub>2</sub>	χ-Fe <sub>5</sub> C <sub>2</sub> , Mn <sub>5</sub> C <sub>2</sub> with B, Mn, V substitution
Co <sub>2</sub> Sm <sub>5</sub>	m28S	Co2RE5 Prototype-Mn5C2	Co <sub>2</sub> RE <sub>5</sub> (RE = Sm,Pr,Nd)
MoAl <sub>3</sub>	mS32	mS32 Prototype-MoAl <sub>3</sub>	MoAl <sub>3</sub> with Fe solubility
ζ'-Ni <sub>13</sub> In <sub>9</sub>	mS44	mS44 Prototype-Ni <sub>13</sub> Ga <sub>9</sub>	ζ'-Ni <sub>13</sub> In <sub>9</sub> , Ni <sub>13</sub> Ga <sub>9</sub> , Pt <sub>13</sub> In <sub>9</sub>
Al <sub>14</sub> Ca <sub>13</sub>	mS54	mS54 Prototype-Al <sub>14</sub> Ca <sub>13</sub>	Al <sub>14</sub> Ca <sub>13</sub>
AlLa, AlCe	oC16	oC16 Prototype-AlCe	AlLa, AlCe
V <sub>2</sub> B <sub>3</sub> , Nb <sub>2</sub> B <sub>3</sub>	oC20	oC20 Prototype-V <sub>2</sub> B <sub>3</sub>	V <sub>2</sub> B <sub>3</sub> , Nb <sub>2</sub> B <sub>3</sub> with Cr and Fe solubility
REFe <sub>2</sub> Al <sub>10</sub>	oC52	oC52 Prototype-YbFe <sub>2</sub> Al <sub>10</sub>	REFe <sub>2</sub> Al <sub>10</sub> (with RE = Y,La,Ce,Pr,Nd,Sm,Gd,Tb,Dy,Ho,Er,Tm,Yb,Lu)
Al <sub>3</sub> Zr <sub>2</sub>	oF40	oF40 Prototype-Al <sub>3</sub> Zr <sub>2</sub>	(Al,Ga) <sub>3</sub> (Hf,Zr) <sub>2</sub>
(Cr,Mo,V)Ni <sub>2</sub>	oI6	oI6 Prototype-MoPt <sub>2</sub>	(Cr,Mo,V)Ni <sub>2</sub> , MoPt <sub>2</sub> , VPt <sub>2</sub>
RESi <sub>2</sub> , RE <sub>3</sub> Si <sub>5</sub> , RE <sub>5</sub> Si <sub>9</sub>	oI12	oI12 Prototype-GdSi <sub>2</sub>	RESi <sub>2</sub> (with RE = Sm,Gd,Tb,Dy) RE <sub>3</sub> Si <sub>5</sub> (with RE = La,Ce) RE <sub>5</sub> Si <sub>9</sub> (with RE = Pr,Nd,Ho)

$\alpha\text{-Al}_2\text{Sr}$	o12I	oI12 Prototype-KHg2	$\alpha\text{-Al}_2\text{Sr}, \text{Ga}_2\text{Sc}, \text{Cu}_2\text{Pr}, \text{Cu}_2\text{Sm}$
$\alpha\text{-Ti}_6\text{Sn}_5$	oI44	oI44 Prototype-Ti6Sn5	$(\text{Nb},\text{Ti})_6\text{Sn}_5$
$\text{Al}_{11}RE_3, \text{Zn}_{11}RE_3$	oI28	oI28 Prototype-Al11La3	$\text{Al}_{11}RE_3$ (with $RE = \text{La,Ce,Pr,Nd,Sm}$ ) $\text{Zn}_{11}RE_3$ (with $RE = \text{Y,Ce,Pr,Nd,Sm,Dy,Gd,Tb,Yb}$ )
$\alpha\text{-PtTi}$	oP4	oP4 Prototype-AuCd	$\alpha\text{-PtTi}$
$\text{Ag}_3\text{Sb}$	oP4a	oP4 Prototype-Ag3Sb	$\text{Ag}_3\text{Sb}$ with Au solubility
$\text{CaMgSi}$	oP12	oP12 Prototype-TiNiSi	$\text{CaMgSi}$
$\text{AlRE}$	oP16	oP16 Prototype-AlEr	$\text{AlRE}$ (with $RE = \text{Pr,Nd,Sm,Gd,Tb,Dy,Ho,Er,Tm,Lu}$ )
$\text{Cu}_4\text{Ti}$	oP20	oP20 Prototype-ZrAu4	$\beta'\text{-Cu}_4\text{Ti}$
$(\text{Co,Mn,Ni})_3\text{Sn}_2$	o20P	oP20 Prototype-Ni3Sn2	$(\text{Co,Mn,Ni})_3\text{Sn}_2$ with Au solubility
$\text{Co}_2\text{Y}_3$	C2Y3	Co2RE3 Prototype-Co2Y3	$\text{Co}_2\text{Y}_3$ with REE solubility
$\text{SiAs}_2, \text{GeAs}_2, \text{SiP}_2$	oP24	oP24 Prototype-GeAs2	$(\text{Ge},\text{Si})(\text{As},\text{P})_2$
$\text{BaSi}_2, \text{SrGe}_2$	o24P	oP24 Prototype-BaSi2	$\text{BaSi}_2, \text{SrGe}_2$
$(\text{Ce,Sm})\text{Cu}_6$	oP28	oP28 Prototype-CeCu6	$\text{CeCu}_6, \text{SmCu}_6$
$RE_5\text{Si}_4, RE_5\text{Sn}_4$	oP36	oP36 Prototype-Sm5Ge4	$RE_5\text{Si}_4$ (with $RE = \text{Y,Tb,Dy,Ho,Er,Tm,Lu}$ ) $RE_5\text{Sn}_4$
$\text{Ca}_7\text{Sn}_6, \text{Ca}_7\text{Ge}_6$	oP52	oP52 Prototype-Ca7Sn6	$\text{Ca}_7\text{Sn}_6, \text{Ca}_7\text{Ge}_6$ with Si solubility
$\delta\text{-NiMo}$	oP56	NiMo-delta	$\delta\text{-NiMo}$ with solubility of Cr, Fe, W
$\text{CoY, NiZr}$	oS8	oS8 Prototype-TII	$\text{CoY, GaCa, GaSc, GeSr, NiLa, NiZr, PbSr}$
$\text{ZrGa}_2, (\text{Nd,Pr})\text{Sn}_2$	oS12	oS12 Prototype-ZrGa2	$\text{ZrGa}_2, (\text{Pr},\text{Nd},\text{Sm})\text{Sn}_2$
$\alpha\text{-GdSn}_3$	oS16	oS16 Prototype-GdSn3	$\alpha\text{-}(\text{Y,Gd,Ho,Er,Tm})\text{Sn}_3$
$\text{Pt}_5\text{Ga}_3, \text{Ni}_3\text{Ga}_3$	o16S	oS16 Prototype-Pt5Ga3	$\text{Pt}_5\text{Ga}_3, \text{Ni}_3\text{Ga}_3$
$\text{LaZn}_4$	oS20	oS20 Prototype-LaZn4	$\text{LaZn}_4$ with Mg solubility
$\text{Co}_3\text{La}_2$	o20S	Co3RE2 Prototype-La2Ni3	$\text{Co}_3\text{La}_2$ with REE solubility
$(\text{Ce,Nd,Pr,Sm})\text{Sb}_2$	oS24	oS24 Prototype-SmSb2	$(\text{Ce,Nd,Pr,Sm})\text{Sb}_2$
$\alpha\text{-Tb}_3\text{Sn}_7, \alpha\text{-Dy}_3\text{Sn}_7, \text{Pr}_3\text{Sn}_7, \text{Sm}_3\text{Sn}_7,$	oS28	oS28 Prototype-Tb3Sn7	$(\text{Pr,Sm,Tb,Dy})_3\text{Sn}_7$
$RE_3\text{Sn}_5$	oS32	oS32 Prototype-Pu3Pd5	$RE_3\text{Sn}_5$ (with $RE = \text{La,Ce,Pr,Nd}$ )
$\text{Li}_7\text{Sn}_2$	oS36	oS36 Prototype-Li7Ge2	$\text{Li}_7\text{Sn}_2$
$\beta\text{-FeSi}_2$	oS48	Luobusaite Prototype-FeSi2	$\beta\text{-FeSi}_2$ with Al solubility
' $\text{Ni}_{10}\text{Zr}_7$ '	oS68	oS68 Prototype-Zr7Ni10	' $\text{Ni}_{10}\text{Zr}_7$ '
$(\text{Ce,Nd,Pr,Sm})_2\text{Sb}$	tI12	tI12 Prototype-La2Sb	$(\text{Ce,Nd,Pr,Sm})_2\text{Sb}$
$\text{Ni}_8(\text{Nb,Ta})$	tI18	tI18 Prototype-Pt8Ti	$\text{Ni}_8(\text{Nb,Ta})$
$\eta\text{-Al}_2\text{Ti}$	tI24	tI24 Prototype-HfGa2	$\eta\text{-Al}_2\text{Ti, Ga}_2\text{Ti, Pb}_2\text{Pr, Pb}_2\text{Nd, Al}_2\text{Mg-metastable}$
$\varepsilon\text{-Al}_3\text{Ti}(lt)$	tI32	tI32 Prototype-TiAl3	$\varepsilon\text{-Al}_3\text{Ti}(lt)$ with Si solubility
$\text{Zn}_{11}RE, \text{CaZn}_{11}$	tI48	tI48 Prototype-BaCd11	$\text{Zn}_{11}RE$ (with $RE = \text{La,Ce,Pr,Nd,Yb}$ ), $\text{CaZn}_{11}$
$RE_{11}\text{Sn}_{10}$	tI84	tI84 Prototype-Ho11Ge10	$RE_{11}\text{Sn}_{10}$ (with $RE = \text{Y,La,Ce,Pr,Nd,Sm,Gd,Tb,Dy,Ho,Er,Tm,Yb,Lu}$ )
$\text{Mg}_{41}RE_5$	tI92	tI92 Prototype-Mg41Ce5	$\text{Mg}_{41}RE_5$ (with $RE = \text{La,Ce,Pr,Nd,Sm}$ )
$\text{Zn}_3(\text{As,P})_2$	i160	tI160 Prototype Cd3As2	$\text{Zn}_3(\text{As,P})_2$

$\alpha$ -FeSi <sub>2</sub> (high-T)	tP3	Ferdisilicite Prototype-FeSi <sub>2</sub>	$\alpha$ -FeSi <sub>2</sub> (high-T) with Al solubility
Pt <sub>3</sub> Al	tP16	tP16 Prototype-Pt <sub>3</sub> Ga	Pt <sub>3</sub> Ga(ht), Pt <sub>3</sub> Al
Al <sub>2</sub> RE <sub>3</sub> , Al <sub>2</sub> Zr <sub>3</sub>	tP20	tP20 Prototype-Al <sub>2</sub> Zr <sub>3</sub>	Al <sub>2</sub> RE <sub>3</sub> (with RE = Y,Gd,Tb,Dy,Ho, Er,Tm,Lu), Al <sub>2</sub> (Hf,Zr) <sub>3</sub> , Li <sub>2</sub> Sr <sub>3</sub> , Zn <sub>2</sub> Zr <sub>3</sub>
$\zeta$ -Al <sub>5</sub> Ti <sub>2</sub>	tP28	tP28 Prototype Al <sub>5</sub> Ti <sub>2</sub>	$\zeta$ -Al <sub>5</sub> Ti <sub>2</sub> with Nb solubility
Ti <sub>3</sub> P, Ti <sub>3</sub> Si	tP32	tP32 Prototype-Ti <sub>3</sub> P	M <sub>3</sub> P (with M = Hf,Ta,Ti,V,Zr) X <sub>3</sub> Si (with X = Nb,Ta,Ti,Zr)
RE <sub>5</sub> Si <sub>4</sub> , (Hf,Ti,Zr) <sub>5</sub> Si <sub>4</sub>	tP36	tP36 Prototype-Zr <sub>5</sub> Si <sub>4</sub>	RE <sub>5</sub> Si <sub>4</sub> (with RE = Sc,La,Ce,Pr,Nd,Pm,Sm, Gd), (Hf,Ti,Zr) <sub>5</sub> Si <sub>4</sub>
$\beta$ -Al <sub>3</sub> Mg <sub>2</sub>	Beta	Beta Prototype-Mg <sub>28</sub> Al <sub>45</sub>	$\beta$ -Al <sub>3</sub> Mg <sub>2</sub> with Li, Zn solubility
$\eta$ -Al <sub>5</sub> Fe <sub>2</sub>	Eta	Eta Prototype-Al <sub>5</sub> Fe <sub>2</sub>	$\eta$ -Al <sub>5</sub> Fe <sub>2</sub> with Mn, Zn solubility
$\phi$ -Mg <sub>6</sub> (Al,Zn) <sub>5</sub>	Phi	Phi AlMgZn	$\phi$ -Mg <sub>6</sub> (Al,Zn,Cu) <sub>5</sub>
$\zeta$ -AlCu or $\zeta$ -Al <sub>9</sub> Cu <sub>11</sub>	Zeta	Zeta Prototype-AlCu	$\zeta$ -Al <sub>9</sub> Cu <sub>11</sub>
<b>E-phase Al<sub>18</sub>Cr<sub>2</sub>Mg<sub>3</sub> Al<sub>20</sub>(Cr,Ti,V)<sub>2</sub>RE, Zn<sub>22</sub>Zr, CaCr<sub>2</sub>Al<sub>20</sub></b>	E	E Prototype-CeCr <sub>2</sub> Al <sub>20</sub>	Al <sub>18</sub> Cr <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> Mn <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> Mo <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> Ta <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> Ti <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> V <sub>2</sub> Mg <sub>3</sub> , Al <sub>18</sub> W <sub>2</sub> Mg <sub>3</sub> Al <sub>20</sub> (Cr,Ti,V) <sub>2</sub> RE, CaCr <sub>2</sub> Al <sub>20</sub> , Zn <sub>22</sub> Zr
<b>Q-Al<sub>3</sub>Cu<sub>2</sub>Mg<sub>9</sub>Si<sub>7</sub></b>	Q	Q Prototype-Th <sub>7</sub> S <sub>12</sub>	Al <sub>3</sub> Cu <sub>2</sub> Mg <sub>9</sub> Si <sub>7</sub>
P phase	P	P phase	CrMoNi with Fe solubility
P1 phase	P1	P1 phase M <sub>3</sub> Ca <sub>2</sub>	(Mg,Cu) <sub>3</sub> Ca <sub>2</sub>
P2 phase	P2	P2 phase MCu <sub>3</sub>	(Mg,Ca)Cu <sub>3</sub>
R phase	R	R phase	(Co,Fe,Ni) <sub>27</sub> W <sub>14</sub> (Co,Cr,Fe,Ni,W) <sub>12</sub>
<b>S-Al<sub>2</sub>CuMg</b>	S	S Prototype-Al <sub>2</sub> CuMg	S-Al <sub>2</sub> CuMg (ordered E1a)
<b>X-REZnMg<sub>12</sub></b>	X	X LPSO-18R	REZnMg <sub>12</sub> (with RE = Y,Gd,Dy,Tb,Ho,Er, Tm, Lu) Long-Period-Stacking-Order
<b><math>\mu</math>-Al<sub>4</sub>Mn, Al<sub>4</sub>Cr</b>	Al4M	Al <sub>4</sub> Mn	Al <sub>4</sub> Mn, Al <sub>4</sub> Cr (metastable Al <sub>4</sub> Fe)
'Al <sub>3</sub> Mn(ht)'	A3M_	oP156 Prototype-Al <sub>11</sub> Mn <sub>4</sub>	Al <sub>29</sub> Mn <sub>10</sub> (high-T)
<b>Al<sub>11</sub>Cr<sub>2</sub></b>	AlCr	Al <sub>11</sub> Cr <sub>2</sub>	Al <sub>11</sub> Cr <sub>2</sub> with Mn solubility
<b><math>\alpha</math>-AlFeSi, <math>\tau_5</math></b>	Tau5	AlFeSi_alpha	stoichiometry ~Al <sub>7</sub> Fe <sub>2</sub> Si (chinese script/rods)
<b><math>\gamma</math>-AlFeSi, <math>\tau_2</math></b>	Tau2	AlFeSi_gamma	stoichiometry $\tau_2$ ~Al <sub>3</sub> FeSi
<b><math>\alpha</math>-AlMnSi</b>	AlM1	AlMnSi_alpha	stoichiometry $\tau_9$ ~Al <sub>9</sub> Mn <sub>2</sub> Si
<b><math>\alpha</math>-AlFeMnSi</b>	AFMS	AlFeMnSi_alpha	stoichiometry ~Al <sub>18</sub> (Fe,Mn) <sub>4</sub> Si <sub>3</sub>
'MgZn' or Mg <sub>12</sub> Zn <sub>13</sub>	MgZn	Mg <sub>12</sub> Zn <sub>13</sub>	Mg <sub>12</sub> Zn <sub>13</sub> with Cu and Al solubility
$\delta$ -AlCu or $\delta$ -Al <sub>2</sub> Cu <sub>3</sub>	Dlta	AlCu_delta	$\delta$ -Al <sub>2</sub> Cu <sub>3</sub> with Zn solubility
$\eta_1$ -AlCu	EtaH	Eta1 Prototype-AlCu	$\eta_1$ -AlCu with Zn solubility
$\gamma$ -AlCu(high-T)	Gam5	gamma Prototype-AlCu	$\gamma$ -Al <sub>4</sub> Cu <sub>9</sub> (high-T) with Zn solubility
$\tau$ -AlCuZn	Tau_	AlCuZn_Tau	$\tau$ -Al <sub>9</sub> Cu <sub>9</sub> Zn <sub>2</sub> (approx.)
B <sub>3</sub> Si	B3Si	B3Si	~B <sub>3</sub> Si
B <sub>6</sub> Si	B6Si	B6Si	~B <sub>6</sub> Si
B <sub>10</sub> Si	B_nS	B_nSi	~B <sub>10</sub> Si

$\beta$ -Zn <sub>3</sub> Li <sub>2</sub>	Li2Z	Zn3Li2	$\beta$ -Li <sub>2</sub> Zn <sub>3</sub>
I-phase REZn <sub>6</sub> Mg <sub>3</sub>	Fm53	I REZn <sub>6</sub> Mg <sub>3</sub>	REZn <sub>6</sub> Mg <sub>3</sub> (with RE = Y,Gd,Tb,Dy,Ho,Er,Tm, Lu)
Mg <sub>3</sub> Ce <sub>2</sub> Zn <sub>3</sub>	MgCZ	Mg3Ce2Zn3	$\sim$ Mg <sub>3</sub> Ce <sub>2</sub> Zn <sub>3</sub>
$\sigma$ -CrMn	HSig	High_Sigma	$\sim$ Cr <sub>2</sub> Mn <sub>7</sub>
pre- $\beta''$ AlMgSi	prB2	pre-B''	complex AlMgSi FCC monoclinic
$\beta''$ AlMgSi	B''	B''	complex AlMgSi non-FCC monoclinic
U2 Mg <sub>4</sub> Si <sub>x</sub> Al <sub>y</sub>	Pnma	U2	complex AlMgSi orthorhombic
$\beta'$ Mg <sub>9</sub> Si <sub>7</sub> Al <sub>3</sub>	P_6	B'	complex AlMgSi hexagonal
Ca(Mg,Zn) <sub>5</sub>	IM1	IM1-CaMgZn	CaMg <sub>4</sub> Zn, CaMgZn <sub>4</sub>
Mg <sub>5</sub> Gd, Mg <sub>5</sub> Sm	Mg5X	cF448 Prototype-Mg5Gd	Mg <sub>5</sub> Gd, Mg <sub>5</sub> Sm
(Li,Mg) <sub>17</sub> Sn <sub>4</sub>	L17X	cF420 Prototype-Li17Pb4	Li <sub>17</sub> Sn <sub>4</sub> with Mg solubility
$\gamma$ -AgMg <sub>4</sub>	hP_G	AgMg4 gamma_hp	$\gamma$ -AgMg <sub>4</sub>
Ag <sub>17</sub> Mg <sub>54</sub>	A17M	oP142 Prototype-Hf54Os17	Ag <sub>17</sub> Mg <sub>54</sub>
Li <sub>2</sub> Ag	L2A	Prototype-Li2Ag	$\sim$ Li <sub>2</sub> Ag
Li <sub>3</sub> Ag	L3A	Prototype-Li3Ag	$\sim$ Li <sub>3</sub> Ag
Li <sub>6</sub> Ag	L6A	Prototype-Li6Ag	$\sim$ Li <sub>6</sub> Ag
REZn <sub>3</sub>	Pmna	oP16 Prototype-YZn3	REZn <sub>3</sub> (with RE = Ce,Dy,Er,Gd,Ho,Lu,Nd,Pr, Sm,Tb,Tm,Pm,Y)
Zn <sub>58</sub> RE <sub>13</sub>	hex5	hP142 Prototype-Cd58Gd13	Zn <sub>58</sub> RE <sub>13</sub> (with RE = Y,Sc,Ce,Pr,Nd,Sm,Gd,Tb,Dy,Ho,Er,Tm,Yb,Lu,Pm)
Zn <sub>22</sub> RE <sub>3</sub>	Z22X	tI100 Prototype-Pu3Zn22	Zn <sub>22</sub> RE <sub>3</sub> (with RE = La,Ce,Pr,Nd,Sm,Pm,Gd)
Al <sub>3</sub> Tb, $\beta$ -Al <sub>3</sub> Y	A3RE	hR12 Prototype-BaPb3	Al <sub>3</sub> Tb, $\beta$ -Al <sub>3</sub> Y
$\tau_1$ -Ce <sub>3</sub> Zn <sub>9</sub> Mg <sub>2</sub>	Tao1	Tau Prototype-Al11La3	$\tau_1$ -Ce <sub>3</sub> Zn <sub>9</sub> Mg <sub>2</sub>
$\tau_2$ -(Ce,Nd) <sub>2</sub> Zn <sub>45</sub> Mg <sub>53</sub>	Tao2	Tao2	$\tau_2$ -(Ce,Nd) <sub>2</sub> Zn <sub>45</sub> Mg <sub>53</sub>
$\tau_4$ -RE <sub>2</sub> Mg <sub>5</sub> Zn <sub>9</sub>	Tao4	Tau4 RE2Zn9Mg5	RE <sub>2</sub> Mg <sub>5</sub> Zn <sub>9</sub> (with RE = Ce,Nd,Pr,Sm,Dy)
$\tau_5$ -RE <sub>3</sub> Zn <sub>30</sub> Mg <sub>13</sub>	Tao5	Tau5 RE3Zn30Mg13	RE <sub>3</sub> Zn <sub>30</sub> Mg <sub>13</sub> (with RE = Y,Ce,Nd,Pr,Sm,Dy)
$\tau_7$ -Ce <sub>20</sub> Zn <sub>81</sub> Mg <sub>19</sub>	Tao7	Tao7	$\tau_7$ -(Ce,Nd) <sub>20</sub> Zn <sub>81</sub> Mg <sub>19</sub>
<b>Al<sub>9</sub>(Co,Fe,Ni)<sub>2</sub>(Sr,Ba)</b>	A9ME	hP12 Prototype-Al9Co2Sr	Al <sub>9</sub> (Co,Fe,Ni) <sub>2</sub> (Sr,Ba)
(Y,Yb)Zn <sub>2</sub>	RZ2L	ReZn2(LT)	(Y,Yb)Zn <sub>2</sub> (low-T)
(Co,Fe) <sub>6</sub> W <sub>6</sub> C	M12C	cF104 Prototype-Fe6W6C	(Co,Fe) <sub>6</sub> W <sub>6</sub> C with Mo solubility
Mn <sub>11</sub> Si <sub>19</sub>	NCL1	tP20 Prototype-Mn11Si19	Mn <sub>11</sub> (Si,Al) <sub>19</sub> Nowotny Chimney Ladder Phase
Al <sub>63</sub> Mo <sub>37</sub>	AM8	AM8 Al63Mo37	Al <sub>63</sub> Mo <sub>37</sub> with Ti solubility
"As <sub>3</sub> P <sub>2</sub> " or "AsP"	AsP	AsP-orthorhombic	$\sim$ "As <sub>3</sub> P <sub>2</sub> "
$\gamma_1$ -FeZn	Gam1	FeZn-Gamma-1	Fe <sub>7</sub> (Fe,Zn) <sub>6</sub> Zn <sub>38</sub>
$\delta_1$ -FeZn	FeZ2	FeZn-delta-1	Fe(Al,Fe,Zn) <sub>3</sub> Zn <sub>13</sub>
Co <sub>11</sub> (Hf,Zr) <sub>2</sub>	gam8	gamma-ortho	Co <sub>11</sub> (Hf,Zr) <sub>2</sub>
CoRE <sub>3</sub>	CoY3	CoY3 Prototype-Fe3C	CoRE <sub>3</sub> (RE = Y,La,Pr,Nd,Sm,Gd,Tb,Dy,Ho,Er,Tm)
M <sub>4</sub> AlC <sub>3</sub>	MAX2	MAX2 Prototype-CsYb3Se4	(Cr,Nb,Ti,V)AlC <sub>3</sub>

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**Table 3: Pure compounds and their phases in the FTlite 8.3 Database**

Al21Pt8	FTLITE	S
Al22Mo5	FTLITE	S
Al23CuFe4	FTLITE	S
Al23V4	FTLITE	S
Al2Au	FTLITE	S
Al2Ba	FTLITE	S1 S2
Al2Ca	FTLITE	S1 S2
Al2CaZn2	FTLITE	S
Al2Cu	FTLITE	S1 S2 S3
Al2Cu3	FTLITE	S
Al2CuLi	FTLITE	S
Al2CuMg	FTLITE	S
Al2Fe	FTLITE	S
Al2Hf	FTLITE	S
Al2Hf3	FTLITE	S
Al2Li	FTLITE	S1 S2
Al2Li18Si6	FTLITE	S
Al2Li3	FTLITE	S
Al2Mn2Si3	FTLITE	S
Al2O3	FTLITE	S1 S2 S3 S4
Al2Pt	FTLITE	S
Al2Sc	FTLITE	S
Al2Sr	FTLITE	S1 S2
Al2Ti	FTLITE	S1 S2
Al2W	FTLITE	S
Al2Y3	FTLITE	S
Al2Zr	FTLITE	S
Al2Zr3	FTLITE	S
Al30Mg23	FTLITE	S
Al39Cu33Zr6	FTLITE	S
Al3BC	FTLITE	S
Al3Ca8	FTLITE	S
Al3Co	FTLITE	S
Al3Cr	FTLITE	S1 S2
Al3Cu	FTLITE	S
Al3Cu2	FTLITE	S
Al3CuCe	FTLITE	S
Al3CuLa	FTLITE	S
Al3Hf	FTLITE	S1 S2
Al3Hf2	FTLITE	S
Al3Hf4	FTLITE	S
Al3Li	FTLITE	S1 S2 S3
Al3Li15Si6	FTLITE	S
Al3Li7Si4	FTLITE	S
Al3Mg	FTLITE	S
Al3MnSi2	FTLITE	S
Al3Mo	FTLITE	S
Al3Nb	FTLITE	S
Al3Ni	FTLITE	S
Al3Ni5	FTLITE	S
Al3Pt2	FTLITE	S
Al3Pt5	FTLITE	S
Al3Sc	FTLITE	S
Al3Ta	FTLITE	S
Al3Ti	FTLITE	S1 S2 S3 S4
Al3V	FTLITE	S1 S2 S3
Al3Y	FTLITE	S1 S2
Al3Zr	FTLITE	S1 S2 S3
Al3Zr2	FTLITE	S
Al3Zr5	FTLITE	S
Al4Ba	FTLITE	S
Al4C3	FTLITE	S
Al4C4Si	FTLITE	S
Al4Ca	FTLITE	S
Al4Ce	FTLITE	S

Al4Cr	FTLITE	S
Al4Fe	FTLITE	S
Al4Li9	FTLITE	S
Al4MgY	FTLITE	S
Al4Mn	FTLITE	S
Al4Mo	FTLITE	S
Al4Sr	FTLITE	S
Al4W	FTLITE	S
Al4Zr5	FTLITE	S
Al53Mg14Li33	FTLITE	S
Al57Cu11Li32	FTLITE	S
Al5Ba4	FTLITE	S
Al5C3N	FTLITE	S
Al5Co2	FTLITE	S
Al5Cu2Mg8Si6	FTLITE	S
Al5Cu6Mg2	FTLITE	S
'Al5CuLi3'	FTLITE	S
Al5Li	FTLITE	S
Al5Mo	FTLITE	S
Al5W	FTLITE	S
Al63Mo37	FTLITE	S
Al6Fe	FTLITE	S
Al6Mn	FTLITE	S
Al6Ni3Si	FTLITE	S
Al77W23	FTLITE	S
Al7Cr	FTLITE	S
Al7Cu2Fe	FTLITE	S
Al7Cu2Zr	FTLITE	S
Al7Cu3Mg6	FTLITE	S
Al7CuFe2	FTLITE	S
Al7CuMn2	FTLITE	S
Al7CuZr2	FTLITE	S
Al7Sr8	FTLITE	S
Al7W3	FTLITE	S
Al8C7B4	FTLITE	S
Al8C7Si	FTLITE	S
Al8Cr5	FTLITE	S1 S2
Al8CrTi3	FTLITE	S
Al8CrY	FTLITE	S
Al8EuFe2	FTLITE	S
Al8FeMg3Si6	FTLITE	S
Al8Mn4Y	FTLITE	S
Al8Mo3	FTLITE	S
Al8V5	FTLITE	S
Al99Mn23	FTLITE	S
Al9Co2	FTLITE	S
Al9Co2Ba	FTLITE	S
Al9Co2Sr	FTLITE	S
Al9Cr3Si	FTLITE	S
Al9Cr4	FTLITE	S1 S2
Al9Cu11	FTLITE	S
Al9Fe2Ba	FTLITE	S
Al9Fe2Sr	FTLITE	S
Al9Ni2Ba	FTLITE	S
Al9Ni2Sr	FTLITE	S
AlAs	FTLITE	S
AlAu	FTLITE	S
AlAu2	FTLITE	S
AlAu4	FTLITE	S
AlB12	FTLITE	S
AlB2	FTLITE	S
AlCa5Zn2	FTLITE	S
AlCeSi2	FTLITE	S
AlCo	FTLITE	S
AlCr2	FTLITE	S

(Al) 4.95 (Cu) 1.053 (Li) 2.997

AlCr2B2	FTLITE	S
AlCr3B4	FTLITE	S
AlCu	FTLITE	S1 S2
AlCu2	FTLITE	S
AlCu3Mn2	FTLITE	S
AlH3	FTLITE	S
AlHf	FTLITE	S
AlHf2	FTLITE	S
AlLi	FTLITE	S1 S2
AlLiSi	FTLITE	S
AlMg	FTLITE	S
AlMgAg	FTLITE	S
AlMgB14	FTLITE	S
AlMo3	FTLITE	S
AlN	FTLITE	S1 S2 S3
AlNaSi	FTLITE	S
AlP	FTLITE	S
AlPt	FTLITE	S
AlPt2	FTLITE	S1 S2
AlPt3	FTLITE	S
AlSb	FTLITE	S
AlSc	FTLITE	S
AlSc2	FTLITE	S
AlSc2Si2	FTLITE	S
AlSn2Zr5	FTLITE	S
AlTi	FTLITE	S
AlY	FTLITE	S
AlY2	FTLITE	S
AlZr	FTLITE	S1 S2
AlZr2	FTLITE	S
AlZr3	FTLITE	S1 S2
As	FTLITE	S1 S2 S3 S4 S5 L
'As2Cu5'	FTLITE	S
AsCu8	FTLITE	S
AsIn	FTLITE	S
Au	FTLITE	S1 S2 S3 L
Au2Bi	FTLITE	S
Au2Pb	FTLITE	S
Au2Ti	FTLITE	S
Au3In	FTLITE	S
Au3Sn12Co5	FTLITE	S
'Au4Zn5'	FTLITE	S
Au5Zn3	FTLITE	S
Au7Ga2	FTLITE	S
Au7Ga3	FTLITE	S
Au7In3	FTLITE	S
Au8Al3	FTLITE	S
'Au8Ga2'	FTLITE	S
AuGa	FTLITE	S
AuGa2	FTLITE	S
AuIn	FTLITE	S
AuIn2	FTLITE	S
AuSb2	FTLITE	S
AuSn	FTLITE	S
AuSn2	FTLITE	S
AuSn4	FTLITE	S
AuTi	FTLITE	S1 S2
AuTi3	FTLITE	S
AuZn3	FTLITE	S
B	FTLITE	S1 S2 S3 S4 S5 S6 S7 L
B2Mo2Ni	FTLITE	S
B4Nd	FTLITE	S
B5Nd2	FTLITE	S
B66Nd	FTLITE	S
BN	FTLITE	S1 S2

Ba	FTLITE	S1	S2	S3	L
Ba10Ga	FTLITE	S			
Ba2Si	FTLITE	S			
Ba2Zn	FTLITE	S			
Ba3Si4	FTLITE	S			
Ba5Ga6	FTLITE	S			
Ba5Si3	FTLITE	S			
Ba8Ga7	FTLITE	S			
BaAl2Si2	FTLITE	S			
BaB6	FTLITE	S			
BaBe13	FTLITE	S			
BaC2	FTLITE	S1	S2		
BaCu	FTLITE	S			
BaCu13	FTLITE	S			
BaGa2	FTLITE	S			
BaGa4	FTLITE	S			
BaGe2	FTLITE	S			
BaH2	FTLITE	S1	S2		
BaSi	FTLITE	S			
BaSi2	FTLITE	S			
BaZn	FTLITE	S			
BaZn13	FTLITE	S			
BaZn2	FTLITE	S			
BaZn5	FTLITE	S			
Be	FTLITE	S1	S2	S3	L
Bi	FTLITE	S1	S2	S3	S4 L
Bi2K	FTLITE	S			
Bi2K3	FTLITE	S			
Bi3In5	FTLITE	S			
Bi3Ni	FTLITE	S			
Bi4K5	FTLITE	S			
BiIn	FTLITE	S			
BiIn2	FTLITE	S			
BiLi	FTLITE	S			
BiNa	FTLITE	S			
C	FTLITE	S1	S2	L	
Ca	FTLITE	S1	S2	S3	L
Ca11Ga7	FTLITE	S			
Ca14Si19	FTLITE	S			
Ca25Ga59	FTLITE	S			
Ca28Ga11	FTLITE	S			
Ca2Cu	FTLITE	S			
Ca2Ge	FTLITE	S			
Ca2Mg55Zn43	FTLITE	S			
Ca2Mg5Zn13	FTLITE	S			
Ca2Ni7	FTLITE	S			
Ca2Pb	FTLITE	S			
Ca2Si	FTLITE	S			
Ca2Sn	FTLITE	S			
Ca3Ga5	FTLITE	S			
Ca3Ga8	FTLITE	S			
Ca3Mg3Zn14	FTLITE	S			
Ca3P2	FTLITE	S			
Ca3Si4	FTLITE	S			
Ca3Zn	FTLITE	S			
Ca4Al3Mg	FTLITE	S			
Ca5Ga3	FTLITE	S			
Ca5Ge3	FTLITE	S			
Ca5Pb3	FTLITE	S			
Ca5Si3	FTLITE	S			
Ca5Zn3	FTLITE	S			
Ca7Ge6	FTLITE	S			
Ca7Mg6Si14	FTLITE	S			
Ca7Sn6	FTLITE	S			
CaAl2Si2	FTLITE	S			

CaAlH5	FTLITE	S
CaB6	FTLITE	S
CaBe13	FTLITE	S
CaC2	FTLITE	S1 S2 S3 S4 L
CaCr2Al20	FTLITE	S
CaCu	FTLITE	S
CaCu5	FTLITE	S
CaGa	FTLITE	S
CaGa2	FTLITE	S1 S2
CaGa4	FTLITE	S1 S2
CaGe2	FTLITE	S
CaH2	FTLITE	S
CaLi2	FTLITE	S
CaMgSi	FTLITE	S
CaNi2	FTLITE	S
CaNi3	FTLITE	S
CaNi5	FTLITE	S
CaO	FTLITE	S
CaPb	FTLITE	S
CaPb3	FTLITE	S
CaSi	FTLITE	S
CaSi2	FTLITE	S
CaTi2Al20	FTLITE	S
CaZn	FTLITE	S
CaZn11	FTLITE	S
CaZn13	FTLITE	S
CaZn2	FTLITE	S
CaZn3	FTLITE	S
CaZn5	FTLITE	S
Ce	FTLITE	S1 S2 S3 S4 L
Ce11Sn10	FTLITE	S
Ce12Y21Mg67	FTLITE	S
Ce13Zn58	FTLITE	S
Ce20Zn81Mg19	FTLITE	S
Ce2Al	FTLITE	S
Ce2Fe17	FTLITE	S
Ce2In	FTLITE	S
Ce2Mg17	FTLITE	S
Ce2Mn3Al	FTLITE	S
Ce2Mn7Al10	FTLITE	S
Ce2Sb	FTLITE	S
Ce2Sn5	FTLITE	S
Ce2Zn17	FTLITE	S1 S2
Ce2Zn45Mg53	FTLITE	S
Ce2Zn9Mg5	FTLITE	S
Ce3Al	FTLITE	S1 S2
Ce3Al11	FTLITE	S1 S2
Ce3Ga2	FTLITE	S
Ce3In	FTLITE	S
Ce3In5	FTLITE	S
Ce3Si2	FTLITE	S
Ce3Si5	FTLITE	S
Ce3Sn	FTLITE	S
Ce3Sn5	FTLITE	S
Ce3Sn7	FTLITE	S
Ce3Zn11	FTLITE	S
Ce3Zn22	FTLITE	S
Ce3Zn30Mg13	FTLITE	S
Ce3Zn9Mg2	FTLITE	S
Ce4Sb3	FTLITE	S
Ce5In4	FTLITE	S
Ce5Mg41	FTLITE	S
Ce5Si3	FTLITE	S1 S2
Ce5Si4	FTLITE	S1 S2
Ce5Sn3	FTLITE	S1 S2

Ce5Sn4	FTLITE	S
Ce6Zn83Mg11	FTLITE	S
Ce9In11	FTLITE	S
CeAl	FTLITE	S
CeAl13Mg6	FTLITE	S
CeAl2	FTLITE	S
CeAl3	FTLITE	S1 S2
CeCr2Al20	FTLITE	S
CeCr4Al5	FTLITE	S
CeCu	FTLITE	S
CeCu2	FTLITE	S
CeCu4	FTLITE	S
CeCu4Al8	FTLITE	S
CeCu5	FTLITE	S
CeCu6	FTLITE	S
CeFe2	FTLITE	S
CeFe2Al10	FTLITE	S
CeIn2	FTLITE	S
CeIn3	FTLITE	S
CeMg	FTLITE	S
CeMg12	FTLITE	S
CeMg2	FTLITE	S
CeMg3	FTLITE	S
CeMn4Al8	FTLITE	S
CeSb	FTLITE	S
CeSb2	FTLITE	S
CeSi	FTLITE	S
CeSi2	FTLITE	S
CeSn3	FTLITE	S
CeTi2Al20	FTLITE	S
CeV2Al20	FTLITE	S
CeZn	FTLITE	S
CeZn11	FTLITE	S
CeZn12	FTLITE	S
CeZn2	FTLITE	S
CeZn2Al2	FTLITE	S
CeZn2Mg	FTLITE	S
CeZn3	FTLITE	S
CeZn5	FTLITE	S
Co	FTLITE	S1 S2 S3 S4 S5 L
Co11Ce24	FTLITE	S
Co11Hf2	FTLITE	S
Co11Zr2	FTLITE	S
Co13La	FTLITE	S
Co16Nb9	FTLITE	S
Co17Ce2	FTLITE	S
Co17Dy2	FTLITE	S
Co17Er2	FTLITE	S
Co17Gd2	FTLITE	S
Co17Ho2	FTLITE	S
Co17La20	FTLITE	S
Co17Lu2	FTLITE	S
Co17Nd2	FTLITE	S
Co17Pr2	FTLITE	S
Co17Pr20	FTLITE	S
Co17Sm2	FTLITE	S
Co17Tb2	FTLITE	S
Co17Tm2	FTLITE	S
Co17Y2	FTLITE	S
Co17Yb2	FTLITE	S
Co19Ce5	FTLITE	S
Co19La5	FTLITE	S
Co19Nd5	FTLITE	S
Co19Pr5	FTLITE	S
Co19Sm5	FTLITE	S

Co23C6	FTLITE	S
Co23Hf6	FTLITE	S
Co23Zr6	FTLITE	S
Co2B	FTLITE	S
Co2Ce	FTLITE	S
Co2Dy	FTLITE	S
Co2Er	FTLITE	S
Co2Gd	FTLITE	S
Co2Hf	FTLITE	S
Co2Ho	FTLITE	S
Co2Lu	FTLITE	S
Co2N	FTLITE	S
Co2Nb	FTLITE	S
Co2Nd	FTLITE	S
Co2Nd5	FTLITE	S
Co2P	FTLITE	S
Co2Pr	FTLITE	S
Co2Pr5	FTLITE	S
Co2Sc	FTLITE	S
Co2Si	FTLITE	S
Co2Sm	FTLITE	S
Co2Sm5	FTLITE	S
Co2Ta	FTLITE	S1    S2    S3
Co2Tb	FTLITE	S
Co2Ti	FTLITE	S1    S2    S3
Co2Tm	FTLITE	S
Co2Y	FTLITE	S
Co2Yb	FTLITE	S
Co2Zr	FTLITE	S
Co3B	FTLITE	S
Co3C2	FTLITE	S
Co3Ce	FTLITE	S
Co3Dy	FTLITE	S
Co3Dy4	FTLITE	S
Co3Er	FTLITE	S
Co3Er4	FTLITE	S
Co3Gd	FTLITE	S
Co3Gd4	FTLITE	S
Co3Ho	FTLITE	S
Co3Ho4	FTLITE	S
Co3La2	FTLITE	S
Co3Lu	FTLITE	S
Co3Lu4	FTLITE	S
Co3Mo	FTLITE	S
Co3Nb	FTLITE	S
Co3Nd	FTLITE	S
Co3Nd2	FTLITE	S
Co3Nd4	FTLITE	S
Co3Pr	FTLITE	S
Co3Pr4	FTLITE	S
Co3Si	FTLITE	S
Co3Sm	FTLITE	S
Co3Sn2	FTLITE	S1    S2
Co3Tb	FTLITE	S
Co3Tb4	FTLITE	S
Co3Tm	FTLITE	S
Co3Tm4	FTLITE	S
Co3V	FTLITE	S
Co3W	FTLITE	S
Co3Y	FTLITE	S
Co3Y2	FTLITE	S
Co3Y4	FTLITE	S
Co3Yb	FTLITE	S
Co4Zn9	FTLITE	S
Co5Ce	FTLITE	S

Co5Dy	FTLITE	S
Co5Er	FTLITE	S
Co5Gd	FTLITE	S
Co5Ho	FTLITE	S
Co5La	FTLITE	S
Co5Nd	FTLITE	S
Co5Pr	FTLITE	S
Co5Sm	FTLITE	S
Co5Tb	FTLITE	S
Co5Tm	FTLITE	S
Co5Y	FTLITE	S
Co5Y8	FTLITE	S
Co6W6C	FTLITE	S
Co7C3	FTLITE	S
Co7Ce2	FTLITE	S
Co7Dy12	FTLITE	S
Co7Dy2	FTLITE	S
Co7Er12	FTLITE	S
Co7Er2	FTLITE	S
Co7Gd2	FTLITE	S
Co7Ho12	FTLITE	S
Co7Ho2	FTLITE	S
Co7La2	FTLITE	S
Co7Nb2	FTLITE	S
Co7Nd2	FTLITE	S
Co7Pr2	FTLITE	S
Co7Sm2	FTLITE	S
Co7Ta2	FTLITE	S
Co7Tb12	FTLITE	S
Co7Tb2	FTLITE	S
Co7Tm2	FTLITE	S
Co7Y2	FTLITE	S
Co7Y6	FTLITE	S
CoB	FTLITE	S
CoDy3	FTLITE	S
CoEr3	FTLITE	S
CoGa3	FTLITE	S
CoGd3	FTLITE	S
CoHf	FTLITE	S
CoHf2	FTLITE	S
CoHo3	FTLITE	S
CoIn2	FTLITE	S
CoIn3	FTLITE	S
CoLa3	FTLITE	S
CoLu3	FTLITE	S
CoMo2B2	FTLITE	S
CoN	FTLITE	S
CoN3	FTLITE	S
CoNd3	FTLITE	S
CoO	FTLITE	S
CoP	FTLITE	S
CoPr3	FTLITE	S
CoSb	FTLITE	S
CoSb2	FTLITE	S
CoSb3	FTLITE	S
CoSc	FTLITE	S
CoSc2	FTLITE	S
CoSc3	FTLITE	S
CoSi	FTLITE	S
CoSi2	FTLITE	S
CoSm3	FTLITE	S
CoSn	FTLITE	S
CoSn2	FTLITE	S
CoTb3	FTLITE	S
CoTi2	FTLITE	S

CoV3	FTLITE	S
CoY	FTLITE	S
CoY3	FTLITE	S
CoZn7	FTLITE	S
CoZr2	FTLITE	S
CoZr3	FTLITE	S
Cr	FTLITE	S1 S2 S3 S4 S5 S6 L
Cr23B6	FTLITE	S
Cr23C6	FTLITE	S
Cr2AlC	FTLITE	S
Cr2B	FTLITE	S1 S2
Cr2C	FTLITE	S
Cr2Hf	FTLITE	S1 S2
Cr2N	FTLITE	S
Cr2O3	FTLITE	S
Cr2P	FTLITE	S
Cr2Ta	FTLITE	S1 S2
Cr2Ti	FTLITE	S1 S2
Cr2Zr	FTLITE	S1 S2 S3
Cr3B2	FTLITE	S
Cr3B4	FTLITE	S
Cr3C	FTLITE	S
Cr3C2	FTLITE	S
Cr3Ga	FTLITE	S1 S2
Cr3Mn5	FTLITE	S
Cr3P	FTLITE	S
Cr3Si	FTLITE	S
Cr5B3	FTLITE	S
Cr5Si3	FTLITE	S1 S2 S3
Cr7C3	FTLITE	S
CrB	FTLITE	S
CrB2	FTLITE	S
CrB4	FTLITE	S
CrGa4	FTLITE	S
CrN	FTLITE	S
CrSi	FTLITE	S
CrSi2	FTLITE	S
CrZn13	FTLITE	S
CrZn17	FTLITE	S
Cu	FTLITE	S1 S2 S3 S4 L
'Cu10Sn3'	FTLITE	S
Cu10Zr7	FTLITE	S
Cu15Si4	FTLITE	S
Cu16Mg6Si7	FTLITE	S
Cu19Si6	FTLITE	S
Cu2Eu	FTLITE	S
Cu2Ho	FTLITE	S
Cu2La	FTLITE	S
Cu2Mg	FTLITE	S1 S2 S3
Cu2Nd	FTLITE	S
Cu2O	FTLITE	S
Cu2P	FTLITE	S
Cu2Pr	FTLITE	S
Cu2Sb	FTLITE	S
Cu2Sc	FTLITE	S
Cu2Sm	FTLITE	S
Cu2Ti	FTLITE	S
Cu2TiZr	FTLITE	S
Cu2Y	FTLITE	S1 S2
Cu2Zr	FTLITE	S
Cu33Si7	FTLITE	S
Cu37La3	FTLITE	S
'Cu3As'	FTLITE	S
Cu3Ge	FTLITE	S
Cu3Mg2Si	FTLITE	S

(Cu) 9.997 (Sn) 3.003

(Cu) 2.94 (As) 1.06

Cu3P	FTLITE	S1	S2			
Cu3Sb	FTLITE	S1	S2			
Cu3Sn	FTLITE	S1	S2			
Cu3Ti2	FTLITE	S				
Cu4In	FTLITE	S				
Cu4La	FTLITE	S				
Cu4Nd	FTLITE	S				
Cu4Pr	FTLITE	S				
Cu4Sb	FTLITE	S				
Cu4Sc	FTLITE	S				
Cu4Sm	FTLITE	S				
'Cu4Sn'	FTLITE	S				(Cu) 3.94 (Sn) 1.06
Cu4Ti3	FTLITE	S				
Cu4Y	FTLITE	S				
Cu51Zr14	FTLITE	S				
Cu5Eu	FTLITE	S				
Cu5Ho	FTLITE	S1	S2			
Cu5La	FTLITE	S				
Cu5Nd	FTLITE	S				
Cu5Pr	FTLITE	S				
Cu5Sm	FTLITE	S				
Cu5Sr	FTLITE	S				
Cu5Zr8	FTLITE	S				
Cu6La	FTLITE	S1	S2			
Cu6Nd	FTLITE	S				
Cu6Pr	FTLITE	S				
'Cu6Sb'	FTLITE	S				(Cu) 5.95 (Sb) 1.05
Cu6Sm	FTLITE	S				
'Cu6Sn5'	FTLITE	S				(Cu) 5.995 (Sn) 5.005
Cu7Ga2	FTLITE	S				
Cu7Ho2	FTLITE	S				
Cu7In3	FTLITE	S				
'Cu7In4'	FTLITE	S				(Cu) 7.04 (In) 3.96
'Cu7Sb2'	FTLITE	S				(Cu) 6.93 (Sb) 2.07
Cu7Y2	FTLITE	S				
Cu8Zr3	FTLITE	S				
Cu9Ho2	FTLITE	S				
Cu9Si2	FTLITE	S				
Cu9Zr2	FTLITE	S				
CuBe2	FTLITE	S				
CuEu	FTLITE	S				
CuEu2	FTLITE	S				
CuH	FTLITE	S				
CuHo	FTLITE	S				
CuLa	FTLITE	S				
CuNd	FTLITE	S				
CuO	FTLITE	S				
CuP2	FTLITE	S				
CuPr	FTLITE	S				
CuSc	FTLITE	S				
CuSm	FTLITE	S1	S2			
CuSr	FTLITE	S				
CuTi2	FTLITE	S				
CuY	FTLITE	S				
CuZr	FTLITE	S				
CuZr2	FTLITE	S				
Dy	FTLITE	S1	S2	S3	S4	L
Dy11Sn10	FTLITE	S				
Dy13Zn58	FTLITE	S				
Dy2Al	FTLITE	S				
Dy2Al10Mg3	FTLITE	S				
Dy2Fe17	FTLITE	S				
Dy2Zn17	FTLITE	S1	S2			
Dy3Al2	FTLITE	S				
Dy3Si4	FTLITE	S				

Dy3Si5	FTLITE	S1	S2			
Dy3Sn7	FTLITE	S1	S2			
Dy3Zn11	FTLITE	S				
Dy3Zn22	FTLITE	S				
Dy5Mg24	FTLITE	S				
Dy5Si3	FTLITE	S				
Dy5Si4	FTLITE	S				
Dy5Sn3	FTLITE	S1	S2			
Dy5Sn4	FTLITE	S				
Dy6Fe23	FTLITE	S				
Dy6Mn23	FTLITE	S				
DyAl	FTLITE	S				
DyAl2	FTLITE	S				
DyAl3	FTLITE	S1	S2			
DyCr2Al20	FTLITE	S				
DyCu4Al8	FTLITE	S				
DyFe2	FTLITE	S				
DyFe2Al10	FTLITE	S				
DyFe3	FTLITE	S				
DyMg	FTLITE	S				
DyMg2	FTLITE	S				
DyMg3	FTLITE	S				
DyMn12	FTLITE	S				
DyMn2	FTLITE	S				
DySi	FTLITE	S				
DySi2	FTLITE	S1	S2			
DySi2Al2	FTLITE	S				
DySn2	FTLITE	S				
DySn3	FTLITE	S1	S2			
DyTi2Al20	FTLITE	S				
DyV2Al20	FTLITE	S				
DyZn	FTLITE	S				
DyZn11	FTLITE	S				
DyZn12	FTLITE	S				
DyZn2	FTLITE	S				
DyZn2Mg	FTLITE	S				
DyZn3	FTLITE	S				
DyZn5	FTLITE	S				
DyZn6Mg3	FTLITE	S				
DyZnMg12	FTLITE	S				
Er	FTLITE	S1	S2	S3	S4	L
Er11Sn10	FTLITE	S				
Er13Zn58	FTLITE	S				
Er2Al	FTLITE	S				
Er2Fe17	FTLITE	S				
Er2Si2Al3	FTLITE	S				
Er2Sn5	FTLITE	S				
Er2Zn17	FTLITE	S1	S2			
Er3Al2	FTLITE	S				
Er3Al20Mg7	FTLITE	S				
Er3Si5	FTLITE	S1	S2			
Er3Zn11	FTLITE	S				
Er3Zn22	FTLITE	S				
Er5Mg24	FTLITE	S				
Er5Si3	FTLITE	S				
Er5Si4	FTLITE	S				
Er5Sn3	FTLITE	S				
Er5Sn4	FTLITE	S				
Er6Fe23	FTLITE	S				
Er6Mn23	FTLITE	S				
ErAl	FTLITE	S				
ErAl2	FTLITE	S				
ErAl3	FTLITE	S				
ErCr2Al20	FTLITE	S				
ErCu4Al8	FTLITE	S				

ErFe2	FTLITE	S
ErFe2Al10	FTLITE	S
ErFe3	FTLITE	S
ErMg	FTLITE	S
ErMg2	FTLITE	S
ErMn12	FTLITE	S
ErMn2	FTLITE	S
ErSi	FTLITE	S
ErSn2	FTLITE	S
ErSn3	FTLITE	S1 S2
ErTi2Al20	FTLITE	S
ErV2Al20	FTLITE	S
ErZn	FTLITE	S
ErZn11	FTLITE	S
ErZn12	FTLITE	S
ErZn2	FTLITE	S
ErZn2Mg	FTLITE	S
ErZn3	FTLITE	S
ErZn5	FTLITE	S
ErZn6Mg3	FTLITE	S
ErZnMg12	FTLITE	S
Eu	FTLITE	S1 S2 S3 L
Eu13Zn58	FTLITE	S
Eu2Mg17	FTLITE	S
Eu2Zn17	FTLITE	S1 S2
Eu3Si5	FTLITE	S
Eu3Zn11	FTLITE	S
Eu3Zn22	FTLITE	S
Eu50Si87	FTLITE	S
Eu5Si3	FTLITE	S
Eu5Si4	FTLITE	S
Eu8Si11	FTLITE	S
EuAl	FTLITE	S
EuAl2	FTLITE	S
EuAl4	FTLITE	S
EuCu4Al8	FTLITE	S
EuMg	FTLITE	S
EuMg2	FTLITE	S
EuMg4	FTLITE	S
EuMg5	FTLITE	S
EuSi	FTLITE	S
EuSi2Al2	FTLITE	S
EuSn3	FTLITE	S
EuTi2Al20	FTLITE	S
EuV2Al20	FTLITE	S
EuZn	FTLITE	S
EuZn11	FTLITE	S
EuZn12	FTLITE	S
EuZn13	FTLITE	S
EuZn2	FTLITE	S
EuZn3	FTLITE	S
EuZn5	FTLITE	S
Fe	FTLITE	S1 S2 S3 S4 S5 S6 S7 L
Fe13Mo2B5	FTLITE	S
Fe18Sc10	FTLITE	S
Fe23B6	FTLITE	S
Fe2B	FTLITE	S
Fe2Mo	FTLITE	S
Fe2Moc	FTLITE	S
Fe2N	FTLITE	S
Fe2Nb	FTLITE	S
Fe2O3	FTLITE	S
Fe2P	FTLITE	S
Fe2Sc	FTLITE	S1 S2
Fe2Si	FTLITE	S

Fe2Ti	FTLITE	S
Fe2W	FTLITE	S
Fe3B	FTLITE	S
Fe3B2	FTLITE	S
Fe3C	FTLITE	S1 S2
Fe3C2	FTLITE	S
Fe3N	FTLITE	S
Fe3O4	FTLITE	S1 S2 S3 S4
Fe3P	FTLITE	S
Fe3Si	FTLITE	S
Fe3Sn2	FTLITE	S
Fe5C2	FTLITE	S1 S2
Fe5Si3	FTLITE	S1 S2
Fe5Sn3	FTLITE	S1 S2 S3
Fe6Sc29	FTLITE	S
Fe6W6C	FTLITE	S
Fe7C3	FTLITE	S
Fe7W6	FTLITE	S
Fe9Sc5	FTLITE	S
FeB	FTLITE	S
FeBe12	FTLITE	S
FeBe2	FTLITE	S
FeBe5	FTLITE	S
FeP	FTLITE	S
FeP2	FTLITE	S
FeP4	FTLITE	S
FeSb	FTLITE	S
FeSb2	FTLITE	S
FeSi	FTLITE	S1 S2 S3
FeSi2	FTLITE	S1 S2 S3
FeSn	FTLITE	S1 S2
FeSn2	FTLITE	S
FeSn2Zr6	FTLITE	S
FeTi	FTLITE	S
Ga	FTLITE	S1 S2 S3 S4 S5 L
Ga2Cu	FTLITE	S
Ga2Pt	FTLITE	S1 S2
Ga2Sc	FTLITE	S
Ga2Ti	FTLITE	S
Ga2Zr	FTLITE	S
Ga2Zr3	FTLITE	S
Ga39Na22	FTLITE	S
Ga3Pt2	FTLITE	S
Ga3Pt5	FTLITE	S
Ga3Sc	FTLITE	S
Ga3Sc5	FTLITE	S
Ga3Ti	FTLITE	S
Ga3Ti2	FTLITE	S
Ga3Ti5	FTLITE	S
Ga3Zr	FTLITE	S
Ga3Zr2	FTLITE	S
Ga3Zr5	FTLITE	S
Ga4Cr3	FTLITE	S
Ga4Cu9	FTLITE	S
Ga4Na	FTLITE	S
Ga4Sc5	FTLITE	S
Ga4Zr5	FTLITE	S
Ga5Zr3	FTLITE	S
Ga6Pt	FTLITE	S
Ga7Pt3	FTLITE	S
Ga8Mo3	FTLITE	S
GaAs	FTLITE	S
GaCr	FTLITE	S
GaN	FTLITE	S
GaP	FTLITE	S

GaPt	FTLITE	S
GaPt2	FTLITE	S
GaPt3	FTLITE	S1 S2 S3
GaSb	FTLITE	S
GaSc	FTLITE	S
GaZr	FTLITE	S1 S2
GaZr2	FTLITE	S
Gd	FTLITE	S1 S2 S3 S4 L
Gd11Sn10	FTLITE	S
Gd12Co7	FTLITE	S
Gd13Zn58	FTLITE	S
Gd2Al	FTLITE	S
Gd2Fe17	FTLITE	S
Gd2Zn17	FTLITE	S1 S2
Gd3Al2	FTLITE	S
Gd3Si5	FTLITE	S1 S2
Gd3Sn7	FTLITE	S
Gd3Zn11	FTLITE	S
Gd3Zn22	FTLITE	S
Gd4Sn11	FTLITE	S
Gd5Si3	FTLITE	S
Gd5Si4	FTLITE	S1 S2
Gd5Sn3	FTLITE	S
Gd5Sn4	FTLITE	S
Gd6Fe23	FTLITE	S
Gd6Mn23	FTLITE	S
GdAl	FTLITE	S
GdAl2	FTLITE	S
GdAl3	FTLITE	S
GdAl4Mg	FTLITE	S
GdCr2Al20	FTLITE	S
GdCu4Al8	FTLITE	S
GdFe2	FTLITE	S
GdFe2Al10	FTLITE	S
GdFe3	FTLITE	S
GdMg	FTLITE	S
GdMg2	FTLITE	S
GdMg3	FTLITE	S
GdMg5	FTLITE	S
GdMn12	FTLITE	S
GdMn2	FTLITE	S
GdSi	FTLITE	S
GdSi2	FTLITE	S1 S2
GdSi2Al2	FTLITE	S
GdSn2	FTLITE	S
GdSn3	FTLITE	S1 S2
GdTi2Al20	FTLITE	S
GdV2Al20	FTLITE	S
GdZn	FTLITE	S
GdZn11	FTLITE	S
GdZn12	FTLITE	S
GdZn2	FTLITE	S
GdZn2Mg	FTLITE	S
GdZn3	FTLITE	S
GdZn5	FTLITE	S
GdZn6Mg3	FTLITE	S
GdZnMg12	FTLITE	S
Ge	FTLITE	S1 S2 S3 L
GeAs	FTLITE	S
GeAs2	FTLITE	S
GeCa	FTLITE	S
GeP	FTLITE	S
H	FTLITE	S L
Hf	FTLITE	S1 S2 S3 L
Hf2Cr	FTLITE	S

Hf2Ni	FTLITE	S
Hf2Ni7	FTLITE	S
Hf2Si	FTLITE	S
Hf3Ga2	FTLITE	S
Hf3N2	FTLITE	S
Hf3N4	FTLITE	S
Hf3P	FTLITE	S
Hf3Si2	FTLITE	S
Hf5Si3	FTLITE	S1 S2
Hf5Si4	FTLITE	S
Hf5Sn3	FTLITE	S1 S2
Hf5Sn4	FTLITE	S
Hf7Ni10	FTLITE	S
HfB	FTLITE	S
HfB2	FTLITE	S
HfH2	FTLITE	S
HfN	FTLITE	S
HfNi	FTLITE	S
HfNi3	FTLITE	S
HfNi5	FTLITE	S
HfSi	FTLITE	S
HfSi2	FTLITE	S
HfSn2	FTLITE	S
Ho	FTLITE	S1 S2 S3 L
Ho11Sn10	FTLITE	S
Ho13Zn58	FTLITE	S
Ho2Al	FTLITE	S
Ho2Al10Mg3	FTLITE	S
Ho2Fe17	FTLITE	S
Ho2Sn5	FTLITE	S
Ho2Zn17	FTLITE	S1 S2
Ho3Al2	FTLITE	S
Ho3Si4	FTLITE	S
Ho3Si5	FTLITE	S1 S2
Ho3Zn11	FTLITE	S
Ho3Zn22	FTLITE	S
Ho5Mg24	FTLITE	S
Ho5Si3	FTLITE	S
Ho5Si4	FTLITE	S
Ho5Si9	FTLITE	S1 S2
Ho5Sn3	FTLITE	S
Ho5Sn4	FTLITE	S
Ho6Fe23	FTLITE	S
Ho6Mn23	FTLITE	S
HoAl	FTLITE	S
HoAl2	FTLITE	S
HoAl3	FTLITE	S
HoCr2Al20	FTLITE	S
HoCu4Al8	FTLITE	S
HoFe2	FTLITE	S
HoFe2Al10	FTLITE	S
HoFe3	FTLITE	S
HoMg	FTLITE	S
HoMg2	FTLITE	S
HoMn12	FTLITE	S
HoMn2	FTLITE	S
HoSi	FTLITE	S
HoSi2Al2	FTLITE	S
HoSiAl2	FTLITE	S
HoSn2	FTLITE	S
HoSn3	FTLITE	S1 S2
HoTi2Al20	FTLITE	S
HoZn	FTLITE	S
HoZn11	FTLITE	S
HoZn12	FTLITE	S

HoZn2	FTLITE	S
HoZn2Mg	FTLITE	S
HoZn3	FTLITE	S
HoZn5	FTLITE	S
HoZn6Mg3	FTLITE	S
HoZnMg12	FTLITE	S
In	FTLITE	S1 S2 S3 L
In2Ag	FTLITE	S
In2Ca	FTLITE	S
In2Li3	FTLITE	S
In2Pt	FTLITE	S
In2Pt3	FTLITE	S1 S2
In3Li13	FTLITE	S
In3Pt2	FTLITE	S
In5Na3	FTLITE	S
In5Pt6	FTLITE	S
In7Pt3	FTLITE	S
In9Na5	FTLITE	S
In9Pt13	FTLITE	S
InAg3	FTLITE	S
InCa3	FTLITE	S
InLi2	FTLITE	S
InLi6	FTLITE	S
InMg2	FTLITE	S
InN	FTLITE	S
InNa	FTLITE	S
InNa2	FTLITE	S
InP	FTLITE	S
InPt	FTLITE	S
InPt2	FTLITE	S
InSb	FTLITE	S
InSiNC	FTLITE	S
K	FTLITE	S1 S2 S3 L
K2O	FTLITE	S
K3Bi	FTLITE	S1 S2
K3Ge	FTLITE	S
KB6	FTLITE	S
KGe	FTLITE	S
KGe4	FTLITE	S
KH	FTLITE	S
KSi	FTLITE	S
KZn13	FTLITE	S
La	FTLITE	S1 S2 S3 S4 L
La11Sn10	FTLITE	S
La13Zn58	FTLITE	S
La22Al53	FTLITE	S
La2Mg17	FTLITE	S
La2Ni3	FTLITE	S
La2Ni7	FTLITE	S1 S2
La2Sn3	FTLITE	S
La2Zn17	FTLITE	S1 S2
La3Al	FTLITE	S
La3Al11	FTLITE	S1 S2
La3Al40Mg17	FTLITE	S
La3Ga2	FTLITE	S
La3Ni	FTLITE	S
La3Si2	FTLITE	S
La3Si5	FTLITE	S
La3Sn5	FTLITE	S
La3Zn11	FTLITE	S
La3Zn22	FTLITE	S
La5Mg41	FTLITE	S
La5Si3	FTLITE	S1 S2
La5Si4	FTLITE	S1 S2
La5Sn3	FTLITE	S1 S2

La5Sn4	FTLITE	S
La5Zn5Mg42	FTLITE	S
La7Ni16	FTLITE	S
La7Ni3	FTLITE	S
LaAl	FTLITE	S
LaAl2	FTLITE	S
LaAl3	FTLITE	S
LaCr2Al20	FTLITE	S
LaCu2Al10	FTLITE	S
LaCu4Al8	FTLITE	S
LaFe2Al10	FTLITE	S
LaMg	FTLITE	S
LaMg12	FTLITE	S
LaMg2	FTLITE	S
LaMg3	FTLITE	S
LaNi	FTLITE	S
LaNi3	FTLITE	S
LaNi5	FTLITE	S
LaSi	FTLITE	S
LaSi2	FTLITE	S
LaSi2Al2	FTLITE	S
LaSn	FTLITE	S
LaSn3	FTLITE	S
LaTi2Al20	FTLITE	S
LaV2Al20	FTLITE	S
LaZn	FTLITE	S
LaZn11	FTLITE	S
LaZn12	FTLITE	S
LaZn13	FTLITE	S
LaZn2	FTLITE	S
LaZn2Al2	FTLITE	S
LaZn2Mg	FTLITE	S
LaZn3	FTLITE	S
LaZn4	FTLITE	S
LaZn5	FTLITE	S
Li	FTLITE	S1    S2    S3    L
Li12Si7	FTLITE	S
Li13Si4	FTLITE	S
Li22Si5	FTLITE	S
Li23Sr6	FTLITE	S
Li2Ga	FTLITE	S
Li2Ga7	FTLITE	S
Li2LaH4	FTLITE	S
Li2O	FTLITE	S
Li2Sb	FTLITE	S
Li2Sr3	FTLITE	S
Li3AlH6	FTLITE	S
Li3Bi	FTLITE	S
Li3Ga14	FTLITE	S
Li3Ga2	FTLITE	S
Li3Ga8	FTLITE	S
Li3Sb	FTLITE	S1    S2
Li4Ba	FTLITE	S
Li5Ga4	FTLITE	S
Li5Ga9	FTLITE	S
Li7Si3	FTLITE	S
LiAlH4	FTLITE	S
LiB3	FTLITE	S
LiGa	FTLITE	S
LiH	FTLITE	S
LiPb	FTLITE	S1    S2
LiZn	FTLITE	S
LiZn4	FTLITE	S
Lu	FTLITE	S1    S2    S3    L
Lu11Sn10	FTLITE	S

Lu13Zn58	FTLITE	S
Lu2Al	FTLITE	S
Lu2Fe17	FTLITE	S
Lu2Sn5	FTLITE	S
Lu2Zn17	FTLITE	S1 S2
Lu3Al2	FTLITE	S
Lu3Si5	FTLITE	S
Lu3Zn11	FTLITE	S
Lu3Zn22	FTLITE	S
Lu5Mg24	FTLITE	S
Lu5Si3	FTLITE	S
Lu5Si4	FTLITE	S
Lu5Sn3	FTLITE	S
Lu6Fe23	FTLITE	S
Lu6Mn23	FTLITE	S
LuAl	FTLITE	S
LuAl2	FTLITE	S
LuAl3	FTLITE	S
LuCu4Al8	FTLITE	S
LuFe2	FTLITE	S
LuFe2Al10	FTLITE	S
LuFe3	FTLITE	S
LuMg	FTLITE	S
LuMg2	FTLITE	S
LuMn2	FTLITE	S
LuMn5	FTLITE	S
LuSi	FTLITE	S
LuSn2	FTLITE	S
LuV2Al20	FTLITE	S
LuZn	FTLITE	S
LuZn11	FTLITE	S
LuZn12	FTLITE	S
LuZn2	FTLITE	S
LuZn2Mg	FTLITE	S
LuZn3	FTLITE	S
LuZn5	FTLITE	S
LuZn6Mg3	FTLITE	S
LuZnMg12	FTLITE	S
Mg	FTLITE	S1 S2 S3 S4 S5 S6 S7 L
Mg11Si7Al3	FTLITE	S
Mg12Zn13	FTLITE	S
Mg17Ba2	FTLITE	S
Mg23Ba6	FTLITE	S
Mg2Ba	FTLITE	S1 S2
Mg2Ca	FTLITE	S1 S2
Mg2Cu	FTLITE	S1 S2
Mg2Ga	FTLITE	S
Mg2Ga5	FTLITE	S
Mg2Ge	FTLITE	S
Mg2Ni	FTLITE	S
Mg2Pb	FTLITE	S
Mg2Pt	FTLITE	S
Mg2Si	FTLITE	S
Mg2Si6Al3	FTLITE	S1 S2
Mg2Sn	FTLITE	S
Mg2Zn	FTLITE	S
Mg2Zn11	FTLITE	S
Mg2Zn3	FTLITE	S
Mg3Bi2	FTLITE	S1 S2
Mg3N2	FTLITE	S1 S2 S3
Mg3P2	FTLITE	S
Mg3Pt	FTLITE	S
Mg3Sb2	FTLITE	S1 S2
Mg3Si6Al2	FTLITE	S1 S2
Mg4Si4Al4	FTLITE	S

Mg4Si6Al	FTLITE	S1	S2
Mg4Si6Al2	FTLITE	S	
Mg4Si7	FTLITE	S	
Mg4Si8	FTLITE	S	
Mg51Zn20	FTLITE	S	
Mg5Ga2	FTLITE	S	
Mg5In2	FTLITE	S	
Mg5Si6	FTLITE	S1	S2
Mg6Pt	FTLITE	S	
Mg6Si3	FTLITE	S	
Mg6Si4	FTLITE	S	
Mg9Si5	FTLITE	S	
Mg9Si7Al3	FTLITE	S	
Mg9Si7Al5	FTLITE	S	
Mg9Si9Al3	FTLITE	S	
MgAl2O4	FTLITE	S	
MgB2	FTLITE	S	
MgB20	FTLITE	S	
MgB4	FTLITE	S	
MgB7	FTLITE	S	
MgBe13	FTLITE	S	
MgCo2	FTLITE	S	
MgGa	FTLITE	S	
MgGa2	FTLITE	S	
MgH2	FTLITE	S	
MgNi2	FTLITE	S	
MgO	FTLITE	S	
MgPt	FTLITE	S	
MgPt3	FTLITE	S	
MgPt7	FTLITE	S	
MgSi	FTLITE	S	
MgSi2Al2	FTLITE	S	
MgSi6Al4	FTLITE	S1	S2
MgZn2	FTLITE	S1	S2
Mn	FTLITE	S1	S2
Mn11Si19	FTLITE	S	
Mn19Sn6	FTLITE	S	
Mn23C6	FTLITE	S	
Mn23Sc6	FTLITE	S	
Mn2B	FTLITE	S	
Mn2Bi	FTLITE	S	
Mn2CaAl10	FTLITE	S	
Mn2N	FTLITE	S	
Mn2P	FTLITE	S	
Mn2Sb	FTLITE	S1	S2
Mn2Sc	FTLITE	S	
Mn2Sn	FTLITE	S	
Mn2Ti	FTLITE	S	
Mn2Ti2	FTLITE	S	
Mn2Zr	FTLITE	S	
Mn3AlC	FTLITE	S	
Mn3B4	FTLITE	S	
Mn3C	FTLITE	S	
Mn3C2	FTLITE	S	
Mn3N	FTLITE	S	
Mn3Ni	FTLITE	S	
Mn3P	FTLITE	S	
Mn3Si	FTLITE	S	
Mn3Ti	FTLITE	S	
Mn5C2	FTLITE	S	
Mn5Si3	FTLITE	S1	S2
Mn5SiC	FTLITE	S	
Mn5Sn3	FTLITE	S	
Mn6N4	FTLITE	S	
Mn6N5	FTLITE	S	

Mn6Si	FTLITE	S				
Mn7C3	FTLITE	S				
Mn7Mo6	FTLITE	S				
Mn8Si2C	FTLITE	S				
Mn9Si2	FTLITE	S				
'Mn9Ti2'	FTLITE	S				(Mn) 8.965 (Ti) 2.035
MnB	FTLITE	S				
MnB2	FTLITE	S				
MnB4	FTLITE	S				
MnBe12	FTLITE	S				
MnBe2	FTLITE	S				
MnBe5	FTLITE	S				
MnBi	FTLITE	S				
MnCa4Al7	FTLITE	S				
MnN	FTLITE	S				
MnNi2	FTLITE	S				
MnNi3	FTLITE	S				
MnO	FTLITE	S				
MnP	FTLITE	S				
MnSc4	FTLITE	S				
MnSi	FTLITE	S				
MnSn2	FTLITE	S				
'MnTi'	FTLITE	S				(Mn) 1.03 (Ti) 0.97
Mo	FTLITE	S1	S2	S3	L	
Mo23C6	FTLITE	S				
Mo2B	FTLITE	S				
Mo2B5	FTLITE	S				
Mo2BC	FTLITE	S				
Mo2C	FTLITE	S				
Mo2Hf	FTLITE	S1	S2			
Mo2Zr	FTLITE	S				
Mo3B2	FTLITE	S				
Mo3Ga	FTLITE	S				
Mo3Ni10B11	FTLITE	S				
Mo3Ni11	FTLITE	S				
Mo3P	FTLITE	S1	S2			
Mo3Si	FTLITE	S				
Mo4Ni3	FTLITE	S				
Mo5Si3	FTLITE	S				
Mo5Sn3	FTLITE	S				
MoB	FTLITE	S1	S2			
MoB12	FTLITE	S				
MoB2	FTLITE	S1	S2			
MoB4	FTLITE	S				
MoC	FTLITE	S				
MoCo2B4	FTLITE	S				
MoCoB	FTLITE	S				
MoGa4	FTLITE	S				
MoN	FTLITE	S1	S2			
MoNi4	FTLITE	S				
MoP	FTLITE	S				
MoPt	FTLITE	S				
MoPt2	FTLITE	S				
MoSi2	FTLITE	S1	S2			
MoZn22	FTLITE	S				
MoZn7	FTLITE	S				
N	FTLITE	L				
Na	FTLITE	S1	S2	S3	L	
Na2C2	FTLITE	S1	S2			
Na2K	FTLITE	S				
Na2LiAlH6	FTLITE	S				
Na2O	FTLITE	S1	S2	S3		
Na3AlH6	FTLITE	S1	S2			
Na3As	FTLITE	S				
Na3Bi	FTLITE	S				

Na3Ge	FTLITE	S
Na3P	FTLITE	S
Na3Sb	FTLITE	S
Na4Ba	FTLITE	S
NaAlH <sub>4</sub>	FTLITE	S
NaB <sub>15</sub>	FTLITE	S
NaB <sub>6</sub>	FTLITE	S
NaBa	FTLITE	S
NaGe	FTLITE	S
NaH	FTLITE	S
NaMgH <sub>3</sub>	FTLITE	S
NaSi	FTLITE	S
NaZn <sub>13</sub>	FTLITE	S
Nb	FTLITE	S L
Nb <sub>2</sub> AlC	FTLITE	S
Nb <sub>2</sub> B <sub>3</sub>	FTLITE	S
Nb <sub>2</sub> C	FTLITE	S
Nb <sub>2</sub> N	FTLITE	S
Nb <sub>2</sub> Zn <sub>3</sub>	FTLITE	S
Nb <sub>3</sub> B <sub>2</sub>	FTLITE	S
Nb <sub>3</sub> B <sub>4</sub>	FTLITE	S
Nb <sub>3</sub> C <sub>2</sub>	FTLITE	S
Nb <sub>3</sub> P	FTLITE	S
Nb <sub>4</sub> FeSi	FTLITE	S
Nb <sub>5</sub> B <sub>6</sub>	FTLITE	S
Nb <sub>5</sub> Si <sub>3</sub>	FTLITE	S1 S2
Nb <sub>5</sub> Sn <sub>3</sub>	FTLITE	S
Nb <sub>6</sub> Fe <sub>7</sub>	FTLITE	S
NbB	FTLITE	S
NbB <sub>2</sub>	FTLITE	S
NbN	FTLITE	S
NbNi <sub>8</sub>	FTLITE	S
NbSn <sub>2</sub>	FTLITE	S
NbZn	FTLITE	S
NbZn <sub>12</sub> Fe	FTLITE	S
NbZn <sub>15</sub>	FTLITE	S
NbZn <sub>2</sub>	FTLITE	S
NbZn <sub>3</sub>	FTLITE	S
NbZn <sub>7</sub>	FTLITE	S
Nd	FTLITE	S1 S2 S3 S4 L
Nd <sub>11</sub> Sn <sub>10</sub>	FTLITE	S
Nd <sub>13</sub> Zn <sub>58</sub>	FTLITE	S
Nd <sub>20</sub> Mg <sub>19</sub> Zn <sub>81</sub>	FTLITE	S
Nd <sub>2</sub> Al	FTLITE	S
Nd <sub>2</sub> Fe <sub>17</sub>	FTLITE	S
Nd <sub>2</sub> Sb	FTLITE	S
Nd <sub>2</sub> Si <sub>3</sub>	FTLITE	S1 S2
Nd <sub>2</sub> Sn <sub>5</sub>	FTLITE	S
Nd <sub>2</sub> Zn <sub>17</sub>	FTLITE	S1 S2
Nd <sub>2</sub> Zn <sub>9</sub> Mg <sub>5</sub>	FTLITE	S
Nd <sub>3</sub> Al	FTLITE	S
Nd <sub>3</sub> Al <sub>11</sub>	FTLITE	S1 S2
Nd <sub>3</sub> Al <sub>50</sub> Mg <sub>22</sub>	FTLITE	S
Nd <sub>3</sub> Si <sub>4</sub>	FTLITE	S
Nd <sub>3</sub> Sn <sub>5</sub>	FTLITE	S
Nd <sub>3</sub> Sn <sub>7</sub>	FTLITE	S
Nd <sub>3</sub> Zn <sub>11</sub>	FTLITE	S
Nd <sub>3</sub> Zn <sub>22</sub>	FTLITE	S
Nd <sub>3</sub> Zn <sub>30</sub> Mg <sub>13</sub>	FTLITE	S
Nd <sub>4</sub> Sb <sub>3</sub>	FTLITE	S
Nd <sub>5</sub> Fe <sub>17</sub>	FTLITE	S
Nd <sub>5</sub> Mg <sub>41</sub>	FTLITE	S
Nd <sub>5</sub> Sb <sub>3</sub>	FTLITE	S
Nd <sub>5</sub> Si <sub>3</sub>	FTLITE	S1 S2
Nd <sub>5</sub> Si <sub>4</sub>	FTLITE	S1 S2

Nd5Si9	FTLITE	S
Nd5Sn3	FTLITE	S
Nd5Sn4	FTLITE	S
Nd6Mn23	FTLITE	S
NdAl1	FTLITE	S
NdAl2	FTLITE	S
NdAl3	FTLITE	S
NdB6	FTLITE	S
NdCr2Al20	FTLITE	S
NdCu4Al8	FTLITE	S
NdFe2Al10	FTLITE	S
NdMg	FTLITE	S
NdMg2	FTLITE	S
NdMg3	FTLITE	S
NdMn2	FTLITE	S
NdSb	FTLITE	S
NdSb2	FTLITE	S
NdSi	FTLITE	S
NdSi2Al2	FTLITE	S
NdSn	FTLITE	S
NdSn2	FTLITE	S
NdSn3	FTLITE	S
NdTl2Al20	FTLITE	S
NdV2Al20	FTLITE	S
NdZn	FTLITE	S
NdZn11	FTLITE	S
NdZn12	FTLITE	S
NdZn2	FTLITE	S
NdZn2Al2	FTLITE	S
NdZn2Mg	FTLITE	S
NdZn3	FTLITE	S
NdZn5	FTLITE	S
Ni	FTLITE	S1    S2    S3    S4    S5    L
Ni10Mn3Si7	FTLITE	S
Ni11As8	FTLITE	S
Ni12P5	FTLITE	S
Ni13Ga9	FTLITE	S
Ni16Mn6Si7	FTLITE	S
Ni16Si7Ti6	FTLITE	S
Ni23C6	FTLITE	S
Ni2B	FTLITE	S
Ni2Cr	FTLITE	S
Ni2Ga	FTLITE	S
Ni2Ga3	FTLITE	S
Ni2Ge	FTLITE	S1    S2
Ni2In	FTLITE	S
Ni2In3	FTLITE	S
Ni2Mg3Al	FTLITE	S
Ni2Mn3Si	FTLITE	S
Ni2MnMg3	FTLITE	S
Ni2Mo	FTLITE	S
Ni2P	FTLITE	S
Ni2Si	FTLITE	S1    S2
Ni2Ta	FTLITE	S
Ni2V	FTLITE	S
Ni2Y	FTLITE	S
Ni2Y3	FTLITE	S
Ni3B	FTLITE	S
Ni3Ga	FTLITE	S
Ni3Ga4	FTLITE	S
Ni3Ga7	FTLITE	S
Ni3Ge	FTLITE	S
Ni3In	FTLITE	S
Ni3In7	FTLITE	S
Ni3Mo	FTLITE	S

Ni3P	FTLITE	S
Ni3Sb	FTLITE	S1 S2
Ni3Si	FTLITE	S1 S2
Ni3Si2	FTLITE	S
Ni3SiTi2	FTLITE	S
Ni3Sn	FTLITE	S1 S2
Ni3Sn2	FTLITE	S
Ni3Ta	FTLITE	S
Ni3Ti	FTLITE	S
Ni3V	FTLITE	S
Ni3Y	FTLITE	S
Ni49Si37Ti14	FTLITE	S
Ni4B3	FTLITE	S
Ni4Si7Ti4	FTLITE	S
Ni4W	FTLITE	S
Ni4Y	FTLITE	S
Ni5AlB4	FTLITE	S
Ni5As2	FTLITE	S
Ni5Ga3	FTLITE	S
Ni5Ge2	FTLITE	S
Ni5Ge3	FTLITE	S
Ni5P2	FTLITE	S
Ni5Sb2	FTLITE	S
Ni5Si2	FTLITE	S
'Ni5Si2B'	FTLITE	S
'Ni5Zr4'	FTLITE	S
Ni6P5	FTLITE	S
Ni6Si2B	FTLITE	S
Ni7Sc2	FTLITE	S
Ni7Y2	FTLITE	S
'Ni7Zr2'	FTLITE	S
Ni8AlB11	FTLITE	S
Ni8Ta	FTLITE	S
'Ni8Y'	FTLITE	S
'Ni8Zr3'	FTLITE	S
Ni9Mn3Si8	FTLITE	S
NiAs	FTLITE	S
NiAs2	FTLITE	S1 S2
NiB	FTLITE	S
NiGa	FTLITE	S1 S2
NiGe	FTLITE	S1 S2
NiIn	FTLITE	S
NiMnSi	FTLITE	S
NiO	FTLITE	S
NiP2	FTLITE	S
NiSb	FTLITE	S
NiSb2	FTLITE	S
NiSc	FTLITE	S
NiSi	FTLITE	S1 S2
NiSi2	FTLITE	S
NiSi4Ti4	FTLITE	S
NiSiTi	FTLITE	S
NiSn	FTLITE	S1 S2
NiSr	FTLITE	S
NiTi2	FTLITE	S1 S2
NiV3	FTLITE	S
NiW	FTLITE	S
NiW2	FTLITE	S
NiY	FTLITE	S
NiY3	FTLITE	S
NiZn8	FTLITE	S
NiZr	FTLITE	S
NiZr2	FTLITE	S
P	FTLITE	S1 S2 S3 S4 L
Pb	FTLITE	S1 S2 S3 S4 S5 S6 S7 L

Pb10Nd11	FTLITE	S
Pb10Pr11	FTLITE	S
Pb2Au	FTLITE	S
Pb2Dy	FTLITE	S
Pb2Nd	FTLITE	S
Pb2Pr	FTLITE	S
Pb2Tb	FTLITE	S
Pb2Zr	FTLITE	S
Pb3Au	FTLITE	S
Pb3Dy	FTLITE	S
Pb3Dy5	FTLITE	S
Pb3Nd	FTLITE	S
Pb3Nd5	FTLITE	S
Pb3O4	FTLITE	S
Pb3Pr	FTLITE	S
Pb3Pr5	FTLITE	S
Pb3Sr	FTLITE	S
Pb3Sr2	FTLITE	S
Pb3Sr5	FTLITE	S
Pb3Tb	FTLITE	S
Pb3Tb5	FTLITE	S
Pb3Zr5	FTLITE	S
Pb4Dy5	FTLITE	S
Pb4Nd3	FTLITE	S
Pb4Nd5	FTLITE	S
Pb4Pr3	FTLITE	S
Pb4Pr5	FTLITE	S
Pb4Sr5	FTLITE	S
Pb4Tb5	FTLITE	S
Pb5Li22	FTLITE	S
Pb5Sr3	FTLITE	S
PbDy	FTLITE	S
PbNd3	FTLITE	S
PbO	FTLITE	S1      S2
PbO2	FTLITE	S
PbPr3	FTLITE	S
PbSr	FTLITE	S
PbSr2	FTLITE	S
PbTb	FTLITE	S
PbZr4	FTLITE	S
Pr	FTLITE	S1      S2      S3      S4      L
Pr11Sn10	FTLITE	S
Pr13Zn58	FTLITE	S
Pr2Al	FTLITE	S
Pr2Fe17	FTLITE	S
Pr2Sb	FTLITE	S
Pr2Sn3	FTLITE	S
Pr2Sn5	FTLITE	S
Pr2Zn17	FTLITE	S1      S2
Pr2Zn9Mg5	FTLITE	S
Pr3Al	FTLITE	S1      S2
Pr3Al11	FTLITE	S1      S2
Pr3Al50Mg22	FTLITE	S
Pr3Si4	FTLITE	S
Pr3Sn	FTLITE	S
Pr3Sn5	FTLITE	S1      S2
Pr3Sn7	FTLITE	S
Pr3Zn11	FTLITE	S
Pr3Zn22	FTLITE	S
Pr3Zn30Mg13	FTLITE	S
Pr4Sb3	FTLITE	S
Pr5Mg41	FTLITE	S
Pr5Sb3	FTLITE	S
Pr5Si3	FTLITE	S1      S2
Pr5Si4	FTLITE	S1      S2

Pr5Si9	FTLITE	S1	S2	
Pr5Sn3	FTLITE	S1	S2	
Pr5Sn4	FTLITE	S1	S2	
Pr6Mn23	FTLITE	S		
PrAl1	FTLITE	S		
PrAl2	FTLITE	S		
PrAl3	FTLITE	S		
PrCr2Al20	FTLITE	S		
PrCu4Al8	FTLITE	S		
PrFe2Al10	FTLITE	S		
PrMg	FTLITE	S		
PrMg12	FTLITE	S		
PrMg2	FTLITE	S		
PrMg3	FTLITE	S		
PrSb	FTLITE	S		
PrSb2	FTLITE	S		
PrSi	FTLITE	S		
PrSi2Al2	FTLITE	S		
PrSn	FTLITE	S		
PrSn2	FTLITE	S		
PrSn3	FTLITE	S		
PrTi2Al20	FTLITE	S		
PrV2Al20	FTLITE	S		
PrZn	FTLITE	S		
PrZn11	FTLITE	S		
PrZn12	FTLITE	S		
PrZn2	FTLITE	S		
PrZn2Al2	FTLITE	S		
PrZn2Mg	FTLITE	S		
PrZn3	FTLITE	S		
PrZn5	FTLITE	S		
Pt	FTLITE	S1	S2	L
Pt17Si8	FTLITE	S1	S2	
Pt25Si7	FTLITE	S		
Pt2Ge	FTLITE	S		
Pt2Ge3	FTLITE	S		
Pt2Si	FTLITE	S1	S2	
Pt2Sn3	FTLITE	S		
'Pt2Ta'	FTLITE	S		(Pt) 2.001 (Ta) 0.999
Pt2V	FTLITE	S		
Pt3Ge	FTLITE	S		
Pt3Ge2	FTLITE	S		
Pt3Pb	FTLITE	S		
Pt3Si	FTLITE	S1	S2	
Pt3Sn	FTLITE	S		
Pt3Ta	FTLITE	S		
Pt3Ti	FTLITE	S		
Pt3Ti4	FTLITE	S		
Pt5Si2	FTLITE	S		
Pt6Si5	FTLITE	S		
Pt8Ti	FTLITE	S		
PtAs2	FTLITE	S		
PtGe	FTLITE	S		
PtGe2	FTLITE	S		
PtPb	FTLITE	S		
PtPb4	FTLITE	S		
PtSb2	FTLITE	S		
PtSi	FTLITE	S		
PtSn	FTLITE	S		
PtSn2	FTLITE	S		
PtSn4	FTLITE	S		
PtTa	FTLITE	S		
'PtTa6'	FTLITE	S		(Pt) 1.001 (Ta) 5.999
PtTi	FTLITE	S1	S2	
PtTi3	FTLITE	S		

Sb	FTLITE	S1	S2	S3	L				
Sb <sub>2</sub> Sn <sub>3</sub>	FTLITE	S							
Sb <sub>2</sub> SnZn	FTLITE	S							
SbFe <sub>2</sub>	FTLITE	S							
SbNi <sub>2</sub>	FTLITE	S							
Sc	FTLITE	S1	S2	S3	S4	L			
Sc <sub>2</sub> Si <sub>3</sub>	FTLITE	S							
Sc <sub>5</sub> Si <sub>3</sub>	FTLITE	S							
Sc <sub>5</sub> Si <sub>4</sub>	FTLITE	S1	S2						
ScCu <sub>4</sub> Al <sub>8</sub>	FTLITE	S							
ScSi	FTLITE	S							
ScV <sub>2</sub> Al <sub>20</sub>	FTLITE	S							
Si	FTLITE	S1	S2	S3	S4	S5	S6	S7	L
Si <sub>2</sub> V	FTLITE	S							
Si <sub>2</sub> W	FTLITE	S							
Si <sub>2</sub> Zr <sub>3</sub>	FTLITE	S							
Si <sub>3</sub> N <sub>4</sub>	FTLITE	S							
Si <sub>3</sub> Ti	FTLITE	S							
Si <sub>3</sub> Zr <sub>5</sub>	FTLITE	S1	S2						
Si <sub>4</sub> Zr <sub>5</sub>	FTLITE	S							
Si <sub>5</sub> V <sub>6</sub>	FTLITE	S							
Si <sub>6</sub> Al <sub>15</sub>	FTLITE	S1	S2						
SiAs	FTLITE	S							
SiAs <sub>2</sub>	FTLITE	S							
SiC	FTLITE	S1	S2	S3	S4	S5	S6		
SiO <sub>2</sub>	FTLITE	S1	S2	S3	S4	S5	S6	S7	S8
SiP	FTLITE	S1	S2						
SiP <sub>2</sub>	FTLITE	S							
SiTa <sub>3</sub>	FTLITE	S							
SiZr	FTLITE	S							
SiZr <sub>2</sub>	FTLITE	S							
Sm	FTLITE	S1	S2	S3	S4	L			
Sm <sub>11</sub> Sn <sub>10</sub>	FTLITE	S							
Sm <sub>13</sub> Zn <sub>58</sub>	FTLITE	S							
Sm <sub>2</sub> Al <sub>1</sub>	FTLITE	S							
Sm <sub>2</sub> Fe <sub>17</sub>	FTLITE	S							
Sm <sub>2</sub> Sb	FTLITE	S							
Sm <sub>2</sub> Sn <sub>3</sub>	FTLITE	S							
Sm <sub>2</sub> Sn <sub>5</sub>	FTLITE	S							
Sm <sub>2</sub> Zn <sub>17</sub>	FTLITE	S1	S2						
Sm <sub>2</sub> Zn <sub>9</sub> Mg <sub>5</sub>	FTLITE	S							
Sm <sub>3</sub> Al <sub>11</sub>	FTLITE	S							
Sm <sub>3</sub> Si <sub>5</sub>	FTLITE	S							
Sm <sub>3</sub> Sn <sub>7</sub>	FTLITE	S							
Sm <sub>3</sub> Zn <sub>11</sub>	FTLITE	S							
Sm <sub>3</sub> Zn <sub>22</sub>	FTLITE	S							
Sm <sub>3</sub> Zn <sub>30</sub> Mg <sub>13</sub>	FTLITE	S							
Sm <sub>4</sub> Sb <sub>3</sub>	FTLITE	S							
Sm <sub>4</sub> Sn <sub>3</sub>	FTLITE	S							
Sm <sub>5</sub> Mg <sub>41</sub>	FTLITE	S							
Sm <sub>5</sub> Sb <sub>3</sub>	FTLITE	S							
Sm <sub>5</sub> Si <sub>3</sub>	FTLITE	S							
Sm <sub>5</sub> Si <sub>4</sub>	FTLITE	S1	S2						
Sm <sub>5</sub> Sn <sub>3</sub>	FTLITE	S							
Sm <sub>5</sub> Sn <sub>4</sub>	FTLITE	S							
Sm <sub>6</sub> Mn <sub>23</sub>	FTLITE	S							
SmAl <sub>1</sub>	FTLITE	S							
SmAl <sub>2</sub>	FTLITE	S							
SmAl <sub>3</sub>	FTLITE	S							
SmCr <sub>2</sub> Al <sub>20</sub>	FTLITE	S							
SmCu <sub>4</sub> Al <sub>8</sub>	FTLITE	S							
SmFe <sub>2</sub>	FTLITE	S							
SmFe <sub>2</sub> Al <sub>10</sub>	FTLITE	S							
SmFe <sub>3</sub>	FTLITE	S							
SmMg	FTLITE	S							

SmMg2	FTLITE	S
SmMg3	FTLITE	S
SmMg5	FTLITE	S
SmMn2	FTLITE	S
SmSb	FTLITE	S
SmSb2	FTLITE	S
SmSi	FTLITE	S
SmSi2	FTLITE	S1 S2
SmSi2Al2	FTLITE	S
SmSn2	FTLITE	S
SmSn3	FTLITE	S
SmTi2Al20	FTLITE	S
SmV2Al20	FTLITE	S
SmZn	FTLITE	S
SmZn11	FTLITE	S
SmZn12	FTLITE	S
SmZn2	FTLITE	S
SmZn2Mg	FTLITE	S
SmZn3	FTLITE	S
SmZn5	FTLITE	S
Sn	FTLITE	S1 S2 S3 S4 S5 L
Sn14Ca12Mg7	FTLITE	S
Sn20Ca31	FTLITE	S
Sn23Ca36	FTLITE	S
Sn24Sr14Mg25	FTLITE	S
Sn2Li5	FTLITE	S
Sn2Na	FTLITE	S
Sn2Zr	FTLITE	S
Sn3Ca	FTLITE	S
Sn3Ca5	FTLITE	S
Sn3Li7	FTLITE	S
Sn3Li8	FTLITE	S
Sn3Na	FTLITE	S
Sn3Na4	FTLITE	S
Sn3Sr	FTLITE	S
Sn3Sr5	FTLITE	S
Sn3SrMg5	FTLITE	S
Sn4As3	FTLITE	S
Sn4Li17	FTLITE	S
Sn4Na	FTLITE	S
Sn4Na15	FTLITE	S
Sn4Na9	FTLITE	S
Sn4P3	FTLITE	S
Sn4Sr	FTLITE	S
Sn5Li13	FTLITE	S
Sn5Li2	FTLITE	S
Sn5Sr3	FTLITE	S
Sn6Na	FTLITE	S
SnAg3	FTLITE	S
SnAs	FTLITE	S
SnCa	FTLITE	S
SnFe2	FTLITE	S
SnLi	FTLITE	S
SnMgLi4	FTLITE	S
SnNa	FTLITE	S1 S2
SnNa3	FTLITE	S
SnNi2	FTLITE	S
SnSr	FTLITE	S
SnSr2	FTLITE	S
SnSrMg	FTLITE	S
SnTi2	FTLITE	S
SnZr4	FTLITE	S
Sr	FTLITE	S1 S2 S3 L
Sr18Zn62Mg20	FTLITE	S
Sr2AlH7	FTLITE	S

Sr2Ge	FTLITE	S
Sr2Mg17	FTLITE	S
Sr2Si	FTLITE	S
Sr2Zn43Mg55	FTLITE	S
Sr35Zn21Mg44	FTLITE	S
Sr4Al138Mg58	FTLITE	S
Sr5Ge3	FTLITE	S
Sr5Si3	FTLITE	S
Sr6Mg23	FTLITE	S
Sr8Ga7	FTLITE	S
Sr9Mg38	FTLITE	S
SrAlH5	FTLITE	S
SrB6	FTLITE	S
SrBe13	FTLITE	S
SrC2	FTLITE	S1 S2
SrGa2	FTLITE	S
SrGa4	FTLITE	S
SrGe	FTLITE	S
SrGe2	FTLITE	S
SrH2	FTLITE	S1 S2
SrMg2	FTLITE	S
SrSi	FTLITE	S
SrSi2	FTLITE	S1 S2
SrSi2Al2	FTLITE	S
SrZn	FTLITE	S
SrZn13	FTLITE	S
SrZn2	FTLITE	S
SrZn5	FTLITE	S1 S2
Ta	FTLITE	S1 S2 L
Ta2B	FTLITE	S
Ta2C	FTLITE	S
Ta2Co	FTLITE	S
Ta2N	FTLITE	S
Ta2Si	FTLITE	S
Ta2Sn3	FTLITE	S
Ta3B2	FTLITE	S
Ta3B4	FTLITE	S
Ta3P	FTLITE	S
Ta3Sn	FTLITE	S
'Ta4C3'	FTLITE	S
Ta5B6	FTLITE	S
Ta5Si3	FTLITE	S1 S2 S3
Ta5Sn3	FTLITE	S
TaB	FTLITE	S1 S2
TaB2	FTLITE	S
TaN	FTLITE	S1 S2
TaN3	FTLITE	S
Tb	FTLITE	S1 S2 S3 L
Tb11Sn10	FTLITE	S
Tb13Zn58	FTLITE	S
Tb2Al	FTLITE	S
Tb2Fe17	FTLITE	S
Tb2Zn17	FTLITE	S1 S2
Tb3Al2	FTLITE	S
Tb3Si5	FTLITE	S
Tb3Sn7	FTLITE	S1 S2
Tb3Zn11	FTLITE	S
Tb3Zn22	FTLITE	S
Tb5Mg24	FTLITE	S
Tb5Si3	FTLITE	S
Tb5Si4	FTLITE	S
Tb5Sn3	FTLITE	S
Tb5Sn4	FTLITE	S
Tb6Fe23	FTLITE	S
Tb6Mn23	FTLITE	S

(Ta) 4 (C) 2 . 66666666666667

TbAl	FTLITE	S
TbAl2	FTLITE	S
TbAl3	FTLITE	S
TbAl4Mg	FTLITE	S
TbCr2Al20	FTLITE	S
TbCu4Al8	FTLITE	S
TbFe2	FTLITE	S
TbFe2Al10	FTLITE	S
TbFe3	FTLITE	S
TbMg	FTLITE	S
TbMg2	FTLITE	S
TbMg3	FTLITE	S
TbMn12	FTLITE	S
TbMn2	FTLITE	S
TbSi	FTLITE	S
TbSi2	FTLITE	S
TbSi2Al2	FTLITE	S
TbSn2	FTLITE	S
TbSn3	FTLITE	S
TbTi2Al20	FTLITE	S
TbV2Al20	FTLITE	S
TbZn	FTLITE	S
TbZn11	FTLITE	S
TbZn12	FTLITE	S
TbZn2	FTLITE	S
TbZn2Mg	FTLITE	S
TbZn3	FTLITE	S
TbZn5	FTLITE	S
TbZn6Mg3	FTLITE	S
TbZnMg12	FTLITE	S
Ti	FTLITE	S1      S2      S3      S4      S5      S6      S7      L
Ti2Al	FTLITE	S
Ti2AlC	FTLITE	S
Ti2C	FTLITE	S
Ti2Cr	FTLITE	S
Ti2H	FTLITE	S
Ti2Mn	FTLITE	S
Ti2N	FTLITE	S1      S2
Ti2O	FTLITE	S
Ti2O3	FTLITE	S1      S2
Ti2Zn	FTLITE	S
Ti3Al	FTLITE	S
Ti3AlC	FTLITE	S
Ti3AlC2	FTLITE	S
Ti3AlSi5	FTLITE	S
Ti3B4	FTLITE	S
Ti3O2	FTLITE	S
Ti3P	FTLITE	S
Ti3Si	FTLITE	S
Ti3Sn	FTLITE	S
Ti4AlSi7	FTLITE	S
Ti5Si3	FTLITE	S1      S2
Ti5Si4	FTLITE	S
Ti5Sn3	FTLITE	S1      S2
Ti6Sn5	FTLITE	S1      S2
TiB	FTLITE	S
TiB2	FTLITE	S
TiC	FTLITE	S
TiH2	FTLITE	S
TiN	FTLITE	S
TiN3	FTLITE	S
TiO	FTLITE	S
TiO2	FTLITE	S1      S2
TiSi	FTLITE	S
TiSi2	FTLITE	S

TiZn	FTLITE	S
TiZn10	FTLITE	S
TiZn15	FTLITE	S
TiZn2	FTLITE	S
TiZn3	FTLITE	S
TiZn5	FTLITE	S
Tm	FTLITE	S1 S2 S3 L
Tm11Sn10	FTLITE	S
Tm13Zn58	FTLITE	S
Tm2Al	FTLITE	S
Tm2Fe17	FTLITE	S
Tm2Sn5	FTLITE	S
Tm2Zn17	FTLITE	S1 S2
Tm3Al2	FTLITE	S
Tm3Si5	FTLITE	S
Tm3Zn11	FTLITE	S
Tm3Zn22	FTLITE	S
Tm5Mg24	FTLITE	S
Tm5Si3	FTLITE	S
Tm5Si4	FTLITE	S
Tm5Sn3	FTLITE	S
Tm6Fe23	FTLITE	S
Tm6Mn23	FTLITE	S
TmAl	FTLITE	S
TmAl2	FTLITE	S
TmAl3	FTLITE	S
TmCu4Al8	FTLITE	S
TmFe2	FTLITE	S
TmFe2Al10	FTLITE	S
TmFe3	FTLITE	S
TmMg	FTLITE	S
TmMg2	FTLITE	S
TmMn12	FTLITE	S
TmMn2	FTLITE	S
TmSi	FTLITE	S
TmSn2	FTLITE	S
TmSn3	FTLITE	S1 S2
TmZn	FTLITE	S
TmZn11	FTLITE	S
TmZn12	FTLITE	S
TmZn2	FTLITE	S
TmZn2Mg	FTLITE	S
TmZn3	FTLITE	S
TmZn5	FTLITE	S
TmZn6Mg3	FTLITE	S
TmZnMg12	FTLITE	S
V	FTLITE	S1 S2 S3 S4 S5 L
V11Ge8	FTLITE	S
V17Ge31	FTLITE	S
V23C6	FTLITE	S
V2AlC	FTLITE	S
V2B3	FTLITE	S
V2Ga5	FTLITE	S
V2Hf	FTLITE	S
V2N	FTLITE	S
V2Ta	FTLITE	S1 S2
V2Zr	FTLITE	S
V3B2	FTLITE	S
V3B4	FTLITE	S
V3C2	FTLITE	S
V3Ga	FTLITE	S
V3Ge	FTLITE	S
V3P	FTLITE	S
V3Si	FTLITE	S
V4Zn5	FTLITE	S

V5C2	FTLITE	S
V5Ge3	FTLITE	S
V5Si3	FTLITE	S1 S2
V6Ga5	FTLITE	S
V7Al45	FTLITE	S
V8Ga41	FTLITE	S
VB	FTLITE	S
VB2	FTLITE	S
VN	FTLITE	S
VSn2	FTLITE	S
VZn16	FTLITE	S
VZn3	FTLITE	S
W	FTLITE	S1 S2 S3 L
W2B	FTLITE	S
W2B9	FTLITE	S
W2Zr	FTLITE	S
W3C2	FTLITE	S
W5Si3	FTLITE	S1 S2
WB	FTLITE	S1 S2
WC	FTLITE	S
WN	FTLITE	S1 S2
Y	FTLITE	S1 S2 S3 S4 L
Y11Sn10	FTLITE	S
Y13Zn58	FTLITE	S
Y15Zn70Mg15	FTLITE	S
Y2Fe17	FTLITE	S
Y2Mn9Al5	FTLITE	S
Y2Sn5	FTLITE	S
Y2Zn17	FTLITE	S1 S2
Y3Si5	FTLITE	S1 S2
Y3Zn11	FTLITE	S
Y3Zn22	FTLITE	S
Y3Zn30Mg13	FTLITE	S
Y5Si3	FTLITE	S
Y5Si4	FTLITE	S
Y5Sn3	FTLITE	S1 S2
Y5Sn4	FTLITE	S
Y6Fe23	FTLITE	S
Y6Mn23	FTLITE	S
YA12	FTLITE	S
YCr2Al20	FTLITE	S
YCu4Al8	FTLITE	S
YCuAl	FTLITE	S
YFe2	FTLITE	S
YFe2Al10	FTLITE	S
YFe3	FTLITE	S
YMn12	FTLITE	S
YMn2	FTLITE	S
YNi5	FTLITE	S
YSi	FTLITE	S
YSi2	FTLITE	S1 S2
YSi2Al2	FTLITE	S
YSn2	FTLITE	S
YSn3	FTLITE	S1 S2
YV2Al20	FTLITE	S
YZn	FTLITE	S
YZn11	FTLITE	S
YZn12	FTLITE	S
YZn2	FTLITE	S1 S2
YZn2Mg	FTLITE	S
YZn3	FTLITE	S
YZn5	FTLITE	S
YZn6Mg3	FTLITE	S
YZnMg12	FTLITE	S
Yb	FTLITE	S1 S2 S3 S4 L



ZrB2	FTLITE	S
ZrC	FTLITE	S
ZrH2	FTLITE	S
ZrN	FTLITE	S
ZrSi2	FTLITE	S

## Modeling

In the thermodynamic assessments, the liquid phase has been described using the Modified Quasichemical Model in the Pair Approximation (MQMPA) in order to evaluate short-range order (SRO) between the elements. Within this model, it is possible to force some binary interactions to respect a Bragg-Williams (BW) type of configurational entropy (which neglects SRO for the specific binary interactions). This makes possible the incorporation of several optimized binary systems coming from the CALPHAD community where a Bragg-Williams configurational entropy with a Redlich-Kister-Muggianu (R-K) polynomial expression of the excess Gibbs energy are typically used. For each binary system in the present database, the type of configurational entropy is given in the "List of optimized systems and calculated binary phase diagrams". In many cases, an important part of the experimental information on a given binary system is associated to solid-liquid equilibria. Hence a thermodynamic model for the liquid phase that takes into account SRO can permit a better fit of the thermodynamic and phase equilibrium data for the important solid phases that constitute the alloy.

Most solid solution phases have been described using sublattice models based on the Compound Energy Formalism (long-range order (LRO)) which include interstitials and vacancies where appropriate. The FCC\_A1 and BCC\_B2 solid solutions have C as an interstitial species. Please note that HCP\_A3 does not have C as an interstitial species.

**Table 4: Optimized binary systems of the FTlite 8.3 Database (quality → Figure 1)**

System	Liquid Model	Comments	Source
Ag - Al	Bragg-Williams R-K Polynomial		S.S. Lim, P.L. Rossiter, J.W. Tibbals, Calphad 19 (1995) 131-142.
Ag - As	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
Ag - Au	MQMPA (Pair Fraction Exp.)	BW liquid converted to MQMPA, no ordered phases	M. Hassam, J. Agren, M. Gaune-Escard, J.P. Bros, Metall. Trans. 21A (1990) 1877-1884.
Ag - B	Bragg-Williams R-K Polynomial		unpublished assessment of Korb (2004) supplied by GTT to SGTE in 2005
Ag - Ba	Bragg-Williams R-K Polynomial		SGTE
Ag - Be	Bragg-Williams R-K Polynomial		SGTE
Ag - Bi	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Ag - C	MQMPA (Pair Fraction Exp.)	graphite saturation	P. Chartrand, CRCT, 2014
Ag - Ca	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", 2014
Ag - Co	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Ag - Cr	Bragg-Williams R-K Polynomial		SGTE
Ag - Cu	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Ag - Fe	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Ag - Ga	Bragg-Williams R-K Polynomial	Stoichiometric Ag <sub>2</sub> Ga	W. Gierlotka, D. Jendrzejczyk-Handzlik / Journal of Alloys and Compounds 509 (2011) 38-42
Ag - Ge	Bragg-Williams R-K Polynomial		P.Y. Chevalier, Thermochimica Acta 130 (1988) 25-32.
Ag - H	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FactPS Database Valid for Ag-rich region</b>	P. Chartrand, CRCT, 2014
Ag - In	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG and M. MEDRAJ, "Thermodynamic and Experimental Study of the Mg-Sn-Ag-In Quaternary System", <i>Journal of Phase Equilibria and Diffusion</i> , Volume 35, Issue 3, (June 2014), pp 284-31.
Ag - K	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
Ag - Li	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
Ag - Mg	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG and M. MEDRAJ, "Thermodynamic and Experimental Study of the Mg-Sn-Ag-In Quaternary System", <i>Journal of Phase Equilibria and Diffusion</i> , Volume 35, Issue 3, (June 2014), pp 284-31.
Ag - Mn	Bragg-Williams R-K Polynomial		I. Karakaya, W. T. Thompson, Bull. Alloy Phase Diagrams, 1990, 11, (5), 80-486
Ag - Mo	MQMPA (Pair Fraction Exp.)		Refitted from unpublished assessment of J.Korb
Ag - N	MQMPA (Pair Fraction Exp.)	Estimated from Cu - N	P. Chartrand, CRCT, 2022
Ag - Na	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", <i>Journal of Chemical Thermodynamics</i> , 66 (2013) 22-33
Ag - Nb	Bragg-Williams R-K Polynomial		SGTE
Ag - Ni	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Ag - O	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FACTPS Database (or SGPS) Valid for Ag-rich region</b>	P. Chartrand, CRCT, 2021

<b>Ag - P</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Ag - Pb</b>	Bragg-Williams R-K Polynomial		Leo Lukas (Unpublished work, 2000) based on original work of Zimmermann (B. Zimmerman, Thesis, University of Stuttgart 1976 "Optimisation by experimental and calculation of the binary and ternary systems of Ag, Bi, Pb and Ti").
<b>Ag - Pt</b>	Bragg-Williams R-K Polynomial		SGnobl
<b>Ag - Sb</b>	Bragg-Williams R-K Polynomial	Stoichiometric Ag <sub>3</sub> Sb	E. Zoro, C. Servant, B. Legendre, Journal of Phase equilibria and Diffusion, 2007, 28, 250-257
<b>Ag - Sc</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Ag - Si</b>	Bragg-Williams R-K Polynomial		P.Y. Chevalier, Thermochimica Acta 113 (1988) 33-41.
<b>Ag - Sn</b>	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG and M. MEDRAJ, "Thermodynamic and Experimental Study of the Mg-Sn-Ag-In Quaternary System", <i>Journal of Phase Equilibria and Diffusion</i> , Volume 35, Issue 3, (June 2014), pp 284-31
<b>Ag - Sr</b>	MQMPA (Pair Fraction Exp.)		Jian Wang, Ph.D. Thesis, Ecole Polytechnique, Montreal, 2014
<b>Ag - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Ag - Ti</b>	Bragg-Williams R-K Polynomial		Mei Li, Changrong Li, Fuming Wang, Weijing Zhang, CALPHAD, 2005, 29, 269-275
<b>Ag - V</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Ag - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Ag - Y</b>	MQMPA (Pair Fraction Exp.)		Jian Wang et al., Intermetallics 136 (2021) 107260
<b>Ag - Zn</b>	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
<b>Ag - Zr</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang & I.-H. Jung, Intermetallics 2010
<b>Al - As</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006 (VLAB Project)
<b>Al - Au</b>	Bragg-Williams R-K Polynomial		J.Murray, H.Okamoto, T.B.Massalski, Bull.Alloy Phase Diags.8 (1987) 20-30 (Modified by A.T.Dinsdale to give compatibility with SGTE unary data and to prevent high-temp. stability of fcc)
<b>Al - B</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006 (VLAB Project)
<b>Al - Ba</b>	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, CRCT, 2004 (VLAB Project)
<b>Al - Be</b>	Bragg-Williams R-K Polynomial		M. Piché, M.A.Sc., École Polytechnique, 2002
<b>Al - Bi</b>	MQMPA (Pair Fraction Exp.)		M. Paliwal and In-Ho Jung, "Thermodynamic modeling of Al-Bi, Al-Sb, Mg-Al-Bi and Mg-Al-Sb systems"Calphad, 2010, vol. 34, pp. 51-63.
<b>Al - C</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2004
<b>Al - Ca</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
<b>Al - Ce</b>	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-La, Al-Ce, Al-Pr, Al-Nd and Al-Sm systems using the Modified Quasicontinuous Model for liquids", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 35(1), p.30-41, 2011.
<b>Al - Co</b>	Bragg-Williams R-K Polynomial	Stoichiometric AlCo-B2	N. Dupin, I. Ansara, La Revue de Metallurgie-CIT/Sceince et Genie des Materiaux, September 1998, 1121-1129, "Thermodynamic evaluation of the system Al-Co"
<b>Al - Cr</b>	MQMPA (Pair Fraction Exp.)		S.Cui and I.-H.Jung, J. Alloys and Compounds, no.708, 2017 pp.887-902
<b>Al - Cu</b>	Bragg-Williams R-K Polynomial		COST 507, pp.28-33; Vol.Data: C. Robelin, CRCT, 2006 (VLAB Project)
<b>Al - Dy</b>	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-Gd, Al-Tb, Al-Dy, Al-Ho and Al-Er systems using a Modified Quasicontinuous Model for the liquid", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 34(4), p.456-466, 2010

Al - Er	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-Gd, Al-Tb, Al-Dy, Al-Ho and Al-Er systems using a Modified Quasichemical Model for the liquid", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 34(4), p.456-466, 2010
Al - Eu	MQMPA (Pair Fraction Exp.)		Liling Jin, Ph.D. Thesis, Ecole Polytechnique (2012)
Al - Fe	MQMPA (Pair Fraction Exp.)	Missing Fe <sub>3</sub> Al D0 <sub>3</sub> (It)	A. T. Phan & Y-B Kang, Acta Materialia 2014
Al - Ga	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2016 modified from A.Watson CALPHAD 1992
Al - Gd	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-Gd, Al-Tb, Al-Dy, Al-Ho and Al-Er systems using a Modified Quasichemical Model for the liquid", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 34(4), p.456-466, 2010
Al - Ge	Bragg-Williams R-K Polynomial		I. Ansara, J. P. Bros, M. Gambino, CALPHAD 1979, 3(3), 225
Al - H	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Al – AlH <sub>3</sub>	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
Al - Hf	Bragg-Williams R-K Polynomial		T.Wang,Z.Jin & J.-C.Zhao, JPE, vol.23, no.5, p.416, 2002
Al - Ho	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-Gd, Al-Tb, Al-Dy, Al-Ho and Al-Er systems using a Modified Quasichemical Model for the liquid", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 34(4), p.456-466, 2010
Al - In	Bragg-Williams R-K Polynomial		I. Ansara, C Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson, CALPHAD 1994, 18(4), 177-222
Al - K	MQMPA (Pair Fraction Exp.)	Estimated from Al-Na	P. Chartrand, CRCT, 2000
Al - La	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-La, Al-Ce, Al-Pr, Al-Nd and Al-Sm systems using the Modified Quasichemical Model for liquids", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 35(1), p.30-41, 2011.
Al - Li	MQMPA (Pair Fraction Exp.)		J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
Al - Lu	MQMPA (Pair Fraction Exp.)		Liling Jin, Ph.D. Thesis, Ecole Polytechnique (2012)
Al - Mg	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2006; Vol.Data: F.Gemme, CRCT, 2003 (VLAB project); C.Aliravci, 2007 (GM Project).
Al - Mn	MQMPA (Pair Fraction Exp.)		M.-S. Kim and Y-B Kang, JPED 2015
Al - Mo	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
Al - N	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Al - AlN	SGTE
Al - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2000 (FACT Consortium Project)
Al - Nb	Bragg-Williams R-K Polynomial		C Servant and I. Ansara, J. Chim. Phys. 1997, 94, 869-888
Al - Nd	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-La, Al-Ce, Al-Pr, Al-Nd and Al-Sm systems using the Modified Quasichemical Model for liquids", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 35(1), p.30-41, 2011.
Al - Ni	Bragg-Williams R-K Polynomial		COST-507 K. Hack, GTT-Technologies, private communication, 2003

<b>Al - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Al - AlO	P. Chartrand, CRCT, 2021
<b>Al - P</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Al - Pb</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2009
<b>Al - Pr</b>	MQMPA (Pair Fraction Exp.)		L. JIN, Y.-B. KANG, P. CHARTRAND and C.D. FUERST, "Thermodynamic evaluation and optimization of Al-La, Al-Ce, Al-Pr, Al-Nd and Al-Sm systems using the Modified Quasichemical Model for liquids", CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry, 35(1), p.30-41, 2011.
<b>Al - Pt</b>	Bragg-Williams R-K Polynomial		K.Wu and Z.Jin, Journal of Phase Equilibria, 21(3), 2000, 221-226
<b>Al - Sb</b>	MQMPA (Pair Fraction Exp.)		M. Palwal and In-Ho Jung, "Thermodynamic modeling of Al-Bi, Al-Sb, Mg-Al-Bi and Mg-Al-Sb systems" Calphad, 2010, vol. 34, pp. 51-63.
<b>Al - Sc</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, A.D. Pelton, P. Chartrand and C. Fuerst, Calphad 32 (2008), pp. 413-422
<b>Al - Si</b>	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, M.A.Sc. thesis, Ecole Polytechnique, 2006; Vol. Data: F. Gemme, CRCT, 2003 (VLAB project)
<b>Al - Sm</b>	MQMPA (Pair Fraction Exp.)		Liling Jin, Ph.D. Thesis, Ecole Polytechnique (2012)
<b>Al - Sn</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang and A.D. Pelton, CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 34 (2010) 180 188
<b>Al - Sr</b>	MQMPA (Pair Fraction Exp.)		P.-A. Anctil, Projet de fin d'études, CRCT, 2003
<b>Al - Ta</b>	Bragg-Williams R-K Polynomial		V.T. Witusiewicz et al. / Intermetallics 18 (2010) 92–106
<b>Al - Tb</b>	MQMPA (Pair Fraction Exp.)		Liling Jin, Ph.D. Thesis, Ecole Polytechnique (2012)
<b>Al - Ti</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
<b>Al - Tm</b>	MQMPA (Pair Fraction Exp.)		Liling Jin, Ph.D. Thesis, Ecole Polytechnique (2012)
<b>Al - V</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
<b>Al - W</b>	Bragg-Williams R-K Polynomial		COST 507, pp.103-108.
<b>Al - Y</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, A.D. Pelton, P. Chartrand and C. Fuerst, Calphad 32 (2008), pp. 413-422
<b>Al - Yb</b>	MQMPA (Pair Fraction Exp.)		Liling Jin, Ph.D. Thesis, Ecole Polytechnique (2012)
<b>Al - Zn</b>	MQMPA (Pair Fraction Exp.)		Liquid: P. Chartrand, CRCT, 2006 (VLAB Project); Vol. Data: F. Gemme, CRCT, 2003 (VLAB Project); solids S. –L. Chen and Y.A. Chang, Calphad, 17 (1993), pp. 113-124.
<b>Al - Zr</b>	Bragg-Williams R-K Polynomial		COST 507, pp.112-116.
<b>As - Au</b>	Bragg-Williams R-K Polynomial		P.J.Spencer, June 1998
<b>As - Bi</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>As - Cu</b>	Bragg-Williams R-K Polynomial	Stoichiometric Cu <sub>3</sub> As	M.Hamalainen, private communication
<b>As - Fe</b>	Bragg-Williams R-K Polynomial	Missing stoichiometric compounds	P. J. Spencer, 2008
<b>As - Ga</b>	Bragg-Williams R-K Polynomial		C. Chatillon, I. Ansara, A. Watson and B. B. Argent: CALPHAD, 1990, 14(2), 203-14.
<b>As - Ge</b>	Bragg-Williams R-K Polynomial		I. Ansara and D. Dutarte: CALPHAD, 1984, 8(4), 323-342
<b>As - In</b>	Bragg-Williams R-K Polynomial		C. Chatillon, I. Ansara, A. Watson and B. B. Argent: CALPHAD, 1990, 14(2), 203-14.
<b>As - Ni</b>	Bragg-Williams R-K Polynomial		S. Uhland et al., CALPHAD 25[1] (2001) 109-124
<b>As - P</b>	Bragg-Williams R-K Polynomial		I Ansara, C Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson: CALPHAD 1994, 18(4), 177-222
<b>As - Pb</b>	Bragg-Williams R-K Polynomial		M.Hamalainen, private communication
<b>As - Pt</b>	Bragg-Williams R-K Polynomial		M. Li et al. / Journal of Alloys and Compounds 437 (2007) 71–79
<b>As - Sb</b>	Bragg-Williams R-K Polynomial		I Ansara, C Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G.

			Chart, T. Anderson: CALPHAD 1994, 18(4), 177-222
<b>As - Si</b>	MQMPA (Pair Fraction Exp.)		P.-J. Spencer, 2006
<b>As - Sn</b>	Bragg-Williams R-K Polynomial		L.Wu et al., CALPHAD (74) 2021, 102296
<b>As - Zn</b>	MQMPA (Pair Fraction Exp.)		O. Kidari and P. Chartrand, CRCT, 2023
<b>Au - B</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Au - Bi</b>	Bragg-Williams R-K Polynomial		C. Servant, E. Zoro and B. Legendre, Calphad, 30 (2006) 443
<b>Au - C</b>	Bragg-Williams R-K Polynomial		P.J. Spencer (2007)
<b>Au - Co</b>	Bragg-Williams R-K Polynomial		H.Q.Dong et al., J. Electronic Mater., 38(10) 2009, p.2158
<b>Au - Cr</b>	Bragg-Williams R-K Polynomial		P.J. Spencer, (1998)
<b>Au - Cu</b>	Bragg-Williams R-K Polynomial	No ordered L <sub>1</sub> <sub>2</sub> & L <sub>1</sub> <sub>0</sub> phases	B.Sundman, S.G.Fries, A.Oates, Calphad 22 (1998) 335-354
<b>Au - Fe</b>	Bragg-Williams R-K Polynomial		Y.Liu et al., CALPHAD 476 (2009) 79-83
<b>Au - Ga</b>	Bragg-Williams R-K Polynomial		J. Wang, Y. Liu, L. Liu, H. Zhou, Z. Jin, Calphad 35(2) (2011) 242-248
<b>Au - Ge</b>	Bragg-Williams R-K Polynomial		P. Y. Chevalier: Thermochimica Acta, 1989, 141, 217-22
<b>Au - In</b>	Bragg-Williams R-K Polynomial		I.Ansara, J.P.Nabot, Calphad 16 (1992) 13-18
<b>Au - Ni</b>	Bragg-Williams R-K Polynomial		J. Wang et al., CALPHAD (2005) pp.263-268
<b>Au - Pb</b>	Bragg-Williams R-K Polynomial		J.P.Nabot, Thesis, Grenoble,1986
<b>Au - Pt</b>	Bragg-Williams R-K Polynomial		SGnobl
<b>Au - Sb</b>	Bragg-Williams R-K Polynomial		E.Zoro, C.Servant and B.Legendre, J.Thermal Anal. & Calo., 90[2] (2007) pp.347-353
<b>Au - Si</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Au - Sn</b>	Bragg-Williams R-K Polynomial		P.Y.Chevalier, Thermochimica Acta 130 (1988) 1-13
<b>Au - Ti</b>	Bragg-Williams R-K Polynomial		W.Luo et al., CALPHAD 25(1) 2001 19-26
<b>Au - Zn</b>	Bragg-Williams R-K Polynomial	Intermetallic solutions missing – liquid and FCC-A1 only	H.S.Liu et al., Intermetallics 11 (2003) 987–994
<b>B - Ba</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
<b>B - C</b>	MQMPA (Pair Fraction Exp.)		P.-J. Spencer, 2007
<b>B - Ca</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
<b>B - Co</b>	Bragg-Williams R-K Polynomial		SGTE
<b>B - Cr</b>	Bragg-Williams R-K Polynomial		L.M. Pan, unpublished research in SGTE Solution Database, 1991.
<b>B - Cu</b>	Bragg-Williams R-K Polynomial		SGTE
<b>B - Fe</b>	Bragg-Williams R-K Polynomial		B. Hallemans, P.Wollants, J.Roos, J. Phase Equilib. 16 (1995), pp. 137-149.
<b>B - Ga</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>B - Ge</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>B - Hf</b>	Bragg-Williams R-K Polynomial		COST 507, pp.120-122.
<b>B - K</b>	MQMPA (Pair Fraction Exp.)	Estimated	P. Chartrand, CRCT, 2018
<b>B - Li</b>	MQMPA (Pair Fraction Exp.)	Estimated	P. Chartrand, CRCT, 2014
<b>B - Mg</b>	MQMPA (Pair Fraction Exp.)		X.-F. Sheng, CRCT, 2008 (GM Project)
<b>B - Mn</b>	Bragg-Williams R-K Polynomial		B. Sundman et al., JPE, 2010
<b>B - Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
<b>B - N</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for B-BN	SGTE
<b>B - Na</b>	MQMPA (Pair Fraction Exp.)	Estimated	P. Chartrand, CRCT, 2018

<b>B - Nb</b>	Bragg-Williams R-K Polynomial		V.T. Witusiewicz et al. / Journal of Alloys and Compounds 448 (2008) 185–194
<b>B - Nd</b>	Bragg-Williams R-K Polynomial		COST-507
<b>B - Ni</b>	Bragg-Williams R-K Polynomial		L.M. Pan, unpublished research in SGTE Solution Database, 1991.
<b>B - Si</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2007
<b>B - Sn</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>B - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
<b>B - Ta</b>	Bragg-Williams R-K Polynomial		A.A.A.Pinto da Silva, et al., CALPHAD 63 (2018) 107-115
<b>B - Ti</b>	MQMPA (Pair Fraction Exp.)		P.-J. Spencer, CRCT, 2005
<b>B - V</b>	Bragg-Williams R-K Polynomial		L.M. Pan, unpublished research in SGTE Solution Database, 1991.
<b>B - W</b>	Bragg-Williams R-K Polynomial		H. Duschanek, P. Rogl, J. Phase Equilib. 16 (1995), pp. 150-161.
<b>B - Zn</b>	Bragg-Williams R-K Polynomial		P. Chartrand 2014
<b>B - Zr</b>	MQMPA (Pair Fraction Exp.)		X.-F. Sheng, CRCT, 2008 (GM Project)
<b>Ba - Be</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Ca</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2004
<b>Ba - Ce</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Co</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Ba - Cr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Cu</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Ba - Dy</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Er</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Eu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Fe</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Ga</b>	Bragg-Williams R-K Polynomial		X.Li et al., CALPHAD 43 (2013) 52-60
<b>Ba - Gd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - H</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Ba – BaH <sub>2</sub>	P. Chartrand, 2016
<b>Ba - Hf</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Ba - Ho</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2005 (VLAB project)
<b>Ba - La</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2003 (VLAB project)
<b>Ba - Lu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Mg</b>	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, Projet de fin d'études, CRCT, 2004 (VLAB project)
<b>Ba - Mn</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2021
<b>Ba - Na</b>	ideal		Est.
<b>Ba - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Ba - Nd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Ni</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Ba - Pr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ba - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014

Ba - Si	MQMPA (Pair Fraction Exp.)		J.P. Harvey 2005
Ba - Sm	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ba - Sr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ba - Ta	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Ba - Tb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ba - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ba - Tm	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ba - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Ba - W	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
Ba - Y	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ba - Yb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ba - Zn	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014 modified from J.P. Spencer 2006
Ba - Zr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Ca	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Cu	Bragg-Williams R-K Polynomial		P.J. Spencer, 2003
Be - Fe	Bragg-Williams R-K Polynomial	Stoichiometric BeFe	M. Piché, M.A.Sc., École Polytechnique, 2002
Be - K	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Li	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Mg	Bragg-Williams R-K Polynomial		M. Piché, M.A.Sc., École Polytechnique, 2002
Be - Mn	Bragg-Williams R-K Polynomial		M. Piché, M.A.Sc., École Polytechnique, 2002
Be - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Si	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Sn	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Sr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Be - Zn	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Bi - Co	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Bi - Cr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Bi - Cu	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
Bi - Fe	Bragg-Williams R-K Polynomial		from FSstel
Bi - Ga	MQMPA (Pair Fraction Exp.)		B.Kumar et al., CALPHAD 74 (2021)
Bi - Ge	Bragg-Williams R-K Polynomial		P. Y. Chevalier: Thermochimica Acta, 1988, 132, 111-116
Bi - In	Bragg-Williams R-K Polynomial		D.Boa, I.Anbara, Thermochimica Acta 314 (1998) 79-86
Bi - K	MQMPA (Pair Fraction Exp.)		Z.M. CAO, W. XIE, P. CHARTRAND, S.H. WEI, G.W. DU and Z.-Y. QIAO, "Thermodynamic assessment of the Bi-alkali metal (Li, Na, K, Rb) systems using the modified quasichemical model for the liquid phase", CALPHAD, 46 (2014), pp.159-167
Bi - Li	MQMPA (Pair Fraction Exp.)		Z.M. CAO, W. XIE, P. CHARTRAND, S.H. WEI, G.W. DU and Z.-Y. QIAO, "Thermodynamic assessment of the Bi-alkali metal (Li, Na, K, Rb) systems using the modified quasichemical model for the liquid phase", CALPHAD, 46 (2014), pp.159-167
Bi - Mg	MQMPA (Pair Fraction Exp.)		M. Paliwal and In-Ho Jung, "Thermodynamic modeling of Mg-Bi and Mg-Sb systems and short-range-ordering behavior of the liquid solutions"Calphad, 2009, vol. 33, pp. 744-754.
Bi - Mn	Bragg-Williams R-K Polynomial		K. Oikawa, Y. Mitsui, K. Koyama and K. Anzai, Materials Transactions, Vol. 52, No. 11 (2011) pp. 2032 to 2039
Bi - Na	MQMPA (Pair Fraction Exp.)		Z.M. CAO, W. XIE, P. CHARTRAND, S.H. WEI, G.W. DU and Z.-Y. QIAO, "Thermodynamic assessment of the Bi-alkali metal (Li, Na, K, Rb) systems using the modified quasichemical model for the liquid phase", CALPHAD, 46 (2014), pp.159-167

<b>Bi - Ni</b>	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
<b>Bi - Pb</b>	Bragg-Williams R-K Polynomial		D.Boa, I.Ansara, Thermochimica Acta 314 (1998) 79-86
<b>Bi - Sb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand 2018 modified from P.J. Spencer, 2006
<b>Bi - Si</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
<b>Bi - Sn</b>	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
<b>Bi - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Bi - Zn</b>	Bragg-Williams R-K Polynomial		C.Girard, Thesis, Marseille, 1985
<b>C - Ca</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>C - Co</b>	Bragg-Williams R-K Polynomial		A. Fernandez Guillermet: Z. Metallkde., 1987, 78, 700-9
<b>C - Cr</b>	Bragg-Williams R-K Polynomial		B. J. Lee: CALPHAD 1992, 16(2), 121-149
<b>C - Cu</b>	MQMPA (Pair Fraction Exp.)		S. Shubhank and Y.-B.Kang CALPHAD 45 (2014) 127–137
<b>C - Fe</b>	MQMPA (Pair Fraction Exp.)	Incl. metastable carbides	M-S Kim, Y-B Kang JPE 2015
<b>C - Ge</b>	MQMPA (Pair Fraction Exp.)		P.Chartrand, 2014
<b>C - Hf</b>	Bragg-Williams R-K Polynomial		Bittermann and Rogl, JPE, 18(4), p 344, 1997
<b>C - Mg</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>C - Mn</b>	MQMPA (Pair Fraction Exp.)		M.K. Paek, Y-B Kang, CALPHAD 46 (2014), 92-102
<b>C - Mo</b>	Bragg-Williams R-K Polynomial		SGTE
<b>C - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2013
<b>C - Nb</b>	Bragg-Williams R-K Polynomial		WM Huang: Mater. Sci. and Techn. 1990, 6(8), 687-694
<b>C - Ni</b>	Bragg-Williams R-K Polynomial		B. J. Lee: CALPHAD, 1992, 16(2), 121-149
<b>C - P</b>	Bragg-Williams R-K Polynomial		P. Gustafson: Inst. Met. Res. (Report IM-2549, 1990))
<b>C - Pb</b>	Bragg-Williams R-K Polynomial		unpublished assessment of T. G. Chart, NPL 1987
<b>C - Pt</b>	Bragg-Williams R-K Polynomial		GTT, 2005
<b>C - Sb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>C - Si</b>	MQMPA (Pair Fraction Exp.)		M-K Paek, Y-B Kang CALPHAD 46 (2014) 92–102
<b>C - Sn</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2007
<b>C - Ta</b>	Bragg-Williams R-K Polynomial		P. Chartrand, 2003
<b>C - Ti</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2007 (VLAB Project)
<b>C - V</b>	Bragg-Williams R-K Polynomial		WM Huang : Z. Metallkde, 1991, 82, (3), 174-181
<b>C - W</b>	Bragg-Williams R-K Polynomial		P. Gustafson: Report TRITA 0212 (1985), Mat. Sci and Tech. 1986, 2(7), 653-658
<b>C - Zr</b>	Bragg-Williams R-K Polynomial		A. Fernandez Guillermet: J. Alloys Compounds, 1995, 217, 69-89
<b>Ca - Ce</b>	MQMPA (Pair Fraction Exp.)		Rui Gao, Thesis, McGill Univ., 2019
<b>Ca - Co</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Ca - Cr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Ca - Cu</b>	MQMPA (Pair Fraction Exp.)		Jian Wang et al., Calphad 75 (2021) 102325
<b>Ca - Dy</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Er</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Eu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Fe</b>	MQMPA (Pair Fraction Exp.)		S.Cui, M.Paliwal and I.-H. Jung, MetTrans 2014
<b>Ca - Ga</b>	Bragg-Williams R-K Polynomial		X.Li et al., CALPHAD 43 (2013) 52-60
<b>Ca - Gd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Ge</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023

<b>Ca - H</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Ca – CaH <sub>2</sub>	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
<b>Ca - Hf</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Ca - Ho</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - In</b>	MQMPA (Pair Fraction Exp.) + BW		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
<b>Ca - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2003 (VLAB project)
<b>Ca - La</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Li</b>	MQMPA (Pair Fraction Exp.)		J. Wang, I.-H. Jung and P. Chartrand, "Thermodynamic modeling of then Ag-(Ca, Li, Zn) and Ca-(In, Li) binary systems", submitted 2014
<b>Ca - Lu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Mg</b>	MQMPA (Pair Fraction Exp.)		J. Wang, Ph.D. Thesis, Ecole Polytechnique, 2014
<b>Ca - Mn</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006 (GM project)
<b>Ca - Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Na</b>	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", <i>Journal of Chemical Thermodynamics</i> , 66 (2013) 22-33
<b>Ca - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Ca - Nd</b>	MQMPA (Pair Fraction Exp.)		Rui Gao, 2019 (McGill Thesis)
<b>Ca - Ni</b>	Bragg-Williams R-K Polynomial		M. Medraj, 2006
<b>Ca - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Ca - CaO	P. Chartrand, 2000
<b>Ca - Pb</b>	Bragg-Williams R-K Polynomial		V.P.Itkin and C.B.Alcock, <i>J. Phase Equilib.</i> 1992, pp.162-169
<b>Ca - Pr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Si</b>	MQMPA (Pair Fraction Exp.)		M. HEYRMAN and P. CHARTRAND, "Thermodynamic Evaluation and Optimization of the Ca-Si System", <i>J. Phase Equilibria and Diffusion</i> , 27 [3], 220-230, 2006
<b>Ca - Sm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Sn</b>	MQMPA (Pair Fraction Exp.)		Jian Wang, Ph.D. Thesis, Ecole Polytechnique, Montreal, 2014
<b>Ca - Sr</b>	MQMPA (Pair Fraction Exp.)		Jian Wang, Ph.D. Thesis, Ecole Polytechnique, Montreal, 2014
<b>Ca - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Ca - Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Ca - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Ca - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Ca - Y</b>	MQMPA (Pair Fraction Exp.)		Rui Gao 2019 Thesis McGill
<b>Ca - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ca - Zn</b>	MQMPA (Pair Fraction Exp.)		P. Spencer, A.D. Pelton., Y.-B. Kang, P. Chartrand, and C. Fuerst, <i>Calphad</i> 32 (2007), pp. 423-431
<b>Ca - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ce - Co</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Ce - Cr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014

<b>Ce - Cu</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Ce - Eu</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Ce - Fe</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Ce - Gd</b>	MQMPA (Pair Fraction Exp.)	estimated from Ce-Y	Y.-B. Kang, CRCT, 2008
<b>Ce - Ho</b>	MQMPA (Pair Fraction Exp.)	Ideal	Y.-B. Kang, CRCT, 2008
<b>Ce - In</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
<b>Ce - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ce - La</b>	MQMPA (Pair Fraction Exp.)	ideal	Y.-B. Kang, CRCT, 2008
<b>Ce - Li</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2005
<b>Ce - Mg</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, L. JIN, P. CHARTRAND, A. E. GHERIBI, K. BAI and P. WU, "Thermodynamic evaluations and optimizations of binary Mg-light Rare-Earth (La, Ce, Pr,Nd, Sm) Systems", CALPHAD, 38, 100-116, 2012
<b>Ce - Mn</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, A.D. Pelton., P. Chartrand, P. Spencer and C. Fuerst, J. Phase Equil. Diff. 28 (2007), pp. 342-354
<b>Ce - Mo</b>	Bragg-Williams R-K Polynomial		W.Chan, M.C. Gao, O.N. Dogan, P. King, JPEDAV (31) 2010
<b>Ce - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ce - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ce - Pr</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Ce - Sc</b>	MQMPA (Pair Fraction Exp.)	ideal	Y.-B. Kang, CRCT, 2008
<b>Ce - Si</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Ce - Sm</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Ce - Sn</b>	MQMPA (Pair Fraction Exp.)		Kim, Thesis, McGill Univ., 2016
<b>Ce - Sr</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
<b>Ce - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ce - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ce - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ce - Y</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, A.D. Pelton., P. Chartrand, P. Spencer and C. Fuerst, J. Phase Equil. Diff. 28 (2007), pp. 342-354
<b>Ce - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Ce - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Co - Cr</b>	MQMPA (Pair Fraction Exp.)		A.Kusoffsky, B.Jansson, Calphad 21 (1997) 321-333; liquid converted to MQMPA [2022Cha]
<b>Co - Cu</b>	Bragg-Williams R-K Polynomial		P.J.Spencer - L.Kaufman interaction parameters with SGTE element data
<b>Co - Dy</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co - Er</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co - Fe</b>	MQMPA (Pair Fraction Exp.)		A.F.Guillermet, High Temp. High Press. 19 (1988) 477-499; liquid converted to MQMPA [2021Cha]
<b>Co - Ga</b>	Bragg-Williams R-K Polynomial		A. Chari et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 34 (2010) 189–195
<b>Co - Gd</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co - Hf</b>	Bragg-Williams R-K Polynomial		Lu, Cheng, ..., Du J.Alloys & Compounds 2015
<b>Co - Ho</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co - In</b>	Bragg-Williams R-K Polynomial		D. Boa, B. K. Dongui, I. Ansara: CALPHAD 25 (2001) 645-650
<b>Co - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Co - La</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Co - Mg</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018

<b>Co – Mn</b>	Bragg-Williams R-K Polynomial		W.Huang, Calphad 13 (1989) 231-242
<b>Co - Mo</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Co – N</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FACTPS Database Valid for Co – Co <sub>2</sub> N	SGTE
<b>Co – Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Co – Nb</b>	Bragg-Williams R-K Polynomial		K.C.H.Kumar, I.Ansara, P.Wollants, L.Delaey, J.Alloys and Cpd. 267 (1998) 105-112
<b>Co - Nd</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co – Ni</b>	Bragg-Williams R-K Polynomial		A.F.Guillermet, Z.Metallkde. 78 (1987) 639-647; Z.Metallkde. 79 (1988) 524-536
<b>Co - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Co - CoO	P. Chartrand, CRCT, 2021
<b>Co – Pb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Co - Pr</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co - Pt</b>	Bragg-Williams R-K Polynomial		D.E. Kim et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 35 (2011) 323–330
<b>Co – Sb</b>	Bragg-Williams R-K Polynomial		Y.Zhang et al., CALPHAD 33 (2009) 405–414
<b>Co – Sc</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co – Si</b>	Bragg-Williams R-K Polynomial	Stoichiometric CoSi Inverted L-L miscibility gap above 3000°C	S.D.Choi, Calphad 16 (1992) 151-159
<b>Co - Sm</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co – Sn</b>	Bragg-Williams R-K Polynomial		L.Liu et al., J.Electron.Mater.33(9) (2004) 935-939
<b>Co – Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Co – Ta</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Co – Tb</b>	MQMPA (Pair Fraction Exp.)		L. Jing, CRCT, 2021
<b>Co – Ti</b>	Bragg-Williams R-K Polynomial		G.Cacciamani, R.Ferro, I.Ansara, N.Dupin, submitted to "Intermetallics", 1999
<b>Co – V</b>	Bragg-Williams R-K Polynomial	Stoichiometric Co <sub>3</sub> V	J. Bratberg, B. Sundman: J. Phase Equil., (2003), 24(6), 495-503
<b>Co – W</b>	Bragg-Williams R-K Polynomial		Markstrom, Sundman & Frisk JPE (2005)
<b>Co – Y</b>	MQMPA (Pair Fraction Exp.)		L. Jin, P. Chartrand, CRCT, 2020
<b>Co – Zn</b>	Bragg-Williams R-K Polynomial	Missing CoZn <sub>7</sub> (ht) and CoZn(ht)	G.P. Vassilev, M. Jiang: J. Phase Equil. and Diffusion 2004, 25, 259-268
<b>Co – Zr</b>	Bragg-Williams R-K Polynomial		Durga & Kumar, CALPHAD 2010
<b>Cr - Cu</b>	MQMPA (Pair Fraction Exp.)		S. Cui and I-H Jung, CALPHAD 2017
<b>Cr - Dy</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Er</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Eu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Fe</b>	MQMPA (Pair Fraction Exp.)		S. Cui and I-H Jung, CALPHAD 2017
<b>Cr - Ga</b>	Bragg-Williams R-K Polynomial		P.Chartrand, CRCT, 2022 (modified from A. Belgacem-Bouzida et al. / Journal of Alloys and Compounds 397 (2005) 155–160)
<b>Cr - Gd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - H</b>	MQMPA (Pair Fraction Exp.)	estimated from Ti-H; Gaseous species must be taken from FactPS Database Valid for Cr rich region	P. Chartrand 2010
<b>Cr - Hf</b>	Bragg-Williams R-K Polynomial		J.Pavlu et al. CALPHAD 2010
<b>Cr - Ho</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014

<b>Cr - K</b>	MQMPA (Pair Fraction Exp.)	estimated from Fe-K	P. Chartrand 2003
<b>Cr - La</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Li</b>	MQMPA (Pair Fraction Exp.)	estimated from Fe-Li	P. Chartrand 2003
<b>Cr - Lu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Mg</b>	MQMPA (Pair Fraction Exp.)		S. Cui et al., Journal of Alloys and Compounds, 2017
<b>Cr - Mn</b>	Bragg-Williams R-K Polynomial		COST 507, pp.145-148.
<b>Cr - Mo</b>	Bragg-Williams R-K Polynomial		COST 507
<b>Cr - N</b>	Bragg-Williams R-K Polynomial	<b>Gaseous species must be taken from FactPS Database Valid for Cr-CrN Spurious formation of BCC-A2 above 2200°C (when gas NOT included)</b>	SGTE
<b>Cr - Na</b>	MQMPA (Pair Fraction Exp.)	estimated from Fe-Na	P. Chartrand, CRCT, 2003
<b>Cr - Nb</b>	Bragg-Williams R-K Polynomial		J.G.Costa Neto, S.G.Fries, H.-L.Lukas, Calphad 17 (1993) 219-228
<b>Cr - Nd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Ni</b>	Bragg-Williams R-K Polynomial		B.J. Lee, Calphad 16 1992, pp. 121-149.
<b>Cr - O</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FACTPS Database Valid for Cr - CrO</b>	P.Chartrand, CRCT, 2021
<b>Cr - P</b>	Bragg-Williams R-K Polynomial	<b>Valid in the Cr – Cr<sub>2</sub>P region only</b>	SGTE
<b>Cr - Pb</b>	Bragg-Williams R-K Polynomial		P. Chartrand, 2003
<b>Cr - Pr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Pt</b>	Bragg-Williams R-K Polynomial		Oikawa et al. Journal of Magnetism and Magnetic Materials 236 (2001) 220–233
<b>Cr - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Si</b>	MQMPA (Pair Fraction Exp.)		Senlin Cui & In-Ho Jung, Met.Trans.2017
<b>Cr - Sm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Sn</b>	Bragg-Williams R-K Polynomial		R. Jerlerud Perez, B. Sundman: CALPHAD, 25 (2001) 59-66
<b>Cr - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Ta</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cr - Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Ti</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
<b>Cr - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - V</b>	Bragg-Williams R-K Polynomial		B.J.Lee, Z.Metallkde 83 (1992) 292-299
<b>Cr - W</b>	Bragg-Williams R-K Polynomial		P. Gustafson, Calphad 12 (1988), pp. 277-292.
<b>Cr - Y</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Cr - Zn</b>	Bragg-Williams R-K Polynomial		I.Ansera, COST 507 (1998) ISBN 92-828-3902-8, p.158-160
<b>Cr - Zr</b>	Bragg-Williams R-K Polynomial		K.Zeng, M.Hamalainen, I.Ansera, COST 507 (1998) ISBN 92-828-3902-8 p 161-164
<b>Cu – Eu</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - Fe</b>	MQMPA (Pair Fraction Exp.)		Shubhank and Y-B. Kang, CALPHAD 2014
<b>Cu - Ga</b>	Bragg-Williams R-K Polynomial	<b>Missing Cu<sub>9</sub>Ga<sub>4</sub> γ<sub>1</sub>, γ<sub>2</sub>, γ<sub>3</sub> solid solutions</b>	Li et al., CALPHAD 32(2) (2008), 447-453
<b>Cu - Ge</b>	Bragg-Williams R-K Polynomial		Wang et al. J.Alloys Cmpds 2010

<b>Cu - H</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Cu - CuH	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
<b>Cu - Ho</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - In</b>	Bragg-Williams R-K Polynomial	$\gamma$ -CuIn, $\eta$ -CuIn are missing	C.R.Kao, A.Bolcavage et al, J Phase Equilibria 14 (1993) 22-30
<b>Cu - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Cu - La</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - Li</b>	Bragg-Williams R-K Polynomial		COST 507, pp.168-169.
<b>Cu - Mg</b>	MQMPA (Pair Fraction Exp.)		S. Cui and I-H. Jung, 2017
<b>Cu - Mn</b>	MQMPA (Pair Fraction Exp.)		S. Cui and I-H. Jung, 2017
<b>Cu - Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Cu - N</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
<b>Cu - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Cu - Nb</b>	Bragg-Williams R-K Polynomial		B.J. Lee database, private communication to SGTE, 1999.
<b>Cu - Nd</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - Ni</b>	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
<b>Cu - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Cu – Cu <sub>2</sub> O	P. Chartrand, CRCT, 2018
<b>Cu - P</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Cu - Pb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Cu - Pr</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - Pt</b>	Bragg-Williams R-K Polynomial		T.Abe, B.Sundman, H.Onodera, JPEDAV (2006) 27:5-13
<b>Cu - Sb</b>	Bragg-Williams R-K Polynomial		SGTE 2004
<b>Cu - Sc</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - Si</b>	MQMPA (Pair Fraction Exp.)		S. Cui and I-H. Jung, CALPHAD, 2017
<b>Cu - Sm</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - Sn</b>	MQMPA (Pair Fraction Exp.)		J. Wang, CRCT, 2015
<b>Cu - Sr</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Cu - Ti</b>	Bragg-Williams R-K Polynomial		K.C.H. Kumar, I. Ansara, P. Wollants, L. Delaey, Z.Metallkde. 87 (1996), pp. 666-672.
<b>Cu - V</b>	Bragg-Williams R-K Polynomial		B.J. Lee database, private communication to SGTE, 1999.
<b>Cu - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Cu - Y</b>	Bragg-Williams R-K Polynomial		COST 507, pp.182-185.
<b>Cu - Zn</b>	Bragg-Williams R-K Polynomial		Liang, Hsiao, Schmid-Fezter CALPHAD, 2015
<b>Cu - Zr</b>	MQMPA (Pair Fraction Exp.)		D.H.Kang I.H.Jung Intermetallics 2010
<b>Dy - Er</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008 (GM Project)
<b>Dy - Fe</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Dy - Gd</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
<b>Dy - Ho</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
<b>Dy - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Dy - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Dy - Mg</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang and L. Jin, CRCT, 2012
<b>Dy - Mn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012

Dy - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Dy - Nb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
Dy - Nd	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Dy - Pb	MQMPA (Pair Fraction Exp.)		M.Zydzik et al., CALPHAD 46 (2014) 1-17
Dy - Si	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
Dy - Sn	MQMPA (Pair Fraction Exp.)		Kim, Thesis, McGill Univ., 2016
Dy - Sr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Dy - Ta	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
Dy - Tb	MQMPA (Pair Fraction Exp.)	ideal	P. Chartrand, 2014
Dy - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Dy - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Dy - Y	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Dy - Zn	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
Dy - Zr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Er - Fe	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
Er - Gd	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Er - Ho	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Er - K	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Er - Li	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Er - Mg	MQMPA (Pair Fraction Exp.)		Y-B. Kang and L. Jin, CRCT, 2012
Er - Mn	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
Er - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Er - Nb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
Er - Nd	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Er - Sc	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
Er - Si	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
Er - Sn	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
Er - Sr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Er - Ta	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
Er - Tb	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
Er - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Er - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Er - Y	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008 (GM Project)
Er - Zn	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
Er - Zr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Eu - K	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Eu - Li	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Eu - Mg	MQMPA (Pair Fraction Exp.)		Y-B. Kang and L. Jin, CRCT, 2012 (GM Project)
Eu - Mn	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
Eu - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Eu - Nb	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
Eu - Si	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
Eu - Sr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014

<b>Eu - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Eu - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Eu - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Eu - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Eu - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Fe - Gd</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - H</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Fe-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
<b>Fe - Hf</b>	Bragg-Williams R-K Polynomial		M.Idbenali et al., <i>Journal of Alloys and Compounds</i> 456 (2008) 151–158
<b>Fe - Ho</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - In</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2022 modified from M.Ohno and K.Yoh, <i>Materials Transactions</i> , Vol. 50, No. 5 (2009) pp. 1202 to 1207
<b>Fe - K</b>	MQMPA (Pair Fraction Exp.)	from Fe in liquid K	P. Chartrand, 2003
<b>Fe - La</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - Li</b>	MQMPA (Pair Fraction Exp.)	from Fe in liquid Li	P. Chartrand, 2003
<b>Fe - Lu</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - Mg</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2006
<b>Fe - Mn</b>	MQMPA (Pair Fraction Exp.)		M-S Kim, Y-B Kang, JPE, 2015
<b>Fe - Mo</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Fe - N</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Fe-Fe <sub>2</sub> N	SGTE
<b>Fe - Na</b>	MQMPA (Pair Fraction Exp.)	from Fe in liquid Na	P. Chartrand, CRCT, 2003
<b>Fe - Nb</b>	Bragg-Williams R-K Polynomial		Khvan & Hallstedt CALPHAD 2013
<b>Fe - Nd</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - Ni</b>	Bragg-Williams R-K Polynomial		A.Dinsdale, T.Chart, NPL, unpublished work, 1986: I.Ansera - fcc ordering
<b>Fe - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Fe - FeO Missing non-stoichiometric wustite-FeO (considered stoichiometric)	P. Chartrand, CRCT, 2021
<b>Fe - P</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Fe - Pb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Fe - Pr</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - Pt</b>	Bragg-Williams R-K Polynomial		P.Fredriksson & B.Sundman, CALPHAD 25(4) 2001, 535-548
<b>Fe - Sb</b>	Bragg-Williams R-K Polynomial		C.Li et al., CALPHAD 47 (2014) 23-34
<b>Fe - Sc</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - Si</b>	MQMPA (Pair Fraction Exp.)		S. Cui and I-H Jung CALPHAD 2017
<b>Fe - Sm</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - Sn</b>	Bragg-Williams R-K Polynomial		K.C.H.Kumar, P.Wollants, L.Delaey, Calphad 20 (1996) 139-149 (with modifs from P. Chartrand (2018))
<b>Fe - Sr</b>	Bragg-Williams R-K Polynomial		P. Chartrand, CRCT, 2014
<b>Fe - Ta</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Fe - Tb</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014

<b>Fe - Ti</b>	Bragg-Williams R-K Polynomial		P.J. Spencer, 2001 (FACT Consortium)
<b>Fe - Tm</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - V</b>	Bragg-Williams R-K Polynomial		W.Huang, Z.Metallkde 82 (1991) 391-401
<b>Fe - W</b>	Bragg-Williams R-K Polynomial		J-O Andersson and P Gustafson: CALPHAD, 1983, 7(4), 317-326
<b>Fe - Y</b>	MQMPA (Pair Fraction Exp.)		B. Konar and I.-H. Jung, McGill, 2014
<b>Fe - Zn</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand (2018) MQMPA calibrated on the solids of Xiong, Du Liu CALPHAD 2009 (later modifications by M.Jacobs)
<b>Fe - Zr</b>	MQMPA (Pair Fraction Exp.)		J.J. Marin Bejarano, CRCT, 2009
<b>Ga - Ge</b>	Bragg-Williams R-K Polynomial		I Ansara, J P Bros, M Gambino: CAPHAD, 1979, 3, 225-233
<b>Ga - In</b>	Bragg-Williams R-K Polynomial		B. C. Rugg, T. G. Chart: CALPHAD, 1990, 14(2), 115-123
<b>Ga - Li</b>	Bragg-Williams R-K Polynomial		H.Azza et al., J. Phase Equilib. Diffus. (2017) 38:788–795
<b>Ga - Mg</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CALPHAD 2014
<b>Ga - Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
<b>Ga - N</b>	Bragg-Williams R-K Polynomial		J. Unland, B. Onderka, A. Davydov, R. Schmid-Fetzer, J. of Crystal Growth, 256, 33-51 (2003)
<b>Ga - Na</b>	Bragg-Williams R-K Polynomial		J. Wang et al. / Journal of Crystal Growth 307 (2007) 59-65
<b>Ga - Ni</b>	Bragg-Williams R-K Polynomial		Z.-M.Cao, X.Shi, W.Xie, I.Ohnuma, K.Ishida & Z.-Y.Qiao, Rare Met. (2015) 34(2) 864-872
<b>Ga - P</b>	Bragg-Williams R-K Polynomial		I. Ansara, C. Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson: CALPHAD, 1994, 18(4), 177-222
<b>Ga - Pb</b>	Bragg-Williams R-K Polynomial		I. Ansara, F. Ajersch: J. Phase Equil., 1991, 12(1), 73-77
<b>Ga - Pt</b>	Bragg-Williams R-K Polynomial		M. Li et al. / Intermetallics 14 (2006) 826–831
<b>Ga - Sb</b>	Bragg-Williams R-K Polynomial		I. Ansara, C. Chatillon, H. L. Lukas, T. Nishizawa, H. Ohtani, K. Ishida, M. Hillert, B. Sundman, B. B. Argent, A. Watson, T. G. Chart, T. Anderson: CALPHAD, 1994, 18(4), 177-222
<b>Ga - Sc</b>	Bragg-Williams R-K Polynomial		Z.K. Jun et al. / Journal of Alloys and Compounds 463 (2008) 511–515
<b>Ga - Si</b>	Bragg-Williams R-K Polynomial		Olesinski BAPD (6) 1985, 362-364
<b>Ga - Sn</b>	MQMPA (Pair Fraction Exp.)		B.Kumar et al., Metals (2021)
<b>Ga - Sr</b>	Bragg-Williams R-K Polynomial		X.Li et al., CALPHAD 43 (2013) 52-60
<b>Ga - Ti</b>	Bragg-Williams R-K Polynomial		Y.Liu et al., CALPHAD 41 (2013) 140-149
<b>Ga - V</b>	Bragg-Williams R-K Polynomial		D. Ling et al./CALPHAD 51(2015) 125-132
<b>Ga - Zn</b>	MQMPA (Pair Fraction Exp.)		Dutkiewicz, J., Moser, Z., Zabdyr, L., Gohil ,D. D., Chart, T. G., Ansara I., Girard, C.: Bull. Alloy Phase Diagrams, 1990, 11(1), 77-82; liquid converted to MQMPA [2022Cha]
<b>Ga - Zr</b>	Bragg-Williams R-K Polynomial		W. Luo et al. / Journal of Alloys and Compounds 587 (2014) 497–505
<b>Gd - Ho</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
<b>Gd - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Gd - La</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
<b>Gd - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Gd - Mg</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang and L. Jin, CRCT, 2012
<b>Gd - Mn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Gd - Mo</b>	Bragg-Williams R-K Polynomial		Zinkevich, JPE (22) no.1, 2001
<b>Gd - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Gd - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Gd - Nd</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
<b>Gd - Sc</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008

Gd - Si	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
Gd - Sm	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Gd - Sn	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
Gd - Sr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Gd - Ta	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
Gd - Tb	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Gd - Ti	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Gd - V	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Gd - Y	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
Gd - Yb	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
Gd - Zn	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
Gd - Zr	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
Ge - In	Bragg-Williams R-K Polynomial		P. Y. Chevalier: 1989, 155, 227-240
Ge - K	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023 with solids from Y. Wang et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 37 (2012) 72–76
Ge - Mg	MQMPA (Pair Fraction Exp.)		Dmitri Nasyrov and In-Ho Jung, "Thermodynamic modeling of the Mg-Ge-Pb system" Calphad, 2009, vol. 33, pp. 521-529.
Ge - Na	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023 with solids from Y. Wang et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 37 (2012) 72–76
Ge - Ni	Bragg-Williams R-K Polynomial		S.Jin et al./CALPHAD 38 (2012) 23–34
Ge - P	Bragg-Williams R-K Polynomial		H. Dong, X. Tao and M. Paulasto-Krockel, Journal of Electronic Materials (2021) 50:4272-4
Ge - Pb	Bragg-Williams R-K Polynomial		P. Chevalier, Thermochemica Acta, 1989, vol 155, pp. 227-240
Ge - Pt	Bragg-Williams R-K Polynomial		J.S. Wang et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 33 (2009) 561-569
Ge - Sb	Bragg-Williams R-K Polynomial		P. Y. Chevalier, Thermochimica Acta, 1989, 155, 227-240
Ge - Si	Bragg-Williams R-K Polynomial		Z. H. Long, H.S. Liu, Z.P. Jin: J. Alloys Compounds, 479 (2009) 262-267
Ge - Sn	Bragg-Williams R-K Polynomial		Y. Feutelais, B. Legendre, S. G. Fries: CALPHAD, 1996, 20(1), 109-123
Ge - Sr	Bragg-Williams R-K Polynomial		Y. Du et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 33 (2009) 719–722
Ge - V	Bragg-Williams R-K Polynomial		C.P. Wang et al. / Intermetallics 16 (2008) 544-549
Ge - Zn	Bragg-Williams R-K Polynomial		P. Chevalier, Thermochimica Acta, 1989, vol 155, pp. 227-240
H - Li	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Li - LiH	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Mg	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Mg – MgH <sub>2</sub>	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Mn	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Mn-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Na	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Na - NaH	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Ni	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FactPS Database Valid for Ni-rich region	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
H - Si	MQMPA (Pair Fraction Exp.)	Gaseous species must be	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen

		<b>taken from FactPS Database Valid for Si-rich region</b>	Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
<b>H - Sn</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FactPS Database</b>	P. Chartrand, CRCT, 2023; fitting optimized solubility from M.-C.Heuzey (FTmisc).
<b>H - Sr</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FactPS Database Valid for Sr – SrH<sub>2</sub></b>	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
<b>H - Ti</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FactPS Database Valid for Ti – TiH<sub>2</sub></b>	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
<b>H - Zn</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FactPS Database Valid for Zn-rich region</b>	J.-P. HARVEY and P. CHARTRAND, "Modeling the Hydrogen Solubility in Liquid Aluminum Alloys", <i>Metallurgical and Materials Transactions. B, Process Metallurgy and Materials Processing Science</i> , 41(4), p. 908-924, 2010.
<b>H - Zr</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FactPS Database Valid for Zr-rich region</b>	P. Chartrand, CRCT, 2014 from Ti-H of J.P. Harvey 2006
<b>Hf - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Hf - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
<b>Hf - Mg</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Hf - Mn</b>	Bragg-Williams R-K Polynomial		P.Zhou et al., JPE 33(1) 2012 pp.20-28
<b>Hf - Mo</b>	Bragg-Williams R-K Polynomial	Stoichiometric HfMo <sub>2</sub> C15 and C36 phases	G. Shao / Intermetallics 10 (2002) 429–434
<b>Hf - N</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
<b>Hf - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Hf - Nb</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Hf - Ni</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
<b>Hf - Si</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Hf - Sn</b>	Bragg-Williams R-K Polynomial		C. Tang et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 39 (2012) 91–96
<b>Hf - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Hf - Ta</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Hf - Ti</b>	Bragg-Williams R-K Polynomial		COST 507, pp.208-209.
<b>Hf - V</b>	Bragg-Williams R-K Polynomial		Servant JPE 2005
<b>Hf - Zr</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Ho - K</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ho - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ho - Mg</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang and L. Jin, CRCT, 2012
<b>Ho - Mn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Ho - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ho - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ho - Si</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Ho - Sn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>Ho - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ho - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ho - Tb</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
<b>Ho - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014

<b>Ho - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ho - Y</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008
<b>Ho - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Ho - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>In - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
<b>In - Mg</b>	MQMPA (Pair Fr Y. Liang et al. / Journal of Alloys and Compounds 475 (2009) 220–226 action Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG, M. MEDRAJ, "Experimental and thermodynamic study of the Mg-Sn-In-Zn quaternary system", <i>Journal of Alloys and Compounds</i> , 588 (2014) 75-95.
<b>In - Na</b>	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, "Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation", <i>Journal of Chemical Thermodynamics</i> , 66 (2013) 22-33
<b>In - Ni</b>	Bragg-Williams R-K Polynomial		P.Waldner and H.Ipser, Z.Metallk, 93(8) 2002, 825-832
<b>In - P</b>	Bragg-Williams R-K Polynomial		I.Ansara, C.Chatillon, Calphad 18 (1994) 204
<b>In - Pb</b>	Bragg-Williams R-K Polynomial		D.Boa, I.Ansara, <i>Thermochimica Acta</i> 314 (1998) 79-86
<b>In - Pt</b>	Bragg-Williams R-K Polynomial		Y. Liang et al. / <i>Journal of Alloys and Compounds</i> 475 (2009) 220–226
<b>In - Sb</b>	Bragg-Williams R-K Polynomial		T.J.Anderson, <i>Calphad</i> 18 (1994) 206
<b>In - Si</b>	Bragg-Williams R-K Polynomial		R.W.Olesinski, N.Kanani, G.J.Abbaschian, <i>Bull.Alloy Phase Diags.6</i> (1985) 128-130
<b>In - Sn</b>	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG, M. MEDRAJ, "Experimental and thermodynamic study of the Mg-Sn-In-Zn quaternary system", <i>Journal of Alloys and Compounds</i> , 588 (2014) 75-95
<b>In - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>In - Zn;</b>	MQMPA (Pair Fraction Exp.)		J. WANG, P. HUDON, D. KEVORKOV, P. CHARTRAND, I.-H. JUNG, M. MEDRAJ, "Experimental and thermodynamic study of the Mg-Sn-In-Zn quaternary system", <i>Journal of Alloys and Compounds</i> , 588 (2014) 75-95
<b>K - La</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2003
<b>K - Lu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Mg</b>	MQMPA (Pair Fraction Exp.)	from Mg in liquid K	P. Chartrand, 2003
<b>K - Mn</b>	MQMPA (Pair Fraction Exp.)	from Fe-K	P. Chartrand, 2013
<b>K - Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2003
<b>K - Na</b>	MQMPA (Equivalent Fraction Exp.)		P. Chartrand, 2000
<b>K - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>K - Nd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Ni</b>	MQMPA (Pair Fraction Exp.)	from Fe-K	P. Chartrand, 2013
<b>K - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for K – K <sub>2</sub> O	P. Chartrand, CRCT, 2021
<b>K - Pr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Sm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Sr</b>	MQMPA (Pair Fraction Exp.)	estimated	P. Chartrand, 2003
<b>K - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>K - Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Ti</b>	MQMPA (Pair Fraction Exp.)	from Fe-K	P. Chartrand, 2013

<b>K - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - V</b>	MQMPA (Pair Fraction Exp.)	from Fe-K	P. Chartrand, 2013
<b>K - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>K - Y</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>K - Zn</b>	MQMPA (Pair Fraction Exp.)	from Na-Zn	P. Chartrand, 2014
<b>K - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>La - Li</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>La - Mg</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, L. JIN, P. CHARTRAND, A. E. GHERIBI, K. BAI and P. WU, "Thermodynamic evaluations and optimizations of binary Mg-light Rare-Earth (La, Ce, Pr,Nd, Sm) Systems", <i>CALPHAD</i> , 38, 100-116, 2012
<b>La - Mn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>La - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>La - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>La - Nd</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
<b>La - Ni</b>	Bragg-Williams R-K Polynomial		SGTE
<b>La - Sc</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
<b>La - Si</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>La - Sn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>La - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>La - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>La - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>La - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>La - Y</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
<b>La - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>La - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Li - Lu</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Li - Mg</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
<b>Li - Mn</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2005
<b>Li - Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2003
<b>Li - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2001
<b>Li - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Li - Nd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Li - Ni</b>	MQMPA (Equivalent Fraction Exp.)		P. Chartrand, 2003
<b>Li - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species taken from FACTPS Database	P. Chartrand, CRCT, 2021
<b>Li - Pb</b>	MQMPA (Pair Fraction Exp.)	Liquid only	P. Chartrand, 2003
<b>Li - Pr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Li - Sb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2022
<b>Li - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Li - Si</b>	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, CRCT, 2007
<b>Li - Sm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Li - Sn</b>	MQMPA (Pair Fraction Exp.)		J. WANG, J. HAN, I.-H. JUNG, D. BAIROS and P. CHARTRAND, "Thermodynamic optimizations on the binary Li-Sn system and ternary Mg-Sn-Li system", <i>CALPHAD</i> , accepted, (July

			2014).
<b>Li – Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2003
<b>Li – Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
<b>Li – Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Li – Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Li – Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Li – V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Li – W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
<b>Li – Y</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2005
<b>Li – Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Li – Zn</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, CRCT, 2006
<b>Li – Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Lu – Mg</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, L. Jin, CRCT, 2012
<b>Lu – Mn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Lu – Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Lu – Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Lu – Si</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Lu – Sn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>Lu – Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Lu – Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Lu – Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Lu – V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Lu – Yb</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
<b>Lu – Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Lu – Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Mg – Mn</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, A.D. Pelton. P. Chartrand, P. Spencer and C. Fuerst, J. Phase Equil. Diff. 28 (2007), pp. 342-354
<b>Mg – Mo</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Mg – Na</b>	MQMPA (Equivalent Fraction Exp.)		P. Chartrand, CRCT, 2003 (FACT Consortium)
<b>Mg – Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Mg – Nd</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, L. JIN, P. CHARTRAND, A. E. GHERIBI, K. BAI and P. WU, "Thermodynamic evaluations and optimizations of binary Mg-light Rare-Earth (La, Ce, Pr,Nd, Sm) Systems", <i>CALPHAD</i> , 38, 100-116, 2012
<b>Mg – Ni</b>	Bragg-Williams R-K Polynomial		M.Jacobs, COST 507 report (1998) ISBN 92-828-3902-8, p.218-220
<b>Mg – O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Mg - MgO	P. Chartrand, CRCT, 2021
<b>Mg – Pb</b>	MQMPA (Pair Fraction Exp.)		Dmitri Nassirov and In-Ho Jung, "Thermodynamic modeling of the Mg-Ge-Pb system"Calphad, 2009, vol. 33, pp. 521-529.
<b>Mg – Pr</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, L. JIN, P. CHARTRAND, A. E. GHERIBI, K. BAI and P. WU, "Thermodynamic evaluations and optimizations of binary Mg-light Rare-Earth (La, Ce, Pr,Nd, Sm) Systems", <i>CALPHAD</i> , 38, 100-116, 2012
<b>Mg – Pt</b>	Bragg-Williams R-K Polynomial		W.Gierlotka et al., J.Mater.Res.,37(11) 2022,p.1904
<b>Mg – Sb</b>	MQMPA (Pair Fraction Exp.)		M. Paliwal and In-Ho Jung, "Thermodynamic modeling of Mg-Bi and Mg-Sb systems and short-range-ordering behavior of the liquid solutions"Calphad, 2009, vol. 33, pp. 744-754

<b>Mg - Sc</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, A.D. PELTON, P. CHARTRAND and C.D. FUERST, ``Critical Evaluation and Thermodynamic Optimization of the Al-Ce, Al-Y, Al-Sc and Mg-Sc Binary Systems'', CALPHAD, 32(2), 413-422, 2008
<b>Mg - Si</b>	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, M.A.Sc. thesis, Ecole Polytechnique, 2006 (VLAB Project); Vol. Data F. Gemme, CRCT, 2003 (VLAB project)
<b>Mg - Sm</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, L. JIN, P. CHARTRAND, A. E. GHERIBI, K. BAI and P. WU, "Thermodynamic evaluations and optimizations of binary Mg-light Rare-Earth (La, Ce, Pr,Nd, Sm) Systems", CALPHAD, 38, 100-116, 2012
<b>Mg - Sn</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang and A.D. Pelton, CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 34 (2010) 180 188
<b>Mg - Sr</b>	MQMPA (Pair Fraction Exp.)		Jian Wang, Ph.D. Thesis, Ecole Polytechnique, Montreal, 2014
<b>Mg - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2017
<b>Mg - Tb</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Mg - Ti</b>	MQMPA (Pair Fraction Exp.)		X.-F. Sheng, Projet de fin d'etudes, CRCT, 2008
<b>Mg - Tm</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Mg - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Mg - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Mg - Y</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, A.D. PELTON, P. CHARTRAND, P. SPENCER and C. FUERST, "Thermodynamic Database Development of the Mg-Ce-Mn-Y System for Mg Alloy Design", Metall. and Mater. Trans. A, 38A, 1231-1243, 2007.
<b>Mg - Yb</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Mg - Zn</b>	MQMPA (Pair Fraction Exp.)		Liquid: P.J. Spencer, 2006; Solids from COST-507; Vol. Data liquid: F. Gemme, CRCT, 2003; Vol. Data solids C.Aliravci, CRCT, 2007
<b>Mg - Zr</b>	MQMPA (Pair Fraction Exp.)		J.J. Marin Bejarano, CRCT, 2009
<b>Mn - Mo</b>	Bragg-Williams R-K Polynomial		COST-507
<b>Mn - N</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Mn-Mn <sub>6</sub> N <sub>5</sub> <b>Spurious stability of BCC-A2 above 2700°C when gas not present.</b>	SGTE
<b>Mn - Na</b>	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
<b>Mn - Nb</b>	Bragg-Williams R-K Polynomial		Z.-K. Liu & B. Hallstedt CALPHAD 2012
<b>Mn - Nd</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Mn - Ni</b>	Bragg-Williams R-K Polynomial	Tentative intermetallic phases	NPL (National Physics Laboratory, UK), unpublished work, 1989.
<b>Mn - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Mn - MnO	P. Chartrand, CRCT, 2021
<b>Mn - P</b>	Bragg-Williams R-K Polynomial	Valid for Mn-Mn <sub>3</sub> P region	SGTE
<b>Mn - Pb</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Mn - Pr</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Mn - Sb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023
<b>Mn - Sc</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Mn - Si</b>	MQMPA (Pair Fraction Exp.)	Mn <sub>5</sub> Si <sub>2</sub> not present	M.K. Paek, Y-B Kang, CALPHAD 46 (2014), 92-102; A. Shukla, Y.-B. Kang and A.D.
<b>Mn - Sm</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Mn - Sn</b>	Bragg-Williams R-K Polynomial		J. Miettinen: CALPHAD, 2001, 25(1), 43-58
<b>Mn - Sr</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006

<b>Mn - Ta</b>	Bragg-Williams R-K Polynomial		X.Yan et al., J. Alloys & Cmpds 865 (2021) 158715
<b>Mn - Tb</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Mn - Ti</b>	Bragg-Williams R-K Polynomial		N.Saunders, COST 507 (1998) ISBN 92-828-3902-8, p.241-244
<b>Mn - Tm</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Mn - V</b>	Bragg-Williams R-K Polynomial		W. Huang, Met. Trans.A 22A (1991) pp. 1911-1920.
<b>Mn - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Mn - Y</b>	MQMPA (Pair Fraction Exp.)		Y.-B. KANG, A.D. PELTON, P. CHARTRAND, P. SPENCER and C. FUERST, "Thermodynamic Database Development of the Mg-Ce-Mn-Y System for Mg Alloy Design", <i>Metall. and Mater. Trans. A</i> , 38A, 1231-1243, 2007
<b>Mn - Yb</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Mn - Zn</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2005
<b>Mn - Zr</b>	Bragg-Williams R-K Polynomial		H.J. Seifert, J. Groebner, F. Aldinger, F.H. Hayes, G. Effenberg, C. Baetzner, H. Flandorfer, P. Rogl, A. Saccone, R. Ferro, Proc. 3 <sup>rd</sup> . International Magnesium Conference, Ed. G. W. Lorimer, Inst.of Materials, London, 1997, pp. 257-270.
<b>Mo - N</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Mo-Mo <sub>2</sub> N	SGTE
<b>Mo - Na</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2003
<b>Mo - Nb</b>	Bragg-Williams R-K Polynomial		COST-507
<b>Mo - Ni</b>	Bragg-Williams R-K Polynomial		COST-507
<b>Mo - P</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Mo - Pt</b>	Bragg-Williams R-K Polynomial		Ph.D. Thesis Tewfik BENLAHARCHE from U. Lorraine (France) 2008
<b>Mo - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2021; based on Wang et al. Journal of Phase Equilibria and Diffusion Vol. 36 No. 1 2015
<b>Mo - Si</b>	Bragg-Williams R-K Polynomial		P.Y. Chevalier, 2003 Report to SGTE
<b>Mo - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2021
<b>Mo - Ta</b>	Bragg-Williams R-K Polynomial		COST-507
<b>Mo - Ti</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu, CRCT, 2021
<b>Mo - V</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Mo - W</b>	MQMPA (Pair Fraction Exp.)	Ideal liquid	SGTE
<b>Mo - Y</b>	Bragg-Williams R-K Polynomial		W.Chan, M.C. Gao, O.N. Dogan, P. King, JPEDAV (31) 2010; 414-420
<b>Mo - Zn</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2014
<b>Mo - Zr</b>	Bragg-Williams R-K Polynomial	Stoichiometric Mo <sub>2</sub> Zr C15	Perez, B. Sundman, CALPHAD 27 2003, 253-262
<b>N - Nb</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Nb-NbN	from FSstel
<b>N - Ni</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Ni-rich region	SGTE
<b>N - Si</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Si-rich region	SGTE
<b>N - Ta</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS Database Valid for Ta-TaN	SGTE
<b>N - Ti</b>	Bragg-Williams R-K Polynomial	Gaseous species must be taken from FactPS	from FSstel

		<b>Database Valid for Ti-TiN</b>	
<b>N - V</b>	Bragg-Williams R-K Polynomial	<b>Gaseous species must be taken from FactPS Database Valid for V-VN</b>	SGTE
<b>N - W</b>	Bragg-Williams R-K Polynomial	<b>Gaseous species must be taken from FactPS Database Valid for W-rich region</b>	SGTE
<b>N - Zr</b>	Bragg-Williams R-K Polynomial	<b>Gaseous species must be taken from FactPS Database Valid for Zr-ZrN</b>	X. Ma et al. / Journal of Alloys and Compounds 373 (2004) 194–201
<b>Na - Nb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Na - Nd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - Ni</b>	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
<b>Na - O</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species must be taken from FACTPS Database Valid for Na – Na<sub>2</sub>O</b>	P. Chartrand, CRCT, 2021
<b>Na - Pr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - Si</b>	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, M.A.Sc. thesis, Ecole Polytechnique, 2006 (VLAB Project)
<b>Na - Sm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - Sn</b>	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, “Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation”, Journal of Chemical Thermodynamics, 66 (2013) 22-33
<b>Na - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2003
<b>Na - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Na - Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - Ti</b>	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
<b>Na - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - V</b>	MQMPA (Pair Fraction Exp.)	from Fe-Na	P. Chartrand, 2013
<b>Na - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Na - Y</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Na - Zn</b>	MQMPA (Pair Fraction Exp.)		J. WANG, N. MIAO, P. CHARTRAND, I.-H. JUNG, “Thermodynamic evaluation and optimization of the (Na + X) binary systems (X = Ag, Ca, In, Sn, Zn) using combined Calphad and first-principles methods of calculation”, Journal of Chemical Thermodynamics, 66 (2013) 22-33
<b>Na - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Nb - Nd</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - Ni</b>	Bragg-Williams R-K Polynomial		J.-M. Joubert, N.Dupin, B.Sundman CALPHAD 2004
<b>Nb - Pr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - Sc</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - Si</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Nb - Sm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - Sn</b>	Bragg-Williams R-K Polynomial		C. Toffolon, Gachon, B. Sundman, JPE 2002
<b>Nb - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017

<b>Nb - Ta</b>	MQMPA (Pair Fraction Exp.)	Ideal liquid	P. Chartrand, 2017
<b>Nb - Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - Ti</b>	Bragg-Williams R-K Polynomial		COST 507, pp.256-260.
<b>Nb - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - V</b>	Bragg-Williams R-K Polynomial		K.C.H. Kumar, P. Wollants, L. Delaey, <i>Calphad</i> <b>18</b> (1994), pp. 71-79.
<b>Nb - W</b>	Bragg-Williams R-K Polynomial		P. Gustafson, Z. Metallkde <b>79</b> (1988), pp. 388-396.
<b>Nb - Y</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nb - Zn</b>	Bragg-Williams R-K Polynomial		Z. Long, JPED 2016
<b>Nb - Zr</b>	Bragg-Williams R-K Polynomial		A.F. Guillermet, Z. Metallkde. <b>82</b> (1991), pp. 478-487.
<b>Nd - Pb</b>	Bragg-Williams R-K Polynomial		M.Zydzik et al., CALPHAD 46 (2014) 1-17
<b>Nd - Pr</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Nd - Sb</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Nd - Sc</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Nd - Si</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Nd - Sm</b>	MQMPA (Pair Fraction Exp.)	Ideal liquid	Y.-B. Kang, CRCT, 2008
<b>Nd - Sn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Nd - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Nd - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Nd - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Nd - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Nd - Y</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Nd - Yb</b>	MQMPA (Pair Fraction Exp.)		Y.-B. Kang, CRCT, 2008
<b>Nd - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Nd - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ni - O</b>	MQMPA (Pair Fraction Exp.)	Gaseous species must be taken from FACTPS Database Valid for Ni - NiO	P. Chartrand, CRCT, 2021
<b>Ni - P</b>	Bragg-Williams R-K Polynomial	Ni-Ni <sub>6</sub> P <sub>5</sub>	NPL, unpublished work, 1989
<b>Ni - Pb</b>	Bragg-Williams R-K Polynomial		Cui Ping Wang, Xing Jun Liu, I. Ohnuma, R. Kainuma, K. Ishida: CALPHAD, 2000, 24(2), 149-167
<b>Ni - Pt</b>	Bragg-Williams R-K Polynomial		X.-G. Lu et al. / CALPHAD: Computer Coupling of Phase Diagrams and Thermochemistry 33 (2009) 450-456
<b>Ni - Sb</b>	Bragg-Williams R-K Polynomial		Y. Zhang et al., CALPHAD 32 (2008) 378–388
<b>Ni - Sc</b>	Bragg-Williams R-K Polynomial	Stoichiometric NiSc B2	Z. Cao et al. / Thermochimica Acta 586 (2014) 30–39
<b>Ni - Si</b>	Bragg-Williams R-K Polynomial	Spurious Ni <sub>3</sub> Si L <sub>1</sub> <sub>2</sub> below 120°C	M. Lindholm, B. Sundman, Met.Trans.A 26A (1996), pp. 2897-2903.
<b>Ni - Sn</b>	Bragg-Williams R-K Polynomial	Stoichiometric α-Ni <sub>3</sub> Sn <sub>2(lt)</sub>	COST-531
<b>Ni - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ni - Ta</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Ni - Ti</b>	Bragg-Williams R-K Polynomial		C.S. Oh, J. Korean Inst.Met.Mater. <b>33</b> (1995), pp. 129-136.
<b>Ni - V</b>	Bragg-Williams R-K Polynomial		COST 507, pp. 261-263.
<b>Ni - W</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Ni - Y</b>	Bragg-Williams R-K Polynomial		Z. Du, W. Zhang, Report F-96-07, May 1996, Univ.Science and Tech., Beijing.
<b>Ni - Zn</b>	Bragg-Williams R-K Polynomial		SGTE

<b>Ni - Zr</b>	Bragg-Williams R-K Polynomial		G. Ghosh, J.Mater.Res. 9 (1994), pp. 598-616.
<b>O - Pb</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species taken from FACTPS Database Valid for Pb - PbO</b>	P. Chartrand, CRCT, 2021
<b>O - Si</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species taken from FACTPS Database Valid for Si - SiO</b>	P. Chartrand, CRCT, 2021
<b>O - Ti</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species taken from FACTPS Database Valid for Ti - Ti<sub>2</sub>O</b>	P. Chartrand, CRCT, 2021
<b>O - V</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species taken from FACTPS Database Valid for V - V<sub>2</sub>O</b>	P. Chartrand, CRCT, 2021
<b>O - Zn</b>	MQMPA (Pair Fraction Exp.)	<b>Gaseous species taken from FACTPS Database Valid for Zn - ZnO</b>	P. Chartrand, CRCT, 2021
<b>P - Sb</b>	Bragg-Williams R-K Polynomial		I.Anbara, C.Chatillon, Calphad 18 (1994) 208
<b>P - Si</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
<b>P - Sn</b>	Bragg-Williams R-K Polynomial		J.Miettinen, Calphad, Vol. 25, No. 1, pp. 67-78
<b>P - Zn</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2022
<b>Pb - Pr</b>	MQMPA (Pair Fraction Exp.)		M.Zydzik et al., CALPHAD 46 (2014) 1-17
<b>Pb - Pt</b>	Bragg-Williams R-K Polynomial		Z.H.Long et al., J.PEDAV (2009) 30:318–322
<b>Pb - Sb</b>	Bragg-Williams R-K Polynomial		H.Ohtani, K.Okuda, K.Ishida, J.Phase Equilibria 16 (1995) 416-429
<b>Pb - Si</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2022; based on R.W.Olesinski, G.J.Abbaschian, Bull.Alloy Phase Diags. 5 (1984) 271-273
<b>Pb - Sn</b>	Bragg-Williams R-K Polynomial		Based on H.Ohtani, K.Okuda, K.Ishida, J.Phase Equilib.16 (1995) 416-429
<b>Pb - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018 using solids from H.Zhang, C.Zhang, W.W.Wang, Y.Du, P.Zhou, B.Hu, Z.Liu, J.C.Wang, J.Wang, J.Min.Metall.Sect.B-Metall. 53(3) B (2017) 179-187
<b>Pb - Tb</b>	MQMPA (Pair Fraction Exp.)		M.Zydzik et al., CALPHAD 46 (2014) 1-17
<b>Pb - Zn</b>	Bragg-Williams R-K Polynomial		T.Jantzen, P.J.Spencer, Calphad 22 (1998) 417-434
<b>Pb - Zr</b>	Bragg-Williams R-K Polynomial		Dixon P. R., Argent B. B., Chart T. G.: CALPHAD, 1998, 22(3), 397-416
<b>Pr - Sb</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Pr - Si</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Pr - Sn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>Pr - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Pr - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Pr - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Pr - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Pr - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Pr - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Pt - Si</b>	Bragg-Williams R-K Polynomial		L.L. Xu et al. / Computer Coupling of Phase Diagrams and Thermochemistry 32 (2008) 101–105
<b>Pt - Sn</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2023 based on solids from J.Hu et al., Mater. Res. Express 9 (2022) 016507
<b>Pt - Ta</b>	Bragg-Williams R-K Polynomial		SGnobl
<b>Pt - Ti</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Pt - V</b>	Bragg-Williams R-K Polynomial		C.P. Wang et al. / Intermetallics 16 (2008) 544-549
<b>Sb - Si</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
<b>Sb - Sm</b>	Bragg-Williams R-K Polynomial		SGTE
<b>Sb - Sn</b>	Bragg-Williams R-K Polynomial		C.S.Oh, J.H.Shim, B.J.Lee, D.N.Lee, J.Alloys and Cpd. 238

			(1996) 155-166
<b>Sb - Zn</b>	Bragg-Williams R-K Polynomial		L.A.Zabdyr, Calphad 21 (1997) 349-358
<b>Sc - Si</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Sc - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sc - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Sc - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sc - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sc - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2021
<b>Sc - Y</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008
<b>Sc - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Sc - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Si - Sm</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Si - Sn</b>	Bragg-Williams R-K Polynomial		M.H.G.Jacobs, P.J.Spencer, Calphad 20 (1996) 89-91
<b>Si - Sr</b>	MQMPA (Pair Fraction Exp.)		J.-P. Harvey, CRCT, 2005
<b>Si - Ta</b>	Bragg-Williams R-K Polynomial	Spurious inverted L-L miscibility gap above 2700°C	C. Vahlas, P.Y. Chevalier, E. Blanquet, Calphad 13 (1989), pp. 273-292.
<b>Si - Tb</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Si - Ti</b>	MQMPA (Pair Fraction Exp.)		FSstel
<b>Si - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, CRCT, 2018
<b>Si - Tm</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Si - V</b>	Bragg-Williams R-K Polynomial		COST 507, pp. 270-273.
<b>Si - W</b>	Bragg-Williams R-K Polynomial		C. Vahlas, P.Y. Chevalier, E. Blanquet, Calphad 13 (1989), pp. 273-292.
<b>Si - Y</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Si - Yb</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2012
<b>Si - Zn</b>	MQMPA (Pair Fraction Exp.)		A. Shukla, Y.-B. Kang and A.D. Pelton, Calphad 32 (2008) pp. 470-477
<b>Si - Zr</b>	Bragg-Williams R-K Polynomial		COST 507, pp. 280-283.
<b>Sm - Sn</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>Sm - Sr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sm - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Sm - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sm - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sm - Y</b>	MQMPA (Pair Fraction Exp.)	ideal	Y-B. Kang, CRCT, 2008 (GM Project)
<b>Sm - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Sm - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sn - Sr</b>	MQMPA (Pair Fraction Exp.)		Jian Wang, Ph.D. Thesis, Ecole Polytechnique, Montreal, 2014
<b>Sn - Ta</b>	Bragg-Williams R-K Polynomial		Lifang Yan et al., Materials 2021, 14, 1568
<b>Sn - Tb</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>Sn - Ti</b>	Bragg-Williams R-K Polynomial		F.Hayes, COST 507 (1998) ISBN 92-828-3902-8, p.284-28
<b>Sn - Tm</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>Sn - V</b>	MQMPA (Pair Fraction Exp.)		Chen, Gierlotka, and Chen, Journal of ELECTRONIC MATERIALS, Vol. 37, No. 11, 2008
<b>Sn - Y</b>	MQMPA (Pair Fraction Exp.)		Junghwan Kim and In-Ho Jung, McGill, 2016
<b>Sn - Zn</b>	MQMPA (Pair Fraction Exp.)		M. Medraj, Concordia, 2011

<b>Sn - Zr</b>	Bragg-Williams R-K Polynomial	Spurious inverted L-L miscibility gap above 1700°C in the Sn-rich region	J.Korb, K.Hack, COST 507 (1998) ISBN 92-828-3902-8, p.290-292
<b>Sr - Ta</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2018
<b>Sr - Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sr - Ti</b>	Bragg-Williams R-K Polynomial		Y. Peng et al., Journal of Phase Equilibria and Diffusion Vol. 32 No. 1 2011, p.42
<b>Sr - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sr - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Sr - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2016
<b>Sr - Y</b>	MQMPA (Pair Fraction Exp.)		P.J. Spencer, 2006
<b>Sr - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Sr - Zn</b>	MQMPA (Pair Fraction Exp.)		P. Spencer, A.D. Pelton, Y.-B. Kang, P. Chartrand, and C. Fuerst, Calphad 32 (2007), pp. 423-431
<b>Sr - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ta - Tb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ta - Ti</b>	Bragg-Williams R-K Polynomial		COST 507, pp. 293-296.
<b>Ta - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ta - V</b>	MQMPA (Pair Fraction Exp.)		P Chartrand, CRCT 2018 using Danon & Servant JACmpd, 2004
<b>Ta - W</b>	Bragg-Williams R-K Polynomial		A.F. Guillermet, W. Huang, unpublished research, KTH, 1995.
<b>Ta - Y</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ta - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2017
<b>Ta - Zr</b>	Bragg-Williams R-K Polynomial		A.F. Guillermet, J.Alloys and Compounds, 226 (1995), pp. 174-184
<b>Tb - Ti</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Tb - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Tb - Y</b>	MQMPA (Pair Fraction Exp.)		Y-B. Kang, CRCT, 2008 (GM Project)
<b>Tb - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Tb - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ti - Tm</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ti - V</b>	MQMPA (Pair Fraction Exp.)		Liquid from Zhijun Zhu, CRCT, 2021; Solids from N.Saunders, COST 507 (1998) ISBN 92-828-3902-8, p.297-298
<b>Ti - W</b>	Bragg-Williams R-K Polynomial		N.Saunders, COST 507 (1998) ISBN 92-828-3902-8, p.297-298
<b>Ti - Y</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ti - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Ti - Zn</b>	Bragg-Williams R-K Polynomial		K. Doi, S. Ono, H. Ohtani, M. Hasebe: J. Phase Equilib. Diff., 2006, 27(1), 63-74
<b>Ti - Zr</b>	MQMPA (Pair Fraction Exp.)		Liquid from P. Chartrand, CRCT, 2022; Solids from K.C.H. Kumar, P. Wollants, L. Delaey, J.Alloys and Compounds, 225 (1994), pp.121-127.
<b>Tm - V</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Tm - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Tm - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>V - W</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014 changed to MQMPA from FSstel
<b>V - Y</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>V - Yb</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>V - Zn</b>	MQMPA (Pair Fraction Exp.)	estimated	P. Chartrand, 2018 (L. est. From Fe-Zn)
<b>V - Zr</b>	Bragg-Williams R-K Polynomial		COST 507, pp. 303-304.

<b>W - Zr</b>	MQMPA (Pair Fraction Exp.)		Liquid from P. Chartrand, CRCT, 2022; Solids from SGTE
<b>Y - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Y - Zr</b>	Bragg-Williams R-K Polynomial		H. Flandorfer, J. Groebner, A. Stamou, N. Hassiotis, A. Saccone, P. Rogl, R. Wouters, H.J. Seifert, D. Maccio, R. Ferro, G. Haidemenopoulos, L. Delaey, G. Effenberg, Z. Metallkde. 88 (1997), pp.529-538.
<b>Yb - Zn</b>	MQMPA (Pair Fraction Exp.)		Zhijun Zhu and A.D. Pelton, 2014
<b>Yb - Zr</b>	MQMPA (Pair Fraction Exp.)		P. Chartrand, 2014
<b>Zn - Zr</b>	Bragg-Williams R-K Polynomial		SGTE

